Identification of the Qy Excitation of the Primary Electron Acceptor of Photosystem II: CD Determination of Its Coupling Environment

Nicholas Cox,*,† Joseph L. Hughes,†,‡ Ronald Steffen,† Paul J. Smith,† A. William Rutherford,†,‡ Ron J. Pace,† and Elmars Krausz†

Research School of Chemistry, Australian National University, Canberra, ACT 0200, Australia, and iBiTec-S, CNRS URA 2096, Bât 532, CEA Saclay, 91191 Gif-sur-Yvette, France

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Low-temperature absorption and CD spectra, measured simultaneously, are reported from Photosystem II (PS II) reduced with sodium dithionite. Spectra were obtained using PS II core complexes before and after photoaccumulation of PheoD1−, the anion of the primary acceptor. For plant PS II, PheoD1− was generated under conditions in which the primary plastoquinone was present as an anion (QA−) and as a modified species taken to be the neutral doubly reduced hydroquinone (QH2). The bleaches observed upon PheoD1− formation in the presence of QA− are shifted to the blue compared those in the presence of QAH2. This is attributed to the influence of the charge on QA−, and this effect mirrors the well-known electrochromic effect of QA− on the neutral pigments. The absorption bleaches induced by PheoD1 reduction are species dependent. Structured changes of the CD in the 680–690 nm spectral region are seen upon photoaccumulation of PheoD1− in PS II from plant, Synechocystis and Thermosynechococcus vulcanus. These CD changes are shown to be consistent with the overall electronic assignments of Raszewski et al. [Raszewski et al. Biophys. J. 2008, 95, 105], which place the dominant PheoD1 excitation near 672 nm. CD changes associated with PheoD1 reduction are modeled to arise from the shift and intensity changes of two CD features: one predominately of ChlD1 character, the other predominately PheoD2 in character. The assignments are also shown to account for the Qy absorption changes in samples where the quinone is its charged (QA−) and neutral (QAH2) states.

Introduction

Photosystem II (PS II), a pigment–protein complex found in higher plants, algae, and cyanobacteria, is responsible for the catalytic conversion of water to molecular oxygen in oxygenic photosynthesis. Its core contains the D1 and D2 reaction center polypeptides that bind the redox-active centers involved in charge separation and water oxidation.

Charge separation in PS II occurs upon excitation of P680, a chlorophyll assembly bound to both D1 and D2 protein subunits, resulting in the transfer of an electron to the neighboring pheo (PheoD1) and subsequently to plastoquinone cofactors (Qa and Qb). Electrons that re-reduce P680+ come from the oxygen-evolving complex (OEC) via the redox-active tyrosine residue 161 (Y2) of the D1 protein. The OEC, which is the substrate water-binding site of the PS II protein, is then in turn re-reduced by electrons from the oxidation of water (for review, see ref 1).

Chemical Reduction of PS II. Several chemical treatments have been identified in PS II that allow the electron transfer pathway to be poised in intermediate states. Treatment of PS II with the strong reducing agent, sodium dithionite, at pH 7 reduces the plastoquinone QA and cyto559 in the dark.2,4 Subsequent illumination at <200 K leads to the stable reduction of the primary electron acceptor PheoD1, with P680+ being re-reduced by the cyto559.2,4 The PheoD1−QA− state has a distinctive EPR split signal, and this resonance was first observed in nonoxygenic photosynthetic bacterial reaction centers (BRCs), which are structurally and functionally homologous to the reaction center of PS II.

Sustained illumination of dithionite-treated PS II above 270 K leads to further modification of the quinone (QA). This was proposed to be the charge-neutral quinol (QAH2) form, because of the disappearance of the characteristic semiquinone-iron signal,2,4 and when the iron is in the low spin form, the loss of the semiquinone EPR signal.6 In this case, when the PheoD1 is photoreduced its EPR signal is a simple radical (g ~2.003, p–p 15 G) rather than a split signal,4 consistent with the formation of the diamagnetic QAH2 state. In the current report, we refer to the form of QA following this treatment as the QA−H2 state, since this is the most likely protonation configuration for the double reduced state, though evidence for this form is circumstantial (see van Mieghem et al.1).

Assignments of the Electronic Excitations of the PheoD1 and ChlD1. Klimov et al.8 first reported the optical consequence of photoinduced reduction of PheoD1 in dithionite-treated PS II. Two bleaches were observed in the illuminated-minus-dark spectrum of the chlorin QA−Qy region at 545 nm (fwhm ~ 4 nm) and 683 nm (fwhm ~2 nm). The bleach pattern generated in PS II was similar to the absorption spectrum of Pheo in solution,9 although significantly red-shifted. The QA and Qy bleach positions correlated well with the two known electrochromic shifts induced by the formation of QA−.10

The change in linear dichroism upon photoaccumulation11 of PheoD1 indicated two components within the bleach in the Qy region. The main bleach component at 683 nm was accompanied by what was attributed to a shift of a chl pigment near 680 nm. The orientations of the transition dipoles for the Qy bleach and main QA− bleach component were found to be parallel and...
perpendicular to the membrane plane, respectively. This is consistent with conventionally assigned orientations of the transition dipoles\textsuperscript{12-14} of Pheo\textsubscript{D1} using recent crystallographic results.\textsuperscript{15}

Mutagenesis has identified that the Q\textsubscript{x} bands for Bpheo\textsubscript{b} in the BRC\textsuperscript{16} and for Pheo\textsubscript{D1} in PS II from \textit{Synechocystis}\textsuperscript{17} red shift due to a hydrogen bond to the 13\textsuperscript{1} keto ring substituent. This hydrogen bond also induces a red shift of the main Q\textsubscript{A}\textsuperscript{−} induced shift feature in the Q\textsubscript{A} region in the PS II mutants from \textit{Synechocystis}.\textsuperscript{17} The Q\textsubscript{A}\textsuperscript{−}-induced electrochromism of the BRC mutants\textsuperscript{16} has not been reported. However, the absorption and linear dichroism\textsuperscript{18} suggest that the Bpheo\textsubscript{b} Q\textsubscript{x} band blue shifts due to this hydrogen bond, and we note that there is minimal effect on the other pigments. Thus, by influencing the Bpheo\textsubscript{b} or Pheo\textsubscript{D1} the effects on the optical spectra appears to be dominated by the shift of a single band.

The strong structural and functional similarity between the electron acceptor sides of PS II and the BRC makes the Q\textsubscript{y} and Q\textsubscript{x} assignment of the Pheo\textsubscript{D1} discussed above difficult to reconcile. In the BRC, the Q\textsubscript{A}\textsuperscript{−} induced electrochromism of the Bpheo\textsubscript{b} bands for Q\textsubscript{y} and Q\textsubscript{x} is consistent with a conventional assignment of the \(\Delta\mu\) vectors being approximately in-line with the Bpheo\textsubscript{b} molecular y- and x-axes. This leads to a ~100 cm\textsuperscript{−1} red shift of the Q\textsubscript{b} band and a negligible shift on the Q\textsubscript{x} band.\textsuperscript{19,20} Conversely, in PS II there is a large blue shift of Pheo\textsubscript{D1} Q\textsubscript{x} (the C550 shift),\textsuperscript{10,21} and no readily discernible large red shift in Q\textsubscript{y}\textsuperscript{13,22}.

To reconcile this anomaly, an unconventional \(\Delta\mu\) orientation for Pheo\textsubscript{D1} has been proposed.\textsuperscript{22} This places \(\Delta\mu_{y}\) aligned perpendicular to the membrane plane and \(\Delta\mu_{x}\) parallel to the plane.

Recent Models for PSII Electronic Structure. Detailed consideration of optical lineshapes has recently led to the development of theoretical models for the electronic structure of the PSII reaction center.\textsuperscript{14,23-25} As input for these models, exciton coupling values for the reaction center pigments were estimated on the basis of structural data.\textsuperscript{15,26} followed by a global evolutionary fitting procedure to a wide range of optical data. From this approach, site energies for the reaction center pigments were estimated as well as the excitation states of the reaction center.

Raszewski et al. fit absorption-difference data for Pheo\textsubscript{D1} photoinduced reduction\textsuperscript{23} and Q\textsubscript{A}\textsuperscript{−} electrochromism.\textsuperscript{14} In this modeling, neither the main bleach upon Pheo\textsubscript{D1} photoinduced reduction nor the main Q\textsubscript{A}\textsuperscript{−} induced electrochromism feature directly identify the Pheo\textsubscript{D1} position. There are three salient features arising from this model:

(i) The Pheo\textsubscript{D1}{\textendash}Chl\textsubscript{D1} dipole–dipole coupling is ~50 cm\textsuperscript{−1}.
(ii) The lowest energy pigment transition is predominantly localized on Chl\textsubscript{D1}, at ~682 nm in D1/D2/cyt\textsubscript{b559} and ~685 nm in core complexes.\textsuperscript{14}
(iii) The exciton transition that is dominated by the Pheo\textsubscript{D1} absorption appears at ~672 nm in both D1/D2/cyt\textsubscript{b559} and core preparations.\textsuperscript{14,23} This is at significantly shorter wavelength than suggested by earlier photoinduced reduction and mutagenesis studies, as discussed above (~684 nm in PSII core complexes).

In the recent modeling of Novoderezhkin et al.,\textsuperscript{24,25} a charge-transfer state involved in the P\textsubscript{D1}{\textendash}P\textsubscript{D2} pigments is included. Charge transfer states are not included in the model of Raszewski et al.\textsuperscript{14,22} In the Novoderezhkin et al. model,\textsuperscript{24,25} the lowest energy exciton transition (~681 nm) is dominated by Chl\textsubscript{D2} but includes other reaction center pigments. In this treatment, Pheo\textsubscript{D1} contributes significantly to all the delocalized reaction center excitations. However, Novoderezhkin et al.\textsuperscript{24,25} do not include Pheo\textsubscript{D1} photoinduced reduction data or Q\textsubscript{A}\textsuperscript{−}-induced electrochromism data in their analysis.

Experimental Test of the Current Acceptor Side Model. In this study, we directly examine the question as to where Pheo\textsubscript{D1} predominantly absorbs. This allows experimental estimation of the maximum coupling strength between this pigment and Chl\textsubscript{D1} and assignment of the dominant Q\textsubscript{y} feature in the Q\textsubscript{A}\textsuperscript{−} induced electrochromic pattern. We present new spectral data, taking advantage of the improved resolution available by optical measurements taken at 1.7 K and our ability to measure CD (Abbreviations: CD, circular dichroism; EPR, electron paramagnetic resonance; PS II, Photosystem II; BRC, bacterial reaction center; Q\textsubscript{A}, primary plastoquinone A acceptor of PS II; chl\textsubscript{x}, chlorophyll\textsubscript{x}; pheo\textsubscript{a}, pheophytin\textsubscript{a}; cyt\textsubscript{b559}, cytochrome \(b_{559}\), fwhm, full width at half-maximum) as well as accurate absorption spectra simultaneously. Specifically, we address the following: (i) absorption changes associated with photoinduced reduction of Pheo\textsubscript{D1} with the quinone as either Q\textsubscript{A}\textsuperscript{−} or Q\textsubscript{A}H\textsubscript{2}; (ii) corresponding CD changes; and (iii) the species dependence of the above absorption and CD changes.

Materials and Methods

Preparation of Spinach Core Complexes. PS II core complexes from spinach were made according to the method of Smith et al.\textsuperscript{27} The cores displayed activity of ~2500–4000 \(\mu\)mol O\textsubscript{2} per mg chl/h and were stored at 1–3 mg chl/mL in 400 mM sucrose, 20 mM NaCl, 2 mM MgCl\textsubscript{2}, and 0.3 g/L DDM, pH 6.5 at 88 °C until use. \textit{Synechocystis} sp. strain PCC 6803 (\textit{Synechocystis}) cores were provided by R. Debus.\textsuperscript{28} \textit{Thermosynechococcus vulcanus} (\textit{T. vulcanus}) cores were provided by J.-R. Shen.\textsuperscript{29,30}

Dark Reduction of Q\textsubscript{A} to Q\textsubscript{A}\textsuperscript{−}. Reduction of Q\textsubscript{A} to Q\textsubscript{A}\textsuperscript{−} was achieved by adding sodium dithionite to a 1 mg chl/mL aqueous suspension of a PS II core complex in the dark. Dithionite solutions (1 M HEPES, pH 7.0 (HCl)) were prepared 1–2 min prior to use. A ~2 \(\mu\)L volume of this dithionite solution was added to ~20 \(\mu\)L of PS II sample in an Eppendorf tube. This solution was rapidly introduced into an optical cell,\textsuperscript{31} which was then attached to the sample rod and transferred to the cryostat. The sample was frozen to 4 K over a time period of ~40 s. The time taken to complete the entire procedure was 2–3 min. The final concentration of dithionite in the sample containing the PS II was ~6 mg/mL.

Double Reduction of Q\textsubscript{y} to Q\textsubscript{y}H\textsubscript{2}. Double reduction to form the Q\textsubscript{y}H\textsubscript{2} state was performed on samples initially reduced to the Q\textsubscript{y}\textsuperscript{−} state (see above). After chemical treatment with sodium dithionite to form the Q\textsubscript{y}\textsuperscript{−} state and introduction into optical cells, but prior to freezing, samples were illuminated for ~5 min at room temperature (285 K) and then allowed to relax for 1 h in the dark in an O\textsubscript{2}-free (gaseous He) atmosphere. Samples were then frozen to 4 K.

Photoinduced Reduction of Pheo\textsubscript{D1} in either the Q\textsubscript{A}\textsuperscript{−} or Q\textsubscript{A}H\textsubscript{2} “Dark” State. Photoinduced reduction of the Pheo\textsubscript{D1} state was achieved by visible illumination with either an Ar\textsuperscript{+} laser or halogen lamp. All low-temperature illuminations (1.7 K) of the PS II core complex samples used the argon laser. The laser was tuned to 514 nm, with output of 300 mW/cm\textsuperscript{2} at the sample. Illuminations were typically 15–20 min in length. High-temperature (285 K), white light illuminations used a 150 W quartz halogen lamp imaged onto the sample through a 10 cm water heat filter for ~1 min.

Optical Sample Protocols. The PS II preparation was diluted with a 1:1 (v/v) mix of ethylene glycol and glycerol as a glasing agent. The final concentration of this glasing agent was 40%
v/v. This liquid sample was introduced into a strain-free, quartz-windowed cell assembly with path length of 200 µm and mounted on a sample rod. The rod was inserted into an Oxford Instruments Spectromag 4 cryostat, through a helium gas lock, the latter fitted with quartz observation/illumination windows. The lock provided an anaerobic environment for room-temperature illumination. To achieve glasses of good quality, the sample rod was plunged into liquid helium. Absorption and CD data were collected simultaneously on a custom-built spectrometer designed in our laboratory. It employed a 0.75 m Spex Czerny-Turner single monochromator using a 1200 lines/mm grating blazed at 500 nm, with dispersion of 1.1 nm/mm.

**Qx Fitting.** A simultaneous fitting of all pheo, Qx data was undertaken (see Figure 7A–E). This data included (i) the pheo, iron ox absorption profile (in both Qx and Qχ– states); (ii) the PheoD1 bleach (in both Qχ– and QχH2 states); and (iii) the electrochromic pattern induced by the reduction of Qχ.

A least-squares minimization routine was employed to find the optimal solution. This fitting assumed the oscillator strength for both pheo, ox Qx transitions to be the same. The line shape of each pheo, transition (PheoD1 and PheoD2) was approximated by the sum of two Gaussians. Both Gaussians describing the PheoD1 were broadened by a factor of 1.5 to fit the PheoD1 bleach data in the QχH2 state (Figure 7D). For a complete description of the fitting procedure, see the Discussion and Supporting Information s4.

**Simulations of the Qχ Exciton Scheme.** Simple excitation calculations were undertaken for the PS II originating from different species in order to simultaneously fit the CD difference and absorption difference spectra upon PheoD1 reduction. Our approach is based on the model of Raszewski et al., which we evaluate on the basis of our new data on core complexes. The six core pigments of the reaction center were included in the calculation. Charge transfer states were not included. The coupling matrix of the reaction center were included in the calculation. Charge transfer states were not included. The coupling matrix of Raszewski et al. was used. Here, excitonic couplings between the reaction center pigments were calculated using the ab initio TrEsp method (see Raszewski et al.). These calculations were based on the recent 3.0 Å crystal structure of PS II. The coupling matrix is given in the Supporting Information s5. Transition dipole moments were taken to be 4.4 D for chlα and 3.4 D for pheo as per Raszewski et al.

For the calculation of the difference spectra, the absorbance and CD spectra were calculated twice, first with all pigments present and second without the PheoD1. Electrochromic shifts of the transition energies of the remaining pigments were accounted for as described in Raszewski et al. In this work, the electrochromic shifts induced by Qχ– and PheoD1– were estimated at three different orientations: 0°, 13°, and 18° off the N8–N9 axis toward Nc (see Supporting Information s5). All three angles gave approximately the same result. Simulations presented in the text assume a Δμ orientation for all pigments to be ~18° off the N8–N9 axis toward Nc. The homogeneous line shapes of the optical transitions were not calculated. These were considered to have only a relatively small influence on the estimated profile of the inhomogeneous absorption bands. As in Raszewski et al. 2008, static disorder in optical transitions was taken into account in our simulations. All pigment site energies were assumed to have a Gaussian distribution. (see Supporting Information s5 and s6). The mean site energies for the six reaction center pigments were varied, starting with the values used by Raszewski et al. so as to account for the D1/D2/cytb559 data reported in Krauss et al. It was found that it was necessary to shift the site energies of the pigment transitions only slightly (~60 cm–1) to reproduce the 2 K absorption profile. This shift is attributed to the fact that the simplified calculations presented here did not include the effects of electron–phonon coupling, which would manifest itself in a homogeneous component to the absorption band shape. Simulations reported here reproduced the absorption, CD, and PheoD1 bleach data of D1/D2/cytb559 reaction centers reasonably well. This is shown in the Supporting Information s5.

The site energies obtained from the D1/D2/cytb559 data were then further adjusted as necessary to fit the core complex data. The mean site energy of the ChlD1 was shifted to lower energy (~100 cm–1) and its site distribution narrowed (~50 cm–1). A similar modification to the ChlD1 pigment site energy/distribution was used in fitting PS II core complex data (see Raszewski et al.). In addition, the mean site energy of the PheoD2 was shifted to lower energy (~100 cm–1) to rationalize the structured CD difference.

### Results

**Plant PS II Core Complexes. Absorption Spectra of PS II Cores Following PheoD1 Photoaccumulation in the Qχ– State.** Figure 1 shows the absorption spectrum of an untreated (trace (a), black) and dithionite-treated (trace (b), red) PS II spinach core complex sample over the 450–750 nm region. The addition of dithionite in the dark (see Materials and Methods) reduced Qχ and cytb559, as is known from earlier work. The reduced form of cytb559 gave rise to a new band at 555.6 nm reviewed in Stewart et al., which is shown in detail in Figure 2.

Electronic excitations of the quinone (Qχ) do not appear in the visible region. The redox state of the quinone can, however, be inferred via its electrostatic influence on surrounding pigments, particularly via the well-known “C550” electrochromic shift on the Qχ pheo excitation (544.0 nm). This band shifts 2–3 nm to the blue as a consequence of the photoinduced formation of Qχ–. An identical band shift was observed in the (dark) absorption spectrum of chemically reduced PS II core complexes. Apart from the C550 shift, little overall change was seen in the absorption spectrum compared to that of untreated PS II core complexes. This establishes that there has been no chemical reduction or major change in electronic structure of the PS II core pigment assembly upon dithionite treatment (also see Supporting Information Figure S1).

Illumination of dithionite reduced PS II spinach core complexes at 1.7 K yielded stable bleaches at 544.0, 555.6, and 683.8 nm as identified in traces (c,d) in Figure 1 and Figure 2. The bleaches at 544.0 and 555.6 nm correspond to the loss of PheoD1 and cytb559, respectively. The overall spectral changes in the chlorin Qχ region were close to conservative, i.e., such
that the negative and positive absorption changes balance, with no net change in absorption. Most notable is the apparent narrow bleach at 683.8 nm (fwhm 2 nm). Positive absorption changes (665–680 nm) include features at 672 and 680 nm. Similar absorption changes were induced upon 285 K illumination. Their magnitude is \( \sim 5 \times \) larger than the 2 K bleach (see Supporting Information, Figure S2).

Annealing the sample at 90 K for 2–5 min led to the reversal of a broad derivative pattern in the \( Q_x \) region (trace (e) Figure 1) without a measurable change in the \( Q_y \) region (trace (e) Figure 2). Therefore, these conservative changes in absorption occurred under conditions where no reoxidation of \( \text{PheoD}_{1}^- \) occurred. We attributed the changes in \( Q_y \) upon annealing to reversible photophysical processes associated with the CP43/CP47 proximal antennae subunits that are analogous to the spectral changes seen in isolated CP43.35,36 All quantifications of \( \text{PheoD}_{1}^- \) bleaching determined in the current work was made on data from samples annealed to 90 K.

Control EPR measurements were made in conjunction with the optical measurements to clearly identify the redox state of the quinone in the dark reduced samples. In nonilluminated samples, the \( g \sim 1.9 \; Q_x^- - Fe^{2+} \) signal\(^2\) was observed, but no cytochrome b599 signal (see Supporting Information Figure S3A). Low-temperature illumination (<200 K) generated a split signal (~50 G) previously assigned in both PS II\(^2,4\) and nonoxygenic bacterial reaction centers (BRCs)\(^3\) to the interaction of the \( \text{PheoD}_{1}^- \) anion \((S = 1/2)\) with the antiferromagnetically coupled \( Q_x^- - Fe^{2+} \) \((S = 5/2)\) complex (see Supporting Information Figure S3B).

**Quantification of \( \text{PheoD}_{1}^- \) Formation in the \( Q_x^- \) State: cytochrome b599 as the Electron Donor at 1.7 K.** The degree of \( \text{PheoD}_{1}^- \) reduction induced by 1.7 K illumination can be deduced from the area of the 544.0 nm \( Q_x^- \) bleach, relative to the area of the original \( Q_y \) absorption. Illumination at 1.7 K bleached \( \sim 10\% \) of the entire \( Q_x \) pheo\(_{a}\) band (i.e., 20% of \( \text{PheoD}_{1}^- \)). Cytochrome b599 oxidation can be calculated in the same way. Approximately 20% of the cytochrome b599 absorption was lost upon 1.7 K illumination. The two signals scale quantitatively, and thus, the process can be described as formation of the \( \text{PheoD}_{1}^- - \text{cytochrome b599} \) radical pair. Illumination at 285 K (trace (f), Figure 2) led to loss of absorption equal to one pheo\(_{a}\) (i.e., 100% of \( \text{PheoD}_{1}^- \)). We did not detect the oxidation of cytochrome b599 under these conditions. Any oxidized cytochrome b599 was presumably re-reduced by the excess dithionite. A description of this quantification, including the procedure used to establish an appropriate baseline, is given in the Supporting Information (Figure S4).

The apparent narrow bleach (fwhm 2 nm) centered at 683.8 nm accounts for a loss of absorbance of \( \sim 0.5 \pm 0.1\% \) of the total chlorin \( Q_x \) region \((650–700 \text{ nm})\). This absorbance envelope consists of \( \sim 33 \text{ chl}_{a} \) and 2 pheo\(_{a}\) molecules per PS II core complex.\(^2\) Thus, the area of the bleach corresponds to a 25 \% loss of pheo\(_{a}\), assuming the pheo\(_{a}\) oscillator strength is 0.7 of one chl\(_{a}\).\(^3\) Hence, the apparent narrow \( Q_y \) bleach is in excess of what could be attributed to the actual \( \text{PheoD}_{1}^- \) bleach, as quantified by \( Q_y \), by up to 50%. The bleach thus cannot be readily described as absorption loss associated with an isolated pheo\(_{a}\). Illumination at 285 K generated a bleach of \( \sim 5 \times \) greater area (see Supporting Information Figure S2).

**Comparison of \( \text{PheoD}_{1}^- \) Formation in the \( Q_x^- \) and \( Q_{AH}^- \) States.** Panels A and B of Figure 3 compare the illuminated-minus-dark difference spectra for PS II prepared from spinach...
with charged \((Q_A^-)\) or neutral \((Q_AH_2)\) forms of the primary quinone acceptor. In both samples, 1.7 K illumination leads to cytb599 acting as the sole terminal electron donor. The \(Q_A\) bleach position is shifted 2–3 nm to the red in the \(Q_AH_2\) state (trace (b) panel A Figure 3) compared to the bleach in the \(Q_A^-\) state (trace (a) panel A Figure 3). This is consistent with the PheoD1 no longer experiencing the electrochromic influence of the anion, \(Q_A^-\). The \(Q_A\) bleach in the \(Q_AH_2\) state broadens to ∼3 nm fwhm and undergoes a small (∼0.1–0.2 nm) red shift when compared to the bleach in the \(Q_A^-\) state (Figure 3 panel B). Apart from these shifts, the bleach patterns in both states are very similar.

The shift in the \(Q_Y\) region is more readily observed in the double difference spectrum in Figure 3 panel C, trace (d). Here, the \(Q_A^-\) illuminated-minus-dark difference is subtracted from the \(Q_AH_2\) difference (Figure 3, panel B, traces (a) and (b), respectively). This subtraction in \(Q_Y\) yields a derivative feature corresponding to the main feature in the electrochromic shift pattern, with precisely the same 684.5 nm crossing point. If the \(Q_Y\) bleach position were merely broadened in the \(Q_AH_2\) state, a second differential shape would be observed in the double difference spectrum. The derivative feature seen in the double difference spectrum of Figure 3C has ∼60–70% of the peak-to-peak amplitude of the large feature seen in the electrochromic pattern of untreated PS II. Some of this reduction can be attributed to an increase in the inhomogeneous broadening evident in the apparent \(Q_Y\) bleach observed in the \(Q_AH_2\) state.

EPR measurements on PS II samples in the relaxed \(Q_AH_2\) state showed no \(Q_A^-\)-Fe\(^{3+}\) resonance, i.e., after 285 K illumination and a 1 h dark relaxation of the sample at 285 K (see Supporting Information Figure S3A). Low-temperature illumination did not generate the split signal that is characteristic of the PheoD1\(^-\)–\(Q_A^-\) state, but two other signals were observed. A \(^{31}\)P signal \(^{39}\) was seen under illumination at 5 K and a stable isolated radical centered at \(g \sim 2.003\) with a peak-to-peak spacing of 15 G was accumulated (see Supporting Information Figure S3C).

**CD Changes upon PheoD1\(^-\) Formation.** The CD spectra of a dithionite-treated spinach core complex and of untreated samples are very similar (see Supporting Information Figure S1). This further indicates that the electronic structure of the PS II pigments has not changed significantly. The CD change associated with PheoD1 reduction is not only structured, but the net area change (Figure 4) is not consistent with the CD of chl\(_b\) or pheo as isolated pigments. With the sign and magnitude of the CD of monomeric chl\(_b\) or pheo as reported (\(\Delta A \approx -1.5 \times 10^{-4}\)) \(^{38}\) the net change in CD, seen in Figure 4, when scaled for the fractional PheoD1 bleach (∼20%) is the opposite sign and 4–5× larger than the CD of monomeric pigments.

The CD change near 683.8 nm appeared as a structured feature, with components narrower than the absorption depletion seen at this wavelength. An analogous change in CD that is 5× larger was observed upon 285 K illumination, where complete reduction of PheoD1 was achieved (Supporting Information Figure S2). The CD changes that were observed for samples poised in the \(Q_A^-\) or \(Q_AH_2\) forms were very similar. The CD change observed here in intact PS II core complexes isolated from spinach is significantly different from an earlier study using D1/D2/cytb559 preparations isolated from pea.\(^{40}\) It is unclear whether this difference represents real species variation between spinach and peas or is intrinsic to the two different types of PS II preparations. This is addressed in more detail in Krausz et al.\(^{33}\)

**Species Comparison of PheoD1\(^-\) Photoaccumulation. Absorption Changes upon PheoD1\(^-\) Formation at 1.7 K.** The illuminated-minus-dark spectra observed upon 1.7 K photoinduced reduction of the PheoD1 in PS II from higher plant (for \(Q_A^-\) and \(Q_AH_2\) forms) and cyanobacteria (\(Q_A^-\) form) were similar (Figure 5). The spectra of the \(Q_X\) region demonstrate that PheoD1 reduction occurs, and they were used to quantify that cytbb565 acts as the sole electron donor at 1.7 K. Systematic differences between species are seen in the positions of the \(Q_X\) bands (see also Peterson Årskold et al.\(^{22}\)) and in the positions of the narrow \(Q_Y\) bleach. We compare first the results for spinach and Synechocystis and then discuss T. vulcanus.

The PheoD1 fraction that is photobleached is the same in plant as in Synechocystis, when monitored by \(Q_X\) (∼20%). The same mismatch in integrated intensity of \(Q_X\) and the apparent \(Q_Y\) bleach was observed. The apparent narrow \(Q_Y\) bleach is in excess of what could be attributed to the actual PheoD1 bleach, as quantified by \(Q_X\), by up to 50%. All difference spectra are close to conservative. For the \(Q_A^-\) form, in the \(Q_Y\) region the fwhm of the bleaches are 2 and 2.5 nm for spinach and Synechocystis,
respectively. The Q_AH_2 form for spinach gave an apparent Q_T bleach with a fwhm of 3 nm. Positive absorption features in the 665–680 nm region were similar in both species, with notable features at 672 and 680 nm in spinach and 672 and 678 nm in Synechocystis. The Q_X and apparent Q_T bleach positions are blue-shifted in Synechocystis compared to spinach.

The bleaches seen in PS II from T. vulcanus (Figure 5, trace (d)) retain the key features seen in plant PS II. A narrow (2.5 nm fwhm) apparent Q_T bleach at 685.8 nm was observed. In T. vulcanus, the Q_Y bleach was to the blue compared to spinach, as in Synechocystis. However, unlike Synechocystis the Q_Y bleach for T. vulcanus was to the red compared to spinach. The fractional PheoD1 bleach (as estimated from Q_Y) corresponded to 10–15% of reaction centers. This is significantly less than the bleach seen in either spinach or Synechocystis.

Upon annealing to 90 K, this relatively small PheoD1 Q_X bleach decayed by ~50%. In other species measured (spinach and Synechocystis), the corresponding bleach was stable at 90 K. As a consequence, the bleach spectra reported in Figure 5 trace (d) for T. vulcanus is the difference between a spectrum taken immediately after illumination at 2 K and a spectrum taken before illumination. The other species studied were annealed at 90 K to remove photophysical artifacts (see the above section Absorption Spectra of PS II core complexes following PheoD1 photoaccumulation in the Q_A state). As with spinach and Synechocystis, the net area of the apparent Q_T bleach near 685 nm overestimated the degree of PheoD1 reduction by up to ~50%.

**CD Changes upon PheoD1- Formation at 1.7 K.** The CD changes in all the species are very distinctive and appear as structured signals about the narrow absorption bleach features (Figure 6). In all the CD difference spectra, there is a positive lobe approximately in the position of the bleach.

In the 650–680 nm region, there may be some other smaller changes in the CD spectra upon photoinduced PheoD1 reduction, but these are broad and comparable to the noise level. These CD data are shot-noise-limited, which leads to signal-to-noise ratio being somewhat lower in the 650–680 nm region, owing to the higher absorption in this region. Only for Synechocystis are there any features of magnitude comparable to the structured changes in the region of the main narrow absorption bleach.

**Discussion**

**Q_X PheoD1 Transition in Higher Plant PS II: Absorption Spectra.** A summary of our analysis of the pheoa Q_X transitions in PS II core complexes from spinach is shown in Figure 7. To extract the pheoa absorption profile from the data, we performed an iterative baseline correction procedure that required self-consistency within all the data sets presented in Figure 7 (see Supporting Information S4). We present (Figure 7) the baseline-subtracted data and the corresponding fits to (i) the absorption spectra in the Q_A and Q_X states (panels A and B), (ii) the bleaches in the Q_A and Q_AH_2 states (panels C and D), and (iii) the (Q_A−minus-Q_AH_2) electrochromic shift (panel E).

A global fit of all the data was made, as described in the Materials and Methods and in the Supporting Information S4. The use of single Gaussians for each pheoa does not adequately describe the data. The line shapes for each pheoa were then modeled as a sum of two Gaussians. Equal total areas were attributed to each pheoa. For one of the pheoa absorption bands, the use of two Gaussians was unnecessary, as the fitting process yielded two bands of equivalent width and position (Q_A/Q_AH_2 18477 cm⁻¹, fwhm = 17.0 nm). The other pheoa band is

| Table 1: PheoD1 Absorption Bleach Positions and Corresponding CD Turning Points |
|-------------------|-------------------|
| Q_X bleach position (nm) | Q_Y bleach position (nm) | Q_AH_2 bleach position (nm) |
| Spinach (Q_A) | 544.0 | 683.8 | 682.4 | 683.5 | 685.1 |
| Spinach (Q_AH_2) | 545.6 | 684.0 | 682.0 | 683.7 | 685.4 |
| Synechocystis (Q_A) | 541.2 | 681.8 | 678.2 | 681.1 | 684.6 |
| T. vulcanus (Q_A) | 541.4 | 685.8 | 683.3 | 685.5 | - |

All bleach positions and CD turning points are summarized in Table 1.
dominated by a narrow component (\(Q_{\chi}/Q_{\chi}H_2\) 18 386/18 332 cm\(^{-1}\), fwhm = 5.0 nm), with a broader contribution (\(Q_{\chi}\) 18 536/18 481 cm\(^{-1}\), fwhm = 11.2 nm) to higher energy by 150 cm\(^{-1}\). The higher-energy component may be attributed to phonon/vibrational side-structure. In order to fit the pheoa bleach obtained in the presence of Q\(_{\chi}H_2\), a broadening factor of \(\sim 1.5\) was introduced for the pheoa that we assign to PheoD\(_{\chi}\).

From the analysis described above, it can be shown that, in the Q\(_{\chi}\)-induced electrochromism, the entire pheoa Q\(_{\chi}\) band does not shift uniformly (Figure 7). The shift is predominantly of the band that contains the narrower component. The photoinduced bleaches in both the Q\(_{\chi}H_2\) and Q\(_{\chi}\) states of PS II are also of this absorption band. For bleaches in the Q\(_{\chi}\) state, the EPR split signal shows that the active branch pheoa (PheoD\(_{\chi}\)) is being reduced upon photoillumination. The similarity of the Q\(_{\chi}\) bleach in the Q\(_{\chi}H_2\) state, except for an increase in the inhomogeneous broadening, suggests that the same pheoa is reduced by the illumination. This is a clear demonstration that it is the PheoD\(_{\chi}\) that is responsible for the well-known C550 shift, as expected.

Jankowiak et al.\(^{41}\) modeled the pheoa Q\(_{\chi}\) absorbance in D1/D2/cytb559 reaction centers (RC-5) using two Gaussians of approximately the same width and separated by \(\sim 90\) cm\(^{-1}\). Bleaching of the pheoa Q\(_{\chi}\) absorbion via both chemical and photoinduced reductions was used to assign the lower-energy component to PheoD\(_{\chi}\). This phenomenology is consistent with our description of the pheoa Q\(_{\chi}\) absorption bands for the spinach core complex data (see above), where the average separation of the PheoD\(_{\chi}\) and PheoD\(_{\alpha}\) excitations in the charge-neutral state \((Q_{\chi}H_2)\) is \(\sim 70\) cm\(^{-1}\), and with PheoD\(_{\chi}\) appearing at lower energy. As absorbance features are known to broaden in D1/D2/cytb559 reaction center preparations as compared to core complexes,\(^{33}\) differences in the line shape of the PheoD\(_{\chi}\) excitation between these preparations are not surprising.

**Q\(_{\chi}\) PheoD\(_{\chi}\) Transition in PS II.** As discussed in the Introduction, literature models propose two quite different positions for the dominant Q\(_{\chi}\) PheoD\(_{\chi}\) excitation in PS II; either \(\sim 684\) nm or \(\sim 672\) nm. These two possibilities are discussed below within the context of the most recent crystallographic data,\(^{15}\) which provide the relative positions and orientations of the reaction center pigments. There is little dispute that ChlD\(_{\chi}\) absorbs near \(684\) nm. EPR orientation studies\(^{42}\) and more recently absorption spectra using site-directed mutants\(^{43,44}\) have shown that the \(\delta P\) bleach is predominately localized on the ChlD\(_{\chi}\) pigment.

**Assigning Q\(_{\chi}\) PheoD\(_{\chi}\) at 684 nm.** The Q\(_{\chi}\) PheoD\(_{\chi}\) excitation has historically been assigned to \(\sim 684\) nm, via the narrow bleach seen at this wavelength upon PheoD\(_{\chi}\) reduction and the electrochromic shift seen at a similar wavelength upon Q\(_{\chi}\) reduction (see Introduction). This assignment is however difficult to reconcile with the results presented in this work as it does not account for (i) the area of the apparent Q\(_{\chi}\) bleach; (ii) the position of the apparent bleach in the Q\(_{\chi}\) state, relative to the Q\(_{\chi}H_2\) state; and (iii) the highly structured CD change in the region of the bleach.

The area of the apparent Q\(_{\chi}\) bleach at \(\sim 684\) nm is significantly greater than that for a pheoa once calibrated against the area of the Q\(_{\beta}\) bleach. This result requires the “bleach” to be composite.\(^{11}\) The “bleach” can also be associated with the negative edge of a blue shift on ChlD\(_{\chi}\), appearing at \(\sim 684\) nm, induced by the electric field of the reduced PheoD\(_{\chi}\).

ChlD\(_{\chi}\) is \(\sim 10\) Å from the PheoD\(_{\chi}\).\(^{15}\) The expected electrochromic shift on the ChlD\(_{\chi}\) upon PheoD\(_{\chi}\) reduction is \(80\) cm\(^{-1}\), taking the orientation of \(\Delta\mu_{\delta}\) along the N\(_{\beta}-N_{\delta}\) axis\(^{23}\) (IUPAC nomenclature\(^{46}\)) and using a effective dielectric constant of \(\epsilon = 2\). This value is not significantly altered by rotating the \(\Delta\mu_{\delta}\) direction in the chlorin ring by \(\sim 15^\circ\) in the direction of N\(_{\beta}\) as used in ref 14. This shift is greater than the line width of ChlD\(_{\delta}\). If PheoD\(_{\delta}\) and ChlD\(_{\delta}\) both absorbed near \(684\) nm, the total “bleach” should correspond to the area of 1 chl and 1 pheoa, combined, and consequently, \(\sim 2\) times larger than experimentally observed. Assigning a smaller shift on ChlD\(_{\delta}\) to reduce the observed bleach area would require an unconventional orientation of its \(\Delta\mu_{\delta}\) (70–80° off the N\(_{\alpha}-N_{\delta}\) axis toward N\(_{\gamma}\))

The position of the apparent bleach shifts to lower energy by \(\sim 10\) cm\(^{-1}\) in the Q\(_{\chi}H_2\) state as compared to the Q\(_{\chi}\) state. The expected position of the PheoD\(_{\chi}\) bleach in the Q\(_{\chi}H_2\) state as compared to the Q\(_{\chi}\) state, again using crystallographically determined orientations of the \(\Delta\mu_{\delta}\), now for PheoD\(_{\chi}\), is \(\sim 60\) cm\(^{-1}\) to higher energy.\(^{12,14}\) Thus, assigning the PheoD\(_{\chi}\) excitation at \(\sim 684\) nm would require an unconventional \(\Delta\mu_{\delta}\) orientation for PheoD\(_{\chi}\) as well as an unconventional orientation of \(\Delta\mu_{\\delta}\) for ChlD\(_{\delta}\). An unconventional orientation for \(\Delta\mu_{\delta}\) of PheoD\(_{\chi}\) was proposed by Peterson Årsköld et al.\(^{22}\)

The highly structured CD change associated with reduction of the PheoD\(_{\chi}\) observed in spinach places significant constraints on any description of the electronic structure of the reaction center pigment system. The \(\Delta\alpha/\Delta\beta\) of the CD features is of the order of \(10^{-3}\) and consequently an order of magnitude higher than normally associated with monomeric chl CD. Integration of the total CD change observed establishes that this CD difference is close to conservative. Both these observations suggest that the changes seen have an excitonic origin.

Importantly, the individual features in the CD difference spectrum seen in the region of the apparent Q\(_{\chi}\) bleach are significantly narrower than any absorption (bleach) feature seen. Hence, the CD difference spectrum does not identify the position of a single excitation but instead indicates the overlap of excitations. The highly structured CD feature that coincides with the bleach must therefore arise from at least two exciton states near that position. Furthermore, these two exciton states must lie within each other’s line width, (i.e., the exciton splitting between these two pigments cannot exceed \(\sim 25\) cm\(^{-1}\)).

The structured CD change cannot be readily interpreted within a model where the PheoD\(_{\chi}\) excitation coincides with the bleach position (\(\sim 684\) nm). In such a model, the two overlapping exciton states giving rise to the structured CD change would be associated with PheoD\(_{\chi}\) and ChlD\(_{\chi}\). Taking the conventional orientation of the transition dipole moments for the PheoD\(_{\delta}\) and ChlD\(_{\delta}\), or even allowing them to vary by up to 20°, leads to an dipole–dipole coupling of 50–70 cm\(^{-1}\). This estimate of coupling is reduced by using a transition monopole method but remains at \(\sim 45–55\) cm\(^{-1}\).\(^{14,23–25}\) This value is \(\sim 2\times\) the maximal coupling allowable from the nature of the observed CD changes upon PheoD\(_{\delta}\) reduction. Hence, assigning the PheoD\(_{\chi}\) excitation to be at \(684\) nm would also require the transition dipole orientations of PheoD\(_{\chi}\) or ChlD\(_{\delta}\) (or both) to be unconventional.

**Assigning Q\(_{\chi}\) of PheoD\(_{\chi}\) at 672 nm.** The data presented in this paper are consistent with the dominant PheoD\(_{\chi}\) excitation being at \(672\) nm. Following the basic assignments of Renger and co-workers, our modeling suggests that the exciton state at \(683.6\) nm that is responsible for the apparent Q\(_{\chi}\) bleach is dominated by the accessory ChlD\(_{\chi}\). Upon reduction of PheoD\(_{\chi}\) the “ChlD\(_{\chi}\) exciton” shifts to the blue. The negative edge of this shift appears then as a bleach.

The integrated intensity of the apparent Q\(_{\chi}\) bleach at 683.8 nm is consistent with it being the loss of a chlorophyll, as opposed to the loss of a pheoa (oscillator strength is \(\sim 0.7\) chl\(_{\alpha}\)\(^{18}\)).
The electrochromic shift on ChlD1 induced by the reduction of Q_{A} is calculated to be relatively small. This is consistent with experimental observations of a value of \( \sim 10 \) cm\(^{-1}\). The apparent Q_{B} bleach positions in the Q_{A} and Q_{B}H_{2} systems are different by \( \sim 10 \) cm\(^{-1}\) as is seen in the double-difference spectra, again consistent with its assignment to ChlD1.

Our simulations (discussed below) show that the assignment of the apparent Q_{B} bleach as a shift on the ChlD1 pigment is also consistent with the CD changes observed. Within the model of Raszewski et al., \(^{14,23}\) the exciton states involved in this CD change involve a dominantly ChlD1 state along with a dominantly PheoD2 state. Evidence for such a PheoD2 assignment in D1/D2/cytb559 reaction centers comes from experiments where PheoD2 was substituted by a chemically modified phaeophytin which absorbed well to the blue of Pheo. This led to a loss of absorption at \( \sim 680 \) nm. \(^{47}\) Significant CD changes were also observed at \( \sim 680 \) nm upon PheoD2 replacement.

Within this scheme, the CD difference feature we observed can be understood (see below). The highly structured pattern arises from changes in the CD of the two exciton states (as discussed above), one dominated by ChlD1 and the other by PheoD2. Indeed, PheoD2 is the only inner reaction center pigment to which ChlD1 does not strongly couple. Estimates for the coupling between these two pigments are \( \sim 3 \) cm\(^{-1}\) and thus less than the \( \sim 25 \) cm\(^{-1}\) upper limit required for it to be consistent with the CD data.

**Simultaneous Simulation of the CD Changes and Q_{B} Absorption Bleach Data.** Simple exciton calculations were undertaken, for the PS II originating from different species, to simultaneously fit the CD difference and absorption difference spectra upon PheoD1 reduction (see the Materials and Methods and Supporting Information s5). Our simulation of the absorption bleach upon PheoD1 reduction in PS II cores from spinach (dark (from Figures 3B(b), 6A(b)) Panel A: Positions of the six absorption (exciton) bands in the Q_{A}H_{2} state. Panel B: Positions of the five remaining absorption bands after the bleach of the PheoD1. Panel C: Simulation of the light-minus-dark absorption difference. Panel D: Positions of the six CD bands after the bleach of the PheoD1. Panel E: Positions of the five remaining CD bands after the bleach of the PheoD1. Panel F: Simulation of the light-minus-dark CD difference. Color coding for panels A, B, D, E; black, theorethical absorption/CD spectrum; blue, ChlD1 absorbance/CD band; red, PheoD1 absorbance/CD band; green, PheoD2 absorbance/CD band; cyan + pink, P_{A}H_{2}/P_{B}H_{2} absorbance/CD bands; yellow, ChlD2 absorbance/CD band. Color coding for panels C and F; black, data, absorbance/CD difference; blue, dark PheoD1Q_{A}H_{2} state; red, light PheoD1Q_{A}H_{2} state; green, simulation of absorbance/CD difference (i.e., red trace – blue trace).

PheoD2 has been one of the least experimentally accessible pigments in the PS II core complexes. Its assignment to \( \sim 680 \) nm by Raszewski et al. is simply inferred from its position in D1/D2/cytb559. No experimental data have been available to directly probe the position of the PheoD2 in core complexes. The position and orientation of PheoD2 ensure that it is relatively insensitive to electrochromic effects associated with ChlD1 and PheoD1. Upon addition, a spectral shift of the site energy of PheoD2 to the red will have a smaller influence on the exciton states than other pigments. PheoD2 only couples significantly to ChlD2 and in the Raszewski et al. model; their site energies are separated by \( \sim 200 \) cm\(^{-1}\).

Our overall assignments are consistent with those of Raszewski et al.\(^{14,23}\) with the exception of the placement of PheoD2. The assignment of the “PheoD2 exciton” to \( \sim 685 \) nm represents a shift to lower energy of \( \sim 100 \) cm\(^{-1}\) compared to its position in D1/D2/cytb559. The highly structured CD changes seen upon PheoD1 reduction in PS II core complexes presented in this work closely constrain the position of the PheoD2 exciton in intact PS II core complex preparations. This then points to a significant spectral shift of its position compared to that observed in isolated D1/D2/cytb559 preparations. It is noted that in the model of Raszewski et al.\(^{14,44,45}\) a similar shift in the site energy of the ChlD1 between core complexes and isolated D1/D2/cytb559 reaction centers was required. A more definitive assignment of the “PheoD2 exciton” based on our data may require a more...
rigorous treatment of the homogeneous line shapes of the pigment transitions.

Species Comparison of Absorption Difference Spectra. Changes in the hydrogen bonding from the protein to the 13\(^1\) keto of the Pheo\(_{D1}\) in PS II\(^{17}\) and the BPheo\(_{L}\) in the BRC\(^{16,18}\) have been shown to affect the pigment site energy. In the wild-type BRC, l-Glu104 provides a hydrogen bond to the BPheo\(_{L}\) 13\(^1\) keto, while in Synechocystis, the corresponding residue is D1-Gln130, which does not provide a hydrogen bond for the Pheo\(_{D1}\) 13\(^1\) keto.

In the BRC and Synechocystis systems, mutants have been created that target these residues so as to remove or add a hydrogen bond, respectively. In both systems, it was shown that a hydrogen bond to the (B)Pheo\(_{D1(L)}\) 13\(^1\) keto results in a red-hydrogen bond, respectively. In both systems, it was shown that seen for D1/D2/cytb559 preparations, as monitored by the triplet of the bleach (see Figure 5).

Consequently, there is less overlap of the Pheo\(_{D2}\) and Chl\(_{D1}\) in higher plants, as Glu at position D1\(^{48}\) and Pheo\(_{D1}\) in PS II\(^{17}\) and the BPheo\(_{L}\) in the BRC\(^{16,18}\) consequently, the Pheo\(_{D1}\) Q\(_X\) band is blue-shifted due to the hydrogen bond,\(^{18}\) while in Synechocystis, the dominant Q\(_X\) electrochromic feature is red-shifted.\(^{17}\)

Synechocystis, \textit{T. vulcanus}, and higher plants differ in their D1 polypeptide sequence in the region where Pheo\(_{D1}\) is bound. In higher plants, there is a Glu at position D1\(^{130}\), while the corresponding residue in \textit{T. vulcanus} and Synechocystis is D1-Gln130, at least under normal growth conditions.\(^{48}\) Consequently, the Pheo\(_{D1}\) Q\(_X\) band in higher plants is 2.5 nm to the red as compared to either Synechocystis\(^{17}\) or \textit{T. vulcanus},\(^{49}\) as determined by the C550 electrochromic shift and the position of the bleach (see Figure 5).

Within the model presented above,\(^{14,23}\) the apparent bleach in Q\(_X\) is assigned to a shift on the Chl\(_{D1}\). Our calculations establish that a \(\sim 50\text{ cm}^{-1}\) shift of a Pheo\(_{D1}\) pigment centered at \(\sim 672\text{ nm}\) has a minimal influence on the transition energy of the Chl\(_{D1}\) exciton (\(\sim 2\text{ cm}^{-1}\)). Thus, a change in the Q\(_Y\) bleach position between organisms indicates a variation in the mean site energy distribution of the Chl\(_{D1}\) rather than Pheo\(_{D1}\).

Species Comparison of CD Difference Spectra. The CD difference spectrum in Synechocystis can be rationalized in the same way as for spinach (see Figures 8 and 9). The absorption band for the Chl\(_{D1}\) in this organism shifts to the blue as compared to spinach, as evidenced by the apparent Q\(_X\) bleach. Consequently, there is less overlap of the Pheo\(_{D1}\) and Chl\(_{D1}\) "exciton bands", leading to a broader CD pattern. The transition energy of the accessory Chl\(_{D1}\) in Synechocystis is close to that seen for D1/D2/cytb559 preparations, as monitored by the triplet bleach position.\(^{44,45,50}\) To fit the CD change, the position of the Q\(_Y\) Pheo\(_{D1}\) pigment had to be shifted to the red. This is in accord with the expected electrochromism that is induced due to the presence of Q\(_{X}^{-}\). Arguably, the CD fitting in Synechocystis is significantly better than that for spinach; it provides a better fit to the structured CD change at the position of the bleach (Figure 9F). This is perhaps not surprising, as the crystallographic data that was used as the basis for these simulations comes from cyanobacterial sources, not higher plants.

The CD difference as seen in \textit{T. vulcanus} is most likely a photophysical effect associated with illumination-induced changes of the proximal light-harvesting complexes CP43 and CP47. The Q\(_X\) CD changes seen are virtually lost upon annealing at 90 K with a corresponding \(\sim 50\%\) decrease in Pheo\(_{D1}\) as estimated by the Q\(_Y\) bleach. The CD changes seen are similar to the photophysical CD changes observed in isolated spinach CP43 antenna complexes,\(^{36}\) except red-shifted. The narrow long-wavelength absorption of CP43 in \textit{T. elongatus}, an organism very similar to \textit{T. vulcanus}, is red-shifted as compared to spinach.\(^{51}\) The negative CD feature in \textit{T. vulcanus} core complexes, which is at least partly associated with CP43, appears at 684.8 nm, compared to 683.3 nm for the corresponding feature in spinach core complexes.

As most of the illumination-induced CD changes in \textit{T. vulcanus} are associated with photophysical rather than photochemical changes, modeling becomes problematic. Unlike the other species studied, the lowest energy CD difference feature in \textit{T. vulcanus} is positively signed. Within the model described, the lowest energy CD difference feature is associated with the dark state (prior to Pheo\(_{D1}\) reduction) for \textit{T. vulcanus} is the Chl\(_{D1}\) exciton. This exciton band must have a positive CD sign, and thus, the red-most CD difference feature should be negative and not positive. When a scaled subtraction of the CP43 photophysical CD change is made, the remnant CD change is within the noise and 2---3 times smaller in amplitude than the corresponding change seen in spinach. This small amplitude is consistent with our modeling for \textit{T. vulcanus}, where a significant red shift of the dominantly Chl\(_{D1}\) exciton, relative to its position in other organisms, leads to a decrease in its CD intensity.

Reaction Center of PS II. Frese et al.\(^{52}\) reported an anomalously large Stark signal for the Pheo\(_{D1}\) Q\(_X\) band in D1/D2/cytb559 preparations, as well as nonclassical Stark behavior in the long-wavelength Q\(_X\) region. In BRC mutants, it has also been suggested\(^{53}\) that a classic Liptay-type analysis may be inadequate to describe the Stark spectra. Deviations from the typical Liptay line shape behavior in Stark spectra have been predicted when there is mixing of the states involved in the observed transition with one (or more) other states.\(^{53}\) The mixing of pigment transitions with charge transfer states has been featured in these discussions of non-Liptay Stark behavior.\(^{52,53}\)

Such considerations may be required to further our understanding of the Q\(_{X}^{-}\) and Pheo\(_{D1}\)\(^{-}\)-induced electrochromism in PS II.

**Figure 9.** Fit of light-minus-dark spectra in Synechocystis PS II cores poised in the Q\(_{X}^{-}\) in the Q\(_{X}\) region (see text). Panel A: Positions of the six absorption (exciton) bands in the Q\(_{X}\) state. Panel B: Positions of the five remaining absorption bands after the bleach of the Pheo\(_{D1}\). Panel C: Simulation of the light-minus-dark absorption difference. Panel D: Positions of the 6 CD bands in the Q\(_{X}^{-}\) state. Panel E: Positions of the five remaining CD bands after the bleach of the Pheo\(_{D1}\). Panel F: Simulation of the light-minus-dark CD difference. Color coding for panels A--F are exactly the same as Figure 8.
Novoderezhkin et al.24,25 include charge transfer states in their modeling of PS II optical spectra and incorporate Stark data. However, they do not address QA−-induced electrochromism or PheoD1− photoaccumulation data. As discussed above, Raszewski et al.14,23 do address the latter data, but do not include charge transfer states in their model or Stark data. A significant advancement of our understanding of the electronic structure of the PSII reaction center may arise from an integration of the current theoretical approaches and the new experimental data presented here.

It seems appropriate that charge transfer states and subsequent charge transfer character of pigment excitations be incorporated in any modeling of electrochromic effects in PS II core complexes. Such an approach may also help reconcile the marked difference in electrochromic effects between PS II and BRC. Highlighted is the large electrochromic shift on the QA band of PheoD1 upon QA− formation in PSII compared to the barely measurable electrochromic shift of the BPheoL QA transition in the BRC.20

Summary

Low-temperature absorption and structured CD changes associated with the photo-accumulation of PheoD1− in intact PSII core complexes provide new evidence for the assignment of the reaction center pigments. These results are shown to be consistent with the overall assignments of Raszewski et al.14,23 which place the dominant PheoD1 excitation near 672 nm.

The highly structured CD changes seen upon PheoD1 reduction in PSII core complexes presented in this work closely constrain the positions of the PheoD2 exciton in intact PSII core complex preparations. Simple exciton calculations point to a significant spectral shift of this pigment when compared to that observed in isolated D1/D2/cyt655 preparations.

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Supporting Information Available: Additional material detailing optical control measurements, additional data from room temperature bleaching experiments, EPR controls, and a further description of the model given in the text. This material is available free of charge via the Internet at http://pubs.acs.org.

References and Notes

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