DIFFERENTIAL RESPONSES OF PHOTOSYSTEM II ACTIVITY TO PHOTOOXIDATION IN RED AND GREEN TISSUES OF AMARANTHUS TRICOLOR LEAVES

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Abstract

In order to study the antioxidative potential of amaranthine and its relationships with photoprotection, changes of PS II activity of red and green tissue in the same edible amaranth leaf were compared under photodestruction treatment induced by MV (methyl viologen) or H2O2. In the first 90 min of MV treatment, PS II maintained stable activity; the chlorophyll-fluorescence parameters were close to those of the controls. However, with the extension of treatment time, during 90–300 min of continuous photoinhibitory stress, Fv/Fm, Y (II) and Y (NPQ) of leaf discs decreased significantly, while Y (NO) and Fv were dramatically increased, which showed that the activity of PS II suffered irreversible photoinhibitory damage, suppressed or even completely inhibited. Compared with the control, under 50 mmol L−1 of H2O2 stress for 360 min, Y(NPQ) and Fv decreased, accompanied by Y(NO) and the Fv/Fm decreases, while Y(II), qP and ETR were slightly increased, reflecting that treatment with 50 mmol L−1 H2O2 promoted distribution of light energy in photochemistry reaction. After photodestructive treatments, a lower Fv/Fm, Y (NO) but higher Fv/Fm, Y (II), qP and Y (NPQ) was found in the red leaf discs compared with the green ones. The decrease and the decline rate of the amaranthine content in the green and red leaf discs implied that the PS II of green leaf discs have a higher sensitivity to photoinhibitory damage than the red ones. It is suggested that the higher content of amaranthine may be used as a detoxification mechanism to counteract photoinhibitory in red leaf.

Introduction

The mechanisms of photoinhibitory damage and defense against photoinhibitory stress have always been significant issues. Many biotic and abiotic stresses can increase the generation of reactive oxygen species (ROS) in plants (Wang et al., 2013; Shamim et al., 2013; Kanswal et al., 2013). The increase of reactive oxygen species in cells can induce (1) the generation of photooxidation stress and damage of organelles, in particular the photosynthetic apparatus; and (2) the rapid elimination of ROS by efficient antioxidative systems and refined regulatory mechanisms during photooxidation (Foyer et al., 1994). Moreover, antioxidative enzymes and low molecular-weight organic antioxidants distribute ubiquitously in the leaf, where plant pigments have a close association with antioxidative activity. Among all frequently-found coloration pigments, both carotenoids and anthocyanins are ascertainment to efficiently scavenge free radicals in plants (Gould et al., 2002; Tanaka et al., 2008). However, there is still limited information about the anti-photoinhibitory capacity of another important secondary metabolite, amaranthine, a homologue of anthocyanidins in higher plants (Cai et al., 1998; Wang et al., 2006).

Amaranthine is a class of water-soluble red-violet pigments, accumulated in red roots, stems and leaves of plants of Amaranthus (Amaranthaceae), belonging to Betacyanins of Betalain (Cai et al., 2001; Stintzing & Carle, 2004). As a secondary metabolite of nitrogenous plants, amaranthine is a potential antioxidant because of its abundance of hydroxyl and imino groups (Cai et al., 1998; Strack et al., 2003). Methods for extracting amaranthine and its chemical property analysis have been investigated recently (Zhan et al., 2003; Cai et al., 2005; Gu et al., 2008). However, the accumulation of amaranthine in Amaranthaceae plants and its related physiological function have been rarely reported. Our previous study revealed that red edible amaranth can decrease H2O2 accumulation and increase heat resistance by accumulating amaranthine quickly (Shao et al., 2008). One red cultivar of amaranth exhibited a red-green leaf phenotype, the proportion of red part and green part in the leaf being 1:2 (Xiao et al., 2000). Leaf discs of red or green parts were treated with H2O2 or MV solutions, which provides an efficient and in-depth approach to probe the antioxidative potential of amaranthine and the relationships with photoprotection. To compare antioxidative capability between red tissue and green tissue in the same leaf under photodestruction treatments, chlorophyll fluorescence measurements were carried out with an IMAGING-PAM chlorophyll fluorometer (Heinz Walz GmbH, Effeltrich, Germany). By using chlorophyll fluorescence technology in the present study, more experimental evidence was obtained to extend our understanding of the physiological functions of amaranthine, helping to elucidate the mechanisms that protect the photosynthetic apparatus in plants.

Materials and Methods

Plant materials and growth conditions: The variety of edible amaranth (Amaranthus tricolor L.) is a red cultivar of amaranth. Seeds of Amaranthus tricolor L. were obtained from Nanhai Dali Jiang Zhiqing Seed Company, Guangdong Province, China. Seeds were germinated in plastic potted with soil on the experimental field at South...
Absorption spectra and contents of amaranthine:
Amaranthine content was measured as described by Stintzing et al. (2002). Fresh leaf discs from the edible amaranth (0.2 g FW) were ground in a mortar and pestle. Amaranthine was extracted in 20mL methanol for 30 min followed by centrifugation for 10 min at 4°C and 12000 × g, and sampling of the supernatant. This process was repeated 3 times. Then the precipitation was soaked in 15mL distilled H2O for 30 min. After centrifugation for 10min at 4°C and 12000 × g, the absorption spectra of supernatant were measured spectrophotometrically (450-700nm) with a Visible-UV spectrophotometer (UV-2550, Shimadzu, Japan). The relative amount of amaranthine was indicated by OD198.

Statistical analysis: Statistical analyses were conducted by using Origin 7.5 software (Microsoft Company, Northampton, MA, USA) and tested by a Duncan test.

Results

Absorption spectra and the content of amaranthine:
The absorption spectra (450-700 nm) of methanol extracts from the red or green leaf discs of the same red cultivar of amaranth leaf were recorded (Fig. 1a). Red leaf discs showed a typical amaranthine absorption peak at 538 nm, whereas green leaf discs had no absorption peak in the same wavelength range. The relative amount of amaranthine in red leaf discs was significantly higher than that in green discs (p<0.01) (Fig. 1b).

Effects of photooxidative treatment on Fv/Fm, Fv'/Fm'
represents the maximal PS II quantum yield. Fig. 2 shows an obvious decreasing trend in Fv/Fm in red and green discs during 300 min of MV or H2O2 treatment. In comparison with the control, after 120 min of treatment, Fv/Fm had a more significant decrease under MV treatment than under H2O2 treatment. After 300 min treatment, Fv/Fm in green tissue decreased by 84.4% (MV) and 49.8% (H2O2) in comparison with control, whereas Fv/Fm in red tissue only decreased by 67.1% (MV) and 38.1% (H2O2). Less decline of Fv/Fm in red leaf discs than in green leaf discs indicated that PS II in red discs was better protected against photooxidation.

Effects of photooxidative treatment on Y (II) and qP:
Y (II) indicates the quantum yield of linear electron transport of PS II (Krall & Edward, 1992) and qP is a measure of the oxidation state of the primary quinine acceptor in PS II. During treatment with MV for 90 min, compared with control, Y (II) and qP showed a little increase in red leaf discs, though green leaf discs showed no significant difference (Fig. 3). Nevertheless, after continuous treatment from 120 to 300 min, Y (II) and qP decreased gradually, indicating that the reaction center of PS II and/or electron transport carriers might be damaged by MV treatment. The values of Y (II) and qP in green leaf discs were almost zero following 240 min treatment. In contrast, Y (II) and qP in red leaf discs remained at 0.084 and 0.201, respectively. However, treatment with 50 mmol L-1 of H2O2 did not result in a distinct effect on Y (II) and qP in red and green leaf discs as MV treatment did. Y (II) and qP in the 2 types of leaf discs increased first and maintained a level slightly higher than that of the control in 360 min duration of treatment.
Fig. 1. Absorption spectra and the contents of amaranthine in red or green discs sampled from the same leaf in a red cultivar of amaranth.

Fig. 2. Effects of MV or H$_2$O$_2$ treatment on $F_v/F_m$ in red or green discs from the same leaf in red cultivar of amaranth.
Effects of photooxidative treatment on NPQ and Y (NO): Non-photochemical quenching (NPQ) means the potential of the plant to dissipate excess excitation energy via heat, and Y (NO) indicates damage because of excess light energy in PS II (Kramer et al., 2004). During the process of photooxidative treatment of the two types of leaf discs (Fig. 4), the Y (NPQ) changes were opposite to those of Y (NO). The oxidative damage to the leaf discs by MV was significantly higher than that caused by H$_2$O$_2$ (p<0.01). In the initial MV treatment (0-90 min), Y (NPQ) of green and red leaf discs are the same as the control. At this time leaf discs may maintain PS II electron transport activity by adopting a higher photochemical quenching (qP) (Fig. 3). After 120 min treatment, Y (NPQ) had a sharp decline and Y (NO) rose markedly, suggesting that the antenna system had suffered serious oxidative damage.

Compared with the controls, after 120 min H$_2$O$_2$ treatment of the red and green leaf discs, Y (NPQ) decreased by 17.4% and 33.8%, Y (NO) increased by 19.1% and 28.5%, respectively, showing less damage in PS II than MV treatment. Compared with green leaf discs, in the subsequent 120-300 min of the MV and H$_2$O$_2$ treatments, red leaf discs had the higher Y (NPQ) and lower Y (NO) values, implying that the PS II of red leaf discs displayed better ability of resisting photooxidation than that of green discs.

Effects of photooxidation treatment on $F_{o}$: $F_{o}$ is a relative measure of the original density of excitation energy in pigments in PS II (Georgieva & Yordanov, 1993). Damage of the PS II reaction center or reversible inactivation can cause an increase in $F_{o}$ (Xu & Wu, 1996). In our experiment, the impact of H$_2$O$_2$ on $F_{o}$ was not large, but it had a clear upward trend after 120 min of MV treatment (Fig. 5). The $F_{o}$ of the green and red leaf discs increased by 30.3% and 15.1% compared with the control, respectively. The rise of $F_{o}$ means that the PS II reaction centre had suffered oxidative damage. In the later MV treatment (120-300 min), $F_{o}$ of the green leaf discs was always kept at a relatively high value, while that of the red discs only went up slowly. It suggested that thylakoid membranes of red leaf discs were more stable than in green leaf discs under oxidative stress.

Fig. 3. Effects of MV or H$_2$O$_2$ treatment on Y (II) and qP in red or green discs from the same leaf in a red cultivar of amaranth.
Fig. 4. Effects of MV or H$_2$O$_2$ treatment on Y (NPQ) and Y (NO) in red or green leaf discs of a red cultivar of amaranth.

Fig. 5. Effects of MV or H$_2$O$_2$ treatment on $F_0$ in the red or green discs from the same leaf in a red cultivar of amaranth.
Effects of photooxidation treatment on ETR: Non-cyclic photosynthetic electron transport rate (ETR) of PS II is a reflection of activity in PS II (Genty et al., 1989). As shown in Fig. 6, under exogenous ROS treatment, the patterns of ETR curves of both leaf discs resembled those of Y (II) curves (Fig. 3). ETR decreased significantly under MV treatment, but was slightly greater than the control under H$_2$O$_2$ treatment. In the MV photooxidative treatment, PS II electron transfer rate increased at first (0-90 min), but declined sharply thereafter, indicating that PS II in detached leaf tissue still had a capability of dynamic regulation in responding to mild photooxidation. Subsequently, after 240 min of MV treatment, the ETR of green leaf discs reached zero, whereas the ETR of the red leaf discs remained at $5.07 \pm 0.95 \mu$mol·m$^{-2}$·s$^{-1}$. Clearly, the electron transport function of PS II in red leaf discs was better protected than that in green discs under photooxidative conditions.

Content of Amaranthine in the two types of leaf discs after photooxidative treatment: Amaranthine content is the main difference of the phenotypic characteristics between the green and red leaf discs. In recent years, studies by Florian & Reinhold (2004) have pointed out that amaranthine have antioxidant properties similar to those of anthocyanin. Here in Fig. 7, under the treatment of photooxidation, amaranthine decreased in both types of leaf discs in a time-dependent manner, and amaranthine in red leaf discs declined more obviously. The decreasing rate of amaranthine was 0.0683 (Fig. 7a) and 0.0475 (Fig. 7b) by MV and H$_2$O$_2$, respectively. The result demonstrates that amaranthine, as an antioxidant in the red leaf discs, was consumed by reactions with ROS.

Fig. 6. Effects of MV or H$_2$O$_2$ treatment on ETR in red or green discs from the same leaf in a red cultivar of amaranth.

Fig. 7. Changes in the contents of amaranthine in red or green discs from the same leaf in a red cultivar of amaranth under MV (a) or H$_2$O$_2$ (b) treatment.
Discussion

As a C₃ plant, edible amaranth prefers warm weather and adapts to hot, dry or salt environments (Niu et al., 2004). Among the commonly cultivated varieties, amaranth can be divided into the green amaranth, red amaranth and a red cultivar of amaranth according to the shape and color of their stems and leaves (Guan et al., 1993). To ensure consistency of the material’s physiological state, we selected a red cultivar of amaranth as the material and collected red and green leaf discs, differing in amarathine content, from the same leaf.

Amaranthine is located in the vacuoles of the Amaranthaceae plants and is a biochrome belonging to Betacynins of Betalain. From the test of DPPH• antioxidant activity, it demonstrated a strong capacity as an antioxidant (Cai et al., 2001; Wang et al., 2006). Further, Cai et al., (2003) reported that seven kinds of sugar beet pigments purified from Amaranthaceae were better antioxidants than ascorbic acid, eldrin and acacia catechusic acid. Nevertheless amaranthine possessing high ROS-scavenging activity can scavenge organelle-derived ROS (Neill & Gould, 2003). Moreover, a possibility exists particularly in stress conditions, which some ROS (H₂O₂) leaking from cellular organelles diffuses across the tonoplast entering the vacuoles (Yamasaki et al., 1997). Our previous study also showed that the accumulation of amaranthine and osmoregulation substances in Amaranthus under high temperature may be the physiological basis of its heat tolerance (Shao et al., 2009). In this study, 1.93 fold higher amaranthine content in red leaf discs than in green leaf discs taken from the same leaf provided a suitable means of comparing the differential capability of antioxidation of red or green leaf discs.

At the beginning of treatment with MV or H₂O₂, the PS II activity was relatively stable, and then reduced. In the first 90 min of MV treatment, Fv/Fm did not change markedly and there was no significant enhancement in qP . ETR and Y(II) in both types of leaf discs. Fv rose slightly. As an exogenous electron acceptor, MV can produce superoxide anions, which can induce photooxidative stress in the cell itself. This is a possible reason that PS II can maintain a relative stable activity at the beginning of the MV treatment. These results are consistent with those of exogenous O₂−-induced chlorophyll fluorescence quenching in lettuce (Xu et al., 2002). With the photooxidative treatment, Fv/Fm, Y(II) and Y(NPQ) decreased significantly, while Y(NO) and Fv increase sharply, indicating that photooxidative stress appeared in the cell, and PS II reaction center and electron transfer mediators suffered irreversible oxidative damage after a long accumulation of O₂−. As in the MV treatment, PS II activity was nearly unaffected treated by 50 mmol·L⁻¹ of H₂O₂ at the beginning of 30 min. Fv/Fm was unchanged, whereas qP, ETR and Y (II) increased, while Y (NO) and Fv decreased. With H₂O₂ treatment, Y (NO) and Fv increased, accompanied by decreases in Y (NPQ) and Fv/Fm, while Y (II), qP and ETR were still slightly higher than the control. It indicates that H₂O₂ may play the same role of Hill reagents as MV that was used in the earlier treatment (Neubauer & Schreiber, 1989). Light conversion by PS II in leaf discs was enhanced by 50 mmol·L⁻¹ of H₂O₂, while excitation pressure of PS II aroused by excessive light energy was reduced. Thus, it could not affect photosynthetic electron flow. The view that the ROS have the dual-function of cellular oxidative toxicity and serving as signal molecules has been widely accepted now. Our results suggested that the relative stable state of PS II activity at the beginning of photooxidation treatment was due to induction of the cellular defense mechanisms by signal molecules of ROS.

According to the dynamic changes of chlorophyll fluorescence parameters in the green leaf discs, the sensitivity of PS II to photooxidation or the extent of oxidative damage was higher than that of the red leaf discs. After the MV and H₂O₂ treatment, the green leaf discs had higher values of initial fluorescence F₀ and Y (NO) (Figs. 4 and 5) than these of the red leaf discs, but the values of Y (NPQ) (Fig. 4) and ETR (under MV treatment, Fig. 6) were lower. After 240 min MV treatment, the Y (NPQ) and qP of the green leaf discs approached zero (Fig. 5), while these two parameters of the red leaf discs still maintained a certain value. These results indicated that the PSII of red leaf discs had a higher tolerance to oxidative damage than the green leaf discs. The difference in anti-photooxidation between red discs and green discs from the same leaf is that amaranthine may be used as a detoxification mechanism to ROS in red leaf discs (Fig. 7). It seems that the strong antioxidant of red leaf discs is similar to the anti-photooxidation of purple rice leaves with abundant anthocyanin (Peng et al., 2006). qP kept at relative higher level than control under H₂O₂ treatment, while MV induced an obvious decline in qP. On the other hand, there is little change in Fo under H₂O₂ treatment, whereas a significant increment in Fo under treatment of MV (Fig. 5). The results indicated that the sits of action or damage by MV and H₂O₂ may be different. Probably, MV caused severe damage in PSII reaction center, while H₂O₂ may not bring more damage to PSII.

In general, plants resistant to stresses depend on its cellular protecting mechanisms and restoration of damage capabilities. Similar to the function of anthocyanin, amaranthine, as natural anti-oxidants, also participates in the elimination of ROS to resist stresses intracellular oxidation as well as other anti-oxidants and antioxidant enzymes .Chlorophyll fluorescence quenching is a very complicated process. However, the mechanism of amaranthine underlying this process is still unclear. Previous studies reported that chloroplast ultrastructure of purple duckweed under stresses of acid rain suffered less damage than that of azolla duckweed (Lin et al., 2005). MV-induced inactivation of PS II is closely associated with damage of chloroplast ultrastructure of rice (Lin et al., 2009). Therefore, further studies are needed to explore whether the presence of amaranthine promotes stability of the membrane systems and chloroplast ultrastructure under photooxidation stress.

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