

APPENDICES TO THE THESIS ENTITLED

*The Origin and Evolution of the Tasmanian
Dolerites*

by

Janet M. Hergt

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APPENDIX 1.

SAMPLE LOCALITIES

All samples collected for analysis have been compiled in this appendix. The specimens are arranged in order of their catalogue numbers beginning with the Tasmanian collection. These are followed by the suite of samples from Portal Peak (Antarctica) generously provided by Prof. Gunter Faure and Teresa Mensing, and a group of samples from western Victoria and Kangaroo Island (southern Australia) provided by Prof. Dave Green.

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SAMPLE NUMBER	LATITUDE	LONGITUDE	ROCK-TYPE	LOCATION
84 101	42° 57.9' S	147° 12.5' E	Chilled Margin	0 m from top contact of sheet 2.5 km NE of Longley, south of the road. Red Hill.
84 102	42° 58.2' S	147° 12.0' E	Dolerite	Red Hill intrusion, 2 km NE Longley. Exposed in road cutting
84 103	43° 03.9' S	147° 13.3' E	Dolerite	Red Hill intrusion, summit of Red Hill
84 104	43° 03.2' S	147° 13.2' E	Dolerite	From northern slopes of Red Hill, 300 m ASL
84 105	43° 13.0' S	147° 50.7' E	Sediment	Exposed platform near Remarkable Cave, approx. 5 km south of Port Arthur
84 106	43° 13.0' S	147° 50.7' E	Chilled Margin	". Directly underlying the contact
84 107	"	"	Chilled Margin	". Less than 0.5 m from the contact
84 108	"	"	Chilled Margin	". Approx. 1 m from the contact
84 109	43° 07.0' S	147° 45.0' E	Chilled Margin	1.5 m below contact with Triassic sandstone, roadcutting 2.2 km SE of Nubeena
84 110	"	"	Dolerite	". 2 m below contact
84 111	43° 04.0' S	147° 50.0' E	Chilled Margin	Approx. 1.5 m below contact, roadcutting 1.45 km E of Koonya
84 112	41° 50.3' S	146° 49.6' E	Dolerite	NE edge of Great Lake, a few hundred metres NE of intake station
84 113	41° 50.4' S	146° 49.5' E	Dolerite	From a small gully beside the road at the intake station
84 114	41° 40.7' S	145° 57.6' E	Dolerite	Cradle Mountain, base of Little Horn columns along face track
84 115	41° 40.7' S	145° 57.4' E	Chilled Margin	Cradle Mountain, approx. 5 m above the contact
84 116	41° 41.0' S	145° 57.1' E	Dolerite	Summit of Cradle Mountain
84 117	41° 41.2' S	145° 57.2' E	Dolerite	From track leading to Cradle Summit, 1400 m ASL
84 118	41° 38.6' S	146° 42.9' E	Dolerite	Golden Valley, 1.75 km south of Exton junction
84 119	41° 32.0' S	147° 40.0' E	Dolerite	Ben Lomond, Ski Village. Sample from creek bed adjacent to amenities block
84 120	41° 31.0' S	147° 39.5' E	Dolerite	Ben Lomond, top of Jacobs Ladder. Roadside exposure
84 121	"	"	Dolerite	Ben Lomond, Hanging Corner on Jacobs Ladder
84 122	"	"	Dolerite	Ben Lomond, base of Jacobs Ladder
84 123	43° 04.3' S	147° 13.8' E	Dolerite	Red Hill intrusion, outcrop in Snug River approx. 100 m from eastern contact
84 124	43° 04.4' S	147° 14.2' E	Dolerite	Red Hill intrusion, Snug River, beneath the small bridge west of 84 123
84 125	43° 04.4' S	147° 14.1' E	Dolerite	Red Hill intrusion, Snug River, 100 m or so SE of the bridge
84 126	43° 04.4' S	147° 14.0' E	Dolerite	Red Hill intrusion, Snug River, continuing upstream
84 127	43° 04.5' S	147° 13.9' E	Dolerite	Red Hill intrusion, Snug River, approx the centre of the dyke
84 128	43° 04.7' S	147° 13.3' E	Chilled Margin	Red Hill intrusion, Snug River, 0.5 m below the western contact
84 129	"	"	Chilled Margin	". Only 30 cm from the contact
84 130	"	"	Dolerite	". Approx. 15 m from the western contact
84 131	43° 04.6' S	147° 13.9' E	Dolerite	O'briens Hill. immediately south of Snug River
84 132	43° 04.6' S	147° 13.8' E	Dolerite	O'briens Hill, approx the centre of the dyke on the road south of the river
84 133	43° 04.6' S	147° 13.9' E	Dolerite	Northern flank of O'briens Hill, south of the road above the river 190 m ASL
84 134	43° 04.7' S	147° 13.9' E	Dolerite	Keeping in about the centre of the dyke on O'briens Hill, 215 m ASL
84 135	43° 05.0' S	147° 13.9' E	Dolerite	Near the top of O'briens Hill beside a large clearing, 315 m ASL
84 136	43° 03.6' S	147° 13.1' E	Dolerite	Red Hill, northern part of dyke along Van Morey Rd. 240 m ASL
84 137	43° 03.6' S	147° 12.9' E	Dolerite	". Approx. 350 m east of the western margin. 260 m ASL
84 138	43° 02.3' S	147° 13.7' E	Chilled Margin	Red Hill intrusion, Small roadside exposure along Nierra Rd. 165 m ASL
84 139	43° 04.7' S	147° 13.1' E	Sediment	Red Hill, Snug River, a few metres west of the western contact
84 140	43° 17.0' S	147° 12.0' E	Chilled Margin	Nine Pin Point, dolerite transgressive to Permian sediments in roadcutting
84 141	43° 14.2' S	147° 08.2' E	Chilled Margin	Cygnet, approx. 1 m above a low angle contact with Triassic sandstone, roadcutting
84 142	43° 14.1' S	147° 08.1' E	Sediment	Cygnet, close to the chilled contact
84 143	43° 19.0' S	147° 03.0' E	Chilled Margin	Esperance Point close to contact
84 144	43° 01.8' S	147° 20.0' E	Chilled Margin	Blackmans Bay, 0 m from contact, 4 km S of town outskirts, near the home "Clarach".
84 145	"	"	Dolerite	" 5 m below contact.
84 146	"	"	Sediment	" 1 m above contact
84 147	"	"	Chilled Margin	" 0 m from contact about 20 m south of 84 144
84 148	"	"	Chilled Margin	". Basal contact of thin sheet above the main intrusion
84 149	42° 55.0' S	147° 20.0' E	Chilled Margin	Southern Expressway, 0 m from contact, adjacent to the Mt. Nelson turn-off
84 150	"	"	Chilled Margin	"
84 151	"	"	Sediment	". 1-1.5 m above the contact with the dolerite
84 152	42° 53.9' S	147° 14.6' E	Chilled Margin	Contact at the base of the Mt. Wellington sheet, 905 m ASL
84 153	42° 53.8' S	147° 14.6' E	Dolerite	Mt. Wellington, 920 m ASL
84 154	42° 53.5' S	147° 14.5' E	Dolerite	Mt. Wellington, 940 m ASL
84 155	42° 53.5' S	147° 14.4' E	Dolerite	Mt. Wellington, 958 m ASL
84 156	42° 53.5' S	147° 14.3' E	Dolerite	Mt. Wellington, 965 m ASL
84 157	42° 53.5' S	147° 14.2' E	Dolerite	Mt. Wellington, 985 m ASL
84 158	42° 53.5' S	147° 14.1' E	Dolerite	Mt. Wellington, 1000 m ASL, sample from "The Chalet"
84 159	42° 53.4' S	147° 14.0' E	Dolerite	Mt. Wellington, 1015 m ASL, first in situ dolerite above 84 158
84 160	42° 53.4' S	147° 13.8' E	Dolerite	Mt. Wellington, 1030 m ASL

SAMPLE NUMBER	LATITUDE	LONGITUDE	ROCK-TYPE	LOCATION
84 161	42° 53.4' S	147° 13.5' E	Dolerite	Mt. Wellington, 1060 m ASL, last in situ outcrop before the hairpin bend to the summit
84 162	42° 53.5' S	147° 13.3' E	Dolerite	Mt. Wellington, 1165 m ASL, blasted roadside outcrop
84 163	42° 53.6' S	147° 13.5' E	Dolerite	Mt. Wellington, 1185 m ASL
84 164	42° 53.6' S	147° 13.8' E	Dolerite	Mt. Wellington, 1210-1215 m ASL
84 165	42° 53.8' S	147° 14.2' E	Dolerite	Mt. Wellington, 1260-1265 m ASL
84 166	42° 53.9' S	147° 14.2' E	Dolerite	Mt. Wellington, at most 1 m below 1270 m trig. marker at summit
84 167	42° 55.3' S	147° 18.4' E	Chilled Margin	Southern Expressway, south of 84 149, block of Permian marl in the dolerite.
84 168	41° 50' S	146° 49.5' E	Chilled Margin	Great Lake borehole 5086 (Hydro-Electric Commission) 5" at 1048' 3"
84 169	"	"	Chilled Margin	Great Lake borehole 5086 (Hydro-Electric Commission) 6" at 1046' 9.5"
84 170	"	"	Chilled Margin	Great Lake borehole 5086 (Hydro-Electric Commission) 6.5" at 1044' 8"
84 171	"	"	Dolerite	Great Lake borehole 5086 (Hydro-Electric Commission) 2.5" at 1038' 8"
84 172	"	"	Dolerite	Great Lake borehole 5086 (Hydro-Electric Commission) 4" at 1029' 6"
84 173	"	"	Dolerite	Great Lake borehole 5086 (Hydro-Electric Commission) 4" at 990'
84 174	"	"	Dolerite	Great Lake borehole 5086 (Hydro-Electric Commission) 4.5" at 951'
84 175	"	"	Dolerite	Great Lake borehole 5086 (Hydro-Electric Commission) 4.5" at 907'
84 176	"	"	Dolerite	Great Lake borehole 5086 (Hydro-Electric Commission) 5" at 875'
84 177	"	"	Dolerite	Great Lake borehole 5086 (Hydro-Electric Commission) 2" at 792'
84 178	"	"	Dolerite	Great Lake borehole 5086 (Hydro-Electric Commission) 2" at 745'
84 179	"	"	Dolerite	Great Lake borehole 5086 (Hydro-Electric Commission) 5.5" at 698'
84 180	"	"	Dolerite	Great Lake borehole 5086 (Hydro-Electric Commission) 3" at 643' 6"
84 181	"	"	Dolerite	Great Lake borehole 5086 (Hydro-Electric Commission) 4" at 592'
84 182	"	"	Dolerite	Great Lake borehole 5086 (Hydro-Electric Commission) 4" at 542'
84 183	"	"	Dolerite	Great Lake borehole 5086 (Hydro-Electric Commission) 3" at 495'
84 184	"	"	Dolerite	Great Lake borehole 5086 (Hydro-Electric Commission) 4" at 448' 6"
84 185	"	"	Dolerite	Great Lake borehole 5086 (Hydro-Electric Commission) 3" at 399'
84 186	"	"	Dolerite	Great Lake borehole 5086 (Hydro-Electric Commission) 5.5" at 355' 6"
84 187	"	"	Dolerite	Great Lake borehole 5086 (Hydro-Electric Commission) 4.5" at 338'
84 188	"	"	Dolerite	Great Lake borehole 5086 (Hydro-Electric Commission) 6" at 278'
84 189	"	"	Dolerite	Great Lake borehole 5086 (Hydro-Electric Commission) 4" at 211'
84 190	"	"	Dolerite	Great Lake borehole 5086 (Hydro-Electric Commission) 3" at 165'
84 191	"	"	Dolerite	Great Lake borehole 5086 (Hydro-Electric Commission) 4" at 95'
84 192	"	"	Dolerite	Great Lake borehole 5086 (Hydro-Electric Commission) 5" at 48'
84 193	42° 34.0' S	147° 49.5' E	Sediment	5 km west of Orford, Large quarry near Black Bridge Creek
84 194	42° 01.5' S	147° 30.0' E	Dolerite	1.5 km north of Ross township in railway cutting, 15m or so from contact
84 195	42° 12.0' S	147° 23.5' E	Chilled Margin	1.5 km north of Antill Ponds along Midland Hwy. 0 m from bottom contact
84 196	42° 24.3' S	147° 15.6' E	Chilled Margin	North of Bishton Creek on the Midland Hwy (north of Spring Hill) 1 m from contact
84 197	42° 26.5' S	147° 08.5' E	Dolerite	Apsley, small columns of different orientation described by S Forsyth Dept. Mines
84 198	42° 17.4' S	147° 03.6' E	Dolerite	Bryans Creek 2.5 km W of Woods Quoin. "Rosette rock" described by S Forsyth.
84 199	42° 11.5' S	147° 07.0' E	Chilled Margin	3 km W of Lake Crescent, hornfels "skins" nearby indicate 0 m from contact
84 200	42° 08.0' S	147° 13.0' E	Chilled Margin	3 km NE of Interlaken; pavements with hornfels skins close-by. 1 m from contact
84 201	42° 09.0' S	147° 10.0' E	Dolerite	Interlaken Canal, no contact exposed; ? m from contact
84 202	"	"	Dolerite	Woods Lake, fine-grained dykelets intrude coarser dolerite. Too altered, not analysed.
84 203	42° 26.4' S	147° 14.3' E	Chilled Margin	0 m from contact; small quarry 5.5km S of Spring Hill on Nth side of old Midland Hwy.
84 204	42° 24.6' S	147° 15.7' E	Dolerite	Spring Hill, upper level of rd. cutting. Fine dolerite appears to intrude coarser rock
84 205	41° 14.5' S	146° 33.7' E	Chilled Margin	0 m from contact with Triassic sandstone, E of Harford just west of Rubicon bridge
84 206	43° 01.6' S	147° 15.5' E	Chilled Margin	Margate Borehole 1. Core samples courtesy Dept. Mines Tas. 306.17 m
84 207	"	"	Dolerite	Margate Borehole 1. Core samples courtesy Dept. Mines Tas. 308.15 m
84 208	"	"	Dolerite	Margate Borehole 1. Core samples courtesy Dept. Mines Tas. 310.45 m
84 209	"	"	Dolerite	Margate Borehole 1. Core samples courtesy Dept. Mines Tas. 311.20 m
84 210	"	"	Dolerite	Margate Borehole 1. Core samples courtesy Dept. Mines Tas. 314.20 m
84 211	"	"	Dolerite	Margate Borehole 1. Core samples courtesy Dept. Mines Tas. 320.20 m
84 212	"	"	Dolerite	Margate Borehole 1. Core samples courtesy Dept. Mines Tas. 331.20 m
84 213	42° 46' S	147° 13' E	Chilled Margin	Granton Borehole 1. Core samples courtesy Dept. Mines Tas. 544.07 m
84 214	"	"	Chilled Margin	Granton Borehole 1. Core samples courtesy Dept. Mines Tas. 544.88 m
84 215	"	"	Dolerite	Granton Borehole 1. Core samples courtesy Dept. Mines Tas. 546.00 m
84 216	"	"	Dolerite	Granton Borehole 1. Core samples courtesy Dept. Mines Tas. 553.00 m
84 217	"	"	Dolerite	Granton Borehole 1. Core samples courtesy Dept. Mines Tas. 555.79 m
84 218	"	"	Dolerite	Granton Borehole 1. Core samples courtesy Dept. Mines Tas. 558.84 m
84 219	"	"	Dolerite	Granton Borehole 1. Core samples courtesy Dept. Mines Tas. 560.00 m
84 220	"	"	Dolerite	Granton Borehole 1. Core samples courtesy Dept. Mines Tas. 561.00 m

SAMPLE NUMBER	LATITUDE	LONGITUDE	ROCK-TYPE	LOCATION
84 221	42° 40' S	147° 07' E	Chilled Margin	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 216.38 m
84 222	"	"	Chilled Margin	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 219.23 m
84 223	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 227.38 m
84 224	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 235.24 m
84 225	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 243.47 m
84 226	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 250.38 m
84 227	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 265.61 m
84 228	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 283.08 m
84 229	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 298.65 m
84 230	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 315.41 m
84 231	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 331.49 m
84 232	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 347.45 m
84 233	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 361.04 m
84 234	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 379.45 m
84 235	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 392.46 m
84 236	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 409.53 m
84 237	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 426.76 m
84 238	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 434.10 m
84 239	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 445.99 m
84 240	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 456.18 m
84 241	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 464.46 m
84 242	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 475.12 m
84 243	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 486.33 m
84 244	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 497.55 m
84 245	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 504.32 m
84 246	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 516.14 m
84 247	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 528.09 m
84 248	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 538.05 m
84 249	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 549.27 m
84 250	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 559.30 m
84 251	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 569.94 m
84 252	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 579.13 m
84 253	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 588.78 m
84 254	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 598.95 m
84 255	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 609.25 m
84 256	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 621.10 m
84 257	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 630.05 m
84 258	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 641.32 m
84 259	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 651.10 m
84 260	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 662.11 m
84 261	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 668.76 m
84 262	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 681.81 m
84 263	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 693.12 m
84 264	41° 39' S	148° 06' E	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 67.69 m
84 265	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 82.75 m
84 266	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 97.76 m
84 267	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 112.75 m
84 268	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 121.75 m
84 269	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 133.78 m
84 270	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 142.79 m
84 271	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 151.75 m
84 272	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 163.81 m
84 273	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 178.81 m
84 274	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 187.82 m
84 275	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 199.83 m
84 276	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 208.81 m
84 277	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 220.81 m
84 278	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 229.82 m
84 279	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 241.78 m
84 280	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 253.84 m

SAMPLE NUMBER	LATITUDE	LONGITUDE	ROCK-TYPE	LOCATION
84 281	41° 39' S	148° 06' E	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 262.84 m
84 282	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 274.86 m
84 283	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 283.86 m
84 284	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 295.87 m
84 285	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 301.87 m
84 286	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 313.90 m
84 287	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 322.92 m
84 288	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 334.93 m
84 289	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 346.92 m
84 290	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 355.94 m
84 291	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 367.96 m
84 292	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 379.97 m
84 293	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 388.97 m
84 294	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 397.99 m
84 295	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 406.99 m
84 296	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 416.00 m
84 297	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 425.01 m
84 298	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 443.00 m
84 299	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 449.00 m
84 300	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 451.95 m
84 301	"	"	Chilled Margin	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 452.60 m
84 302	42° 51' S	147° 15' E	Dolerite	Glenorchy Borehole. Core samples courtesy Dept. Mines Tas. 199.95 m
84 303	"	"	Dolerite	Glenorchy Borehole. Core samples courtesy Dept. Mines Tas. 201.96 m
84 304	"	"	Dolerite	Glenorchy Borehole. Core samples courtesy Dept. Mines Tas. 203.50 m
84 305	"	"	Dolerite	Glenorchy Borehole. Core samples courtesy Dept. Mines Tas. 205.94 m
84 306	"	"	Dolerite	Glenorchy Borehole. Core samples courtesy Dept. Mines Tas. 210.44 m
84 307	"	"	Dolerite	Glenorchy Borehole. Core samples courtesy Dept. Mines Tas. 217.40 m
84 308	"	"	Dolerite	Glenorchy Borehole. Core samples courtesy Dept. Mines Tas. 234.52 m
84 309	"	"	Dolerite	Glenorchy Borehole. Core samples courtesy Dept. Mines Tas. 257.33 m
84 310	"	"	Dolerite	Glenorchy Borehole. Core samples courtesy Dept. Mines Tas. 286.36 m
84 311	"	"	Dolerite	Glenorchy Borehole. Core samples courtesy Dept. Mines Tas. 304.34 m
84 312	"	"	Dolerite	Glenorchy Borehole. Core samples courtesy Dept. Mines Tas. 324.84 m
84 313	"	"	Dolerite	Glenorchy Borehole. Core samples courtesy Dept. Mines Tas. 345.80 m
84 314	"	"	Dolerite	Glenorchy Borehole. Core samples courtesy Dept. Mines Tas. 366.06 m
84 315	"	"	Dolerite	Glenorchy Borehole. Core samples courtesy Dept. Mines Tas. 387.71 m
84 316	"	"	Dolerite	Glenorchy Borehole. Core samples courtesy Dept. Mines Tas. 405.69 m
84 317	"	"	Dolerite	Glenorchy Borehole. Core samples courtesy Dept. Mines Tas. 424.89 m
84 318	"	"	Dolerite	Glenorchy Borehole. Core samples courtesy Dept. Mines Tas. 447.75 m
84 319	"	"	Dolerite	Glenorchy Borehole. Core samples courtesy Dept. Mines Tas. 460.55 m
84 320	"	"	Dolerite	Glenorchy Borehole. Core samples courtesy Dept. Mines Tas. 480.67 m
84 321	"	"	Dolerite	Glenorchy Borehole. Core samples courtesy Dept. Mines Tas. 482.50 m
84 322	"	"	Dolerite	Glenorchy Borehole. Core samples courtesy Dept. Mines Tas. 489.20 m
84 323	"	"	Chilled Margin	Glenorchy Borehole. Core samples courtesy Dept. Mines Tas. 490.42 m
84 324	"	"	Chilled Margin	Glenorchy Borehole. Core samples courtesy Dept. Mines Tas. 491.95 m
87 85	83° 50' S	165° 40' E	Sediment	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 18.0 m
87 86	"	"	Sediment	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 18.5 m
87 87	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 19.0 m
87 88	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 19.5 m
87 89	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 21.0 m
87 90	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 24.0 m
87 91	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 27.0 m
87 92	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 30.0 m
87 93	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 33.0 m
87 94	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 36.0 m
87 95	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 39.0 m
87 96	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 42.0 m
87 97	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 45.0 m
87 98	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 48.0 m
87 99	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 51.0 m
87 100	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 54.0 m

SAMPLE NUMBER	LATITUDE	LONGITUDE	ROCK-TYPE	LOCATION
87 101	83° 50'S	165° 40' E	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 57.0 m
87 102	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 60.0 m
87 103	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 63.0 m
87 104	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 66.0 m
87 105	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 69.0 m
87 106	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 72.0 m
87 107	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 75.0 m
87 108	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 78.0 m
87 109	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 81.0 m
87 110	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 84.0 m
87 111	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 87.0 m
87 112	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 90.0 m
87 113	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 93.0 m
87 114	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 96.0 m
87 115	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 99.0 m
87 116	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 102.0 m
87 117	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 105.0 m
87 118	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 111.0 m
87 119	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 117.0 m
87 120	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 123.0 m
87 121	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 129.0 m
87 122	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 135.0 m
87 123	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 141.0 m
87 124	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 145.5 m
87 125	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 147.0 m
87 126	"	"	Dolerite	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 148.0 m
87 127	"	"	Sediment	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 149.5 m
87 128	"	"	Sediment	Portal Peak Antarctica. Samples courtesy Gunter Faure and Teresa Mensing. 152.0 m
87 129	37° 35.9' S	141° 43.9' E	Alkali Basalt	Koroite Hill, 3 km E of Coleraine Western Victoria. Sample courtesy Tas. Uni.
87 130	"	"	Tholeiitic Basalt	Koroite Hill, 3 km E of Coleraine Western Victoria. Sample courtesy Tas. Uni.
87 131	"	"	Tholeiitic Basalt	Koroite Hill, 3 km E of Coleraine Western Victoria. Sample courtesy Tas. Uni.
87 132	37° 12.9'S	141° 30.0'E	Alkali Basalt	Harrow, approx. 50 km N Coleraine Western Victoria. Sample courtesy Tas. Uni.
87 133	35° 38.3' S	137° 28.9' E	Tholeiitic Basalt	2.5 km SE Wisanger Kangaroo Is. South Australia. Sample courtesy Tas. Uni.
87 134	35° 44.3' S	138° 00.5' E	Tholeiitic Plug	8 km E Penneshaw, Kangaroo Is. South Australia. Sample courtesy Tas. Uni.
87 135	35° 39.0' S	137° 37.3' E	Tholeiitic Basalt	Quarry sample 1 km W Kingscote Kangaroo Is. S Australia. Sample courtesy Tas. Uni.

APPENDIX 2.

SAMPLE PREPARATION AND ANALYTICAL TECHNIQUES

Various techniques have been employed in the analysis of close to 300 rock samples. The sample preparation and analytical methods are outlined in this appendix and are divided into the following sections:

1. Crushing
2. Determination of H_2O+ , H_2O- , CO_2 and FeO
3. Preparation of glass discs and pressed powder pellets and their analysis by XRF spectrometry
4. Determination of trace element concentrations by INAA
5. Determination of trace element concentrations by SSMS
6. Determination of Sr and Nd isotopic compositions
7. Determination of Pb isotopic compositions
8. Determination of O isotopic compositions

2.1 CRUSHING

The samples collected from outcrops were reduced to gravel sized chips using an hydraulic splitter. Care was taken in removing surfaces bearing metal from hammer blows, rare carbonate veins, and (in the case of many of the chilled margins) rust colored alteration along joint planes.

Specimens obtained from drill core were first ground on a diamond lap to remove any surface contamination produced during drilling, and the saw marks developed during halving of the core (in the case of samples obtained from the Tasmanian Department of Mines). It was also necessary to clean the Antarctic samples in this manner as they consisted of small sawn blocks; in this case splitting to a smaller size was not necessary.

Where there were no constraints on the sample size, approximately 1 kg of the chips were crushed and an aliquot further ground to powder in a tungsten carbide swing mill. In addition to the Antarctic specimens (a maximum of ~10-20 g for most samples), many of the core samples were small enough to be crushed and ground into powder in one step. Between samples the swing mills were cleaned by crushing washed quartz pebbles and rinsing the mills in alcohol.

Powders prepared in this manner were used in all whole-rock analyses with only two exceptions. Because crushing in tungsten carbide contributes Co and Ta to the sample (as well as W), the Co analyses performed on chilled margin samples were obtained by taking a separate aliquot of chips and crushing them in steel mills. The Pb whole-rock isotopic compositions were also obtained from chips saved from the original hand-specimen.

2.2 DETERMINATION OF H_2O+ , H_2O- , CO_2 AND FeO

The method of determining the amount of H_2O- (the hygroscopic water absorbed onto grain surfaces), H_2O+ (the water held in silicate structures) and CO_2 (also held in the matrix)

was by gravimetric analysis. In the procedure used, approximately 0.5 g of rock powder was accurately weighed into a platinum boat (also of accurately known weight) and then placed into an oven set at $\sim 100\text{-}110^\circ\text{C}$ for several hours (or overnight). This removed the hygroscopic water and the loss of weight (H_2O -) was recorded.

The remaining water and CO_2 were liberated at higher temperatures and were determined after removal of H_2O -. The platinum boat was placed in a combustion tube within a furnace (set at $\sim 1000^\circ\text{C}$) for 30 minutes. A stream of nitrogen was passed through the tube to flush the volatiles liberated from the sample. Two absorption tubes were assembled at the end of the combustion tube; the first containing powdered P_2O_5 mixed with quartz granules to absorb water; the second containing sodium hydroxide on support granules which absorbs CO_2 . The tubes were weighed between samples and a blank was measured at the beginning and end of each day. The increase in weight of each tube was recalculated to give the weight percent of H_2O and CO_2 in each sample. These analyses were performed only once for each sample.

The total Fe content determined by XRF analysis as Fe_2O_3 (see later discussion) was resolved into FeO and Fe_2O_3 by a method involving dissolution and titration. In this method approximately 0.5 g of rock powder was weighed into a platinum crucible (containing a small platinum wire to prevent "bumping" during heating) and was dampened with recently boiled distilled water. A mixture of HF(conc) and $\text{H}_2\text{SO}_4(1:1)$ was poured into the crucible (~ 10 ml) and then covered by a platinum lid. The crucible was carefully heated over a bunsen flame until fuming began, at which point the heat was reduced and the mixture was left to react for 3 minutes. The contents of the crucible was carefully rinsed into a beaker containing a solution of boric acid, sulphuric acid, phosphoric acid and a few drops of 0.5% barium diphenylamine sulphonate indicator. This was quickly titrated against a standard solution of $\text{K}_2\text{Cr}_2\text{O}_7$ to determine FeO. Fe_2O_3 was then calculated by difference.

All results were duplicated to eliminate possible errors resulting from incomplete dissolution or oxidation of the solution prior to titration. To check the results further, several samples were re-analysed using a similar but more sophisticated technique developed by Kiss (1977), and the results agreed to within expected error (i.e. $\sim 2\text{-}3\%$).

2.3 PREPARATION OF GLASS DISCS AND PRESSED POWDER PELLETS AND THEIR ANALYSIS BY XRF SPECTROMETRY.

Most of the geochemical data were determined using XRF spectrometry. The major elements were obtained by analysis of fused glass discs and the trace elements were determined from analysis of pressed powder pellets.

The trace element pellets were made by combining ~ 2.5 g of sample powder with a few drops of PVA solution (to help bind the powder) and gently pressed using a hand-held assembly to produce a circular disc. Boric acid powder was packed around and on top of each disc which was then pressed at higher pressure into a rigid pellet. After drying for a few minutes under a heat lamp the samples were ready for analysis.

For the major element analysis fused glass discs were used. Although the methods used are similar to those of Norrish and Hutton (1969), some important differences have been made and elaboration is useful. The glass produced for analysis contains close to 0.28 g of the sample

powder and 1.5 g of the lithium borate flux. Instead of painstakingly weighing exact amounts of these powders, the balance was linked to a computer and disc-drive. The weights were recorded onto discs for later data reduction, saving considerable time and effort in sample preparation. In some labs an oxidizing agent such as NaNO_3 or LiNO_3 is also added at this stage. Instead, the sample-flux mixtures were sintered at 180°C for 15 minutes prior to fusing, and duplicated results using the two methods show good agreement.

After 15 minutes sintering, the samples were transferred to a second furnace (800°C) for 15 minutes. At this temperature the mixture melts but does not necessarily homogenise. For most of the samples, the glasses were remelted and homogenised by being held and agitated in a flame for ~ 10 minutes prior to pouring the melt droplet onto a graphite disc and pressing it flat with an aluminium plunger. The Portal Peak, western Victorian and Kangaroo Island samples were analysed later than most other samples. Instead of holding the samples in a flame to homogenise them, a rocking-furnace was used and the crucibles were taken from this furnace, and the glass poured and pressed into discs.

The theory on which XRF analysis is based is detailed in a number of excellent textbooks and publications and is not dealt with here. The largest error involved in this technique arises from absorption of the fluorescent x-rays by the sample. Clearly, if not all of the x-rays released by atoms of the element measured manage to pass out of the sample, then the concentration calculated will be low compared to the true value. Although the error introduced by absorption decreases with decreasing abundance of the element, proper correction for absorption requires the determination of mass absorption coefficients.

In this study, direct measurements were made (in duplicate) of the Rb and Sr absorption coefficients for each sample analysed. These absorption coefficients were applied to elements with fluorescent wavelengths between 0.7 and 1.0\AA . Other trace elements were corrected by calculating mass absorption coefficients, from major element analyses and published absorption coefficients.

Other errors involved in the determination of concentrations by XRF include corrections for high or non-linear backgrounds (produced by scatter) and interferences by spectra from other elements. These have been corrected using the methods of Norrish and Chappell (1967) by a program developed by Dr. B.W. Chappell.

Calibration of the two XRF spectrometers used in the major and trace element analyses (Siemens SRS300, and Philips PW1400 respectively) was originally performed using synthetic standards prepared using chemicals of high purity, and well documented international standards. These are periodically re-analysed in addition to a number of "in-house" standards calibrated for more routine use.

2.4 DETERMINATION OF TRACE ELEMENT CONCENTRATIONS BY INAA

Samples chosen for instrumental neutron activation analysis (INAA) were packed into small polythene capsules which were sealed by heating. The weight of sample used was generally ~ 0.3 g. No special treatment of the sample powder was required; however, plastic gloves were worn at all times to ensure that the vials remained free from surface contamination.

Each batch included fourteen unknowns and a standard (in this case the in-house basalt standard). These were irradiated in a flux of $\sim 4 \times 10^{12}$ n/cm²/s for 24 hours and the known Fe contents of the samples (from XRF) were used as flux monitors.

After a week of "cooling" the gamma-ray spectra were acquired using a germanium low-energy photon spectrometer for Ce, Nd, Sm, Eu, Gd, Tb, Yb, Lu, Ta, and a lithium-drifted germanium detector for La, Yb, Lu, Ta, Eu, Hf, and Cs (Table 2.1; from Chappell and Hergt, in prep.).

The counts were reduced to concentrations using a computer program developed by Dr. B.W. Chappell, and multiple photopeak data acquired for a single element were weighted inversely as the variance of the errors.

Table 2.1 Instrumental conditions for INAA analysis and coefficient of variation ($\sigma\%$) of the counting error of the various determinations

Count	1	2	3	4	5	6
Approx. days after irradiation	3.5-5	8-11	15-30	30-60	450	450
Detector*	Planar	Coaxial	Planar	Coaxial	Planar	Coaxial
Approx. length of count (sec.)	7500	16000	20000	20000	20000	20000
Elements keV						
Na 1369		0.6				
Fe 1099				0.17		
Fe 1292				0.19		
Sc 889				0.06		0.15
Sc 1121				0.06		0.17
Cr 320				0.8		
Co 1173				0.21		0.09
Co 1333				0.22		0.09
Cs 605				21		1.7
Cs 796				9.1		2.1
Ba 124			3.9			
Ba 216		13				
Ba 496		12				
La 329		2.6				
La 487		1.5				
La 816		3.6				
La 1597		0.6				
Ce 145			1.1			
Nd 91			2.4			
Sm 103	0.4		0.5			
Eu 122			1.2		(0.19)	
Eu 779				9.5		1.3
Eu 1086						1.7
Eu 1408				2.1		0.7
Gd 97					4.3	
Gd 103					5.2	
Tb 87			3.2			
Tb 299				11		
Ho 81	6.0					
Yb 114	5.3					
Yb 177			5.3			
Yb 396		2.2				
Lu 113	8.3					
Lu 208	13	1.7	2.4			
Hf 133			2.1			
Hf 482				3.5		
Ta 100			41		12	
Ta 152			68			
Ta 1221				13		
Th 312				3.2		
U 106	21					

Footnotes to table 2.1

* Coaxial detector was an ORTEC GEM10175 with 17.5% efficiency and resolution of 1.85 keV at 1333 keV. For counts 1 and 3 a 6 cm diameter and for count 5 a 36 mm diameter ORTEC detectors were used, with resolutions of 500 and 700 eV, respectively, at 122 keV. With the 16 mm planar detector that we now have available, better precisions can be obtained on count 3 without loss of resolution.

Data are from the first of six replicate analyses of the Mt Wellington chilled margin sample 84-152. Errors vary slightly from count to count depending on the precise time of counting and the length of that count.

Data show which photopeaks are used for the analysis of each element and when. Numbers are the per cent. counting error in net intensity peak minus background and are a component of the total final error.

An estimate of the absolute counting errors can be obtained by combining these data with the mean values.

The very precise Eu value of 0.19% at 122 keV on count 5 was not used since at the time the resolution of the large planar detector was poor and slightly variable, and other good data for Eu are available.

These data should not be taken as a guide to counting errors in samples of different compositions. e.g. the high Sc in the chilled margins means that count 4 is much less efficient for all other elements measured at that time.

2.5 DETERMINATION OF TRACE ELEMENT CONCENTRATIONS BY SSMS

The trace element compositions of seven samples from western Victoria and Kangaroo Island (supplied by Prof. D.H. Green) were analysed using spark-source mass spectrometry. In this method, each sample is well-mixed with a Lu carbide standard in a ratio of 1:1 (using 100 mg of each powder) in a carefully cleaned agate mill. Electrodes are prepared from this mix by pressing the powder into a ~1 mm diameter hole bored through a polypropylene slug.

After initial sparking (to remove any possible surface contamination), a photoplate is placed in position and a series of 15-16 graded exposures recorded. The species recorded on the photographic plate are separated in the spectrometer according to their mass:charge ratio and include elemental ions, carbides, and polyatomic ions; care is used when reading the plates and reducing the data to eliminate any interferences. The intensities of the lines recorded on the plates (relative to the Lu internal standard) are obtained using a densitometer and are related to concentration by reference to a calibration curve developed by Dr. S.R. Taylor using BCR-1.

2.6 DETERMINATION OF Sr AND Nd ISOTOPIC COMPOSITIONS

Most of the Sr and Nd isotopic data were obtained from spiked samples despite the fact that elemental concentrations of Sr, Rb, Nd and Sm were obtained by other techniques (i.e. XRF and INAA). Aliquots from the same jars of powder used in the geochemical analyses were used for isotopic analysis. Between 50 and 200 mg of sample powder were used, and in most cases (after spiking with ^{85}Rb - ^{84}Sr and ^{147}Sm - ^{150}Nd spike solutions) the powders were dissolved in open beakers. Exceptions are the differentiated samples from the Red Hill Dyke, and five contact sediments, which were dissolved in teflon bombs kept at ~200°C for 2 days.

In most cases, the samples were first dissolved in HF (~1ml), dried and redissolved using a further quantity of HF(conc) with a few drops of HClO_4 (conc) added. This mixture was dried down and allowed to fume for at least an hour to remove any excess perchloric acid and fluorinated silicate. Having converted the fluorides to chlorates, ~2 ml of 6N HCl was

added to convert the chlorates to chlorides. The solution was dried to stickiness and the residue re-dissolved in 2.5 ml of 1N HCl. The solution was then transferred to a labelled centrifuge tube and spun down for several minutes.

These solutions were loaded onto 5 g cation exchange columns conditioned with 1N HCl. The Rb and Sr cuts from this column were dried and passed through 2 g clean-up columns, while the REE (collected in one cut) were dried and passed through pressurised columns using 0.2M methyl-lactic acid as the eluent. The Sm and Nd cuts were collected using an automatic rotating turntable linked to a drop-counter.

In addition to the spiked samples measured, twelve unspiked samples were dissolved to obtain Nd compositions of higher precision using the Finnigan-Mat 261 multiple collector mass spectrometer. These samples had been analysed previously (spiked) and an additional ~50 mg of each were processed in the manner described above. After removing the REE cuts from the 5g columns however, the samples were dried, redissolved in 0.2N HCl and passed through HDEHP columns.

For the spiked samples, the purified Rb, Sr, Sm and Nd cuts were run as the metal species using a rhenium triple-filament assemblage. The samples were loaded onto the side-filaments in distilled 1N HCl, and then analysed using the MSZ mass spectrometer (Clement and Compston, 1972). Data acquisition was performed automatically using programs developed at the ANU. The isotopic ratios were obtained by peak-switching and normalised to correct for mass fractionation ($^{86}\text{Sr}/^{88}\text{Sr}=0.1194$, $^{146}\text{Nd}/^{142}\text{Nd}=0.636151$). In general, 98 $^{87}\text{Sr}/^{86}\text{Sr}$ ratios, or 156 $^{143}\text{Nd}/^{144}\text{Nd}$ ratios are obtained for each particular analysis. No blank corrections were required for any of the ratios.

For the unspiked Nd analyses using the Finnigan-Mat multiple collector mass spectrometer, the intensities of the isotopes 142, 143, 144, 145, 146, and 150 are measured simultaneously. The samples were analysed twice (automatically) to examine how the reproducibility of the results compared with the precision calculated for individual runs. The automatic data acquisition program has been set up to collect 10 blocks of data per analysis, with 12 sets of ratios collected in each block. The precision and accuracy are discussed in Appendix 3.

2.7 DETERMINATION OF Pb ISOTOPIC COMPOSITIONS

Unlike the other isotopic analyses, the four samples chosen for Pb-isotopic analysis were processed using small chips of rock sample rather than the finely-crushed powders. A small piece of each hand-specimen was shattered in a Mg-steel piston assemblage, and interior fragments were hand-picked using clean tweezers. These chips (~0.5 g) were dissolved using HF and HClO₄ in teflon bombs and the residue converted to chlorides using 6N HCl. The solutions were split and a small portion spiked with mixed ^{235}U - ^{208}Pb spike. The Pb was extracted from each split by passing the sample through 2 g Dowex-1 anion exchange columns using distilled solutions of HBr and HCl. The U was extracted from the spiked split by passing the dried cut (obtained from the 2 g columns) down teflon micro-columns using 10 N HCl, saturated ammonium sulphate solution, and ultra-pure water.

The unspiked Pb analyses were performed in duplicate (i.e. the Pb collected was loaded

onto two separate filament beads) to check the reproducibility. Each aliquot of lead was loaded onto an outgassed single rhenium filament using silica-gel and phosphoric acid, and the U was loaded onto the same type of filament using a Ta-oxide slurry.

The main source of error in the Pb results is derived from variable mass fractionation and no correction for blank has been applied. Multiple analyses of NBS-981 were performed and a mass fractionation correction factor was derived by normalising the results to the corrected values of Catanzaro *et al.* (1968). The average correction factor obtained (based on 5 analyses of the standard) is 0.123% per mass unit.

2.8 DETERMINATION OF OXYGEN ISOTOPIC COMPOSITIONS

The technique employed in extracting the oxygen from each of 18 samples and converting it to CO₂ is similar to that described by Clayton and Mayeda (1963). Using this method, 15 mg of sample powder are weighed and carefully transferred into bombs under a back-pressure of Ar. An excess of BF₅ gas is measured into each of the 10 bombs which are then sealed from the rest of the line. Heating jackets are placed over the bombs and the samples are heated overnight (800°C).

The oxygen released from each digested sample is released into a chamber and converted to CO₂ by reaction with a heated carbon electrode. These CO₂ samples are removed from the line by freezing each one into a clean glass tube which is cut from the line with a oxy-acetylene torch.

The mass-spectrometry of the 18 samples analysed in this study was performed by Dr. A.R. Chivas and the results processed and normalised to standard mean ocean water (SMOW) using a computer program developed at ANU. Each gas sample was analysed in duplicate, and repeat analyses were performed using separate sample digestions for all samples. The 2 σ uncertainties for the compositions obtained are less than or equal to $\pm 0.3\text{‰}$ and are listed beside the results in Appendix 5.

APPENDIX 3.

DETERMINATION OF ANALYTICAL PRECISION AND ACCURACY

In geochemical and isotopic research, a high level of accuracy in analytical results is important. To achieve this, laboratories make frequent analysis of international and in-house standards. This helps detect any technical problems, such as instrument drift or errors associated with a change in sample preparation.

The study of a rather homogeneous suite of samples requires that the data be of the highest precision possible. Only in this way can subtle differences in petrogenetic history be recognised and properly assessed.

Geochemical data published for dolerites from Tasmania indicate that the magmas emplaced in the Jurassic were of very uniform composition, at least with respect to the major elements (Edwards, 1942; McDougall, 1962). Also, despite the significant range in initial $^{87}\text{Sr}/^{86}\text{Sr}$ ratios, the Pb isotopic compositions show little variation, and two samples (showing a range in Sr-isotopic compositions) analysed for their Nd-isotopic signatures were identical within error (Allègre *et al.*, 1982).

The apparent homogeneity of the geochemical and isotopic compositions may either be real, or a result of poor precision of the measurements. To better evaluate these possibilities, it was considered important to determine the precision (as well as the accuracy) of the data obtained in this study

In order to determine the precision of individual (in the case of most of the isotopic data) or duplicate (in the case of the geochemical data) analyses, multiple measurements were made for particular samples. The standard deviation was then used to calculate the reproducibility expected for the other results. It is most useful to determine the precision using material of similar composition to the unknowns. For many trace elements, the chilled margin sample from Mt. Wellington (84-152) was chosen.

To evaluate the accuracy of measurements, international standards were analysed in triplicate (or more) and the averages compared to data available in the literature. In addition, the different analytical techniques employed in this study have facilitated a comparison between elemental data obtained by more than one method (e.g. Rb data obtained by XRF may be compared to the results from isotope dilution for the same samples).

Presentation of the results has been divided into two parts; the determination of precision and accuracy (subdivided into the geochemical and isotopic results), and the comparison between concentration data obtained by different instrumental techniques.

3.1 DETERMINATION OF ANALYTICAL PRECISION AND ACCURACY

3.1.1 Precision

The precision of concentration measurements has been determined using a combination of the results obtained for W-2, and additional data obtained from the chilled margin dolerite 84-152, by calculating the standard deviation of multiple analyses. The precision of major

element compositions is based on the eight analyses of W-2 (6 in the case of FeO wet-chemical analysis). Precision of the XRF trace element analyses has not been determined; however, the uncertainty associated with the INAA trace element data has been calculated using six separate analyses of 84-152. The results of these analyses are given in Table 3.1.

Table 3.1 Six replicate analyses of chilled margin sample 84-152

Element	1	2	3	4	5	6	Mean $\pm\sigma$	$\sigma\%$	$\sigma\%/\sqrt{2}$
Na ₂ O	1.539	1.542	1.622	1.553	1.618	1.629	1.584 \pm 0.043	2.73%	1.93%
Sc	40.248	40.585	40.749	40.864	41.031	40.888	40.728 \pm 0.278	0.68%	0.48%
Cr	120.128	121.187	118.603	119.054	120.348	120.057	119.896 \pm 0.931	0.78%	0.55%
Co	63.116	64.451	64.492	64.198	65.242	63.753	64.209 \pm 0.722	1.12%	0.80%
Cs	1.274	1.343	1.359	1.332	1.499	1.233	1.340 \pm 0.091	6.80%	4.81%
Ba	233.685	240.681	251.990	233.123	240.468	209.355	234.884 \pm 14.2	6.06%	4.29%
La	10.711	10.428	10.889	10.802	10.879	10.908	10.770 \pm 0.183	1.69%	1.20%
Ce	23.959	24.532	24.737	25.714	25.352	24.116	24.735 \pm 0.688	2.78%	1.97%
Nd	12.022	12.479	12.958	13.140	12.512	12.189	12.550 \pm 0.431	3.44%	2.43%
Sm	3.051	3.014	3.041	3.090	3.134	3.112	3.074 \pm 0.046	1.49%	1.06%
Eu	0.808	0.816	0.821	0.823	0.833	0.811	0.819 \pm 0.009	1.11%	0.78%
Gd	3.100	3.521	2.995	3.259	3.345	3.029	3.208 \pm 0.204	6.36%	4.50%
Tb	0.573	0.538	0.576	0.600	0.654	0.536	0.580 \pm 0.044	7.58%	5.36%
Ho	0.819	0.795	0.755	0.876	0.866	0.855	0.828 \pm 0.047	5.66%	4.00%
Yb	2.551	2.163	2.292	2.306	2.293	2.350	2.326 \pm 0.127	5.45%	3.86%
Lu	0.363	0.350	0.362	0.365	0.364	0.359	0.361 \pm 0.006	1.54%	1.09%
Hf	1.786	1.850	1.863	2.054	2.012	1.782	1.891 \pm 0.115	6.10%	4.31%
Ta	0.520	0.515	0.517	0.562	0.535	0.537	0.531 \pm 0.018	3.35%	2.37%
Th	3.882	3.685	3.728	3.369	3.327	3.412	3.567 \pm 0.228	6.39%	4.52%
U	1.166	1.059	1.170	1.161	1.294	1.181	1.172 \pm 0.075	6.37%	4.51%

Na₂O as per cent. and other elements as ppm.

σ and $\sigma\%$ are the standard and relative standard errors of individual measurements.

$\sigma\%/\sqrt{2}$ is relative standard error of two measurements, i.e. a duplicated analysis.

For Ba, Ce, Nd, Tb and Hf data would be improved by use of a larger detector for count 3 (see footnote to Table 2.1).

These analyses are all of the same sample and assume a fixed FeO value (8.77%) and hence errors in the XRF measurement of FeO are not a component in the above data; this would have a slight effect only for Sc, Cr and Eu so that these figures are also good estimates of precision when different samples are measured.

For twenty elements, this table shows the mean and the standard deviation of individual measurements and the coefficient of variation ($\sigma\%/\sqrt{2}$). The coefficient of variation divided by $\sqrt{2}$ is also shown since this is the estimate of the error in a duplicated analysis, as made on all

other chilled margin samples. The maximum value of $\sigma\%/\sqrt{2}$ is 5.36% (Tb), for nine elements it is less than two per cent. and for Sc, Cr Co and Eu it is below one per cent.

The errors shown in Table 3.1 are in general much better than those normally claimed for the INAA method. For example, in the study of precision of a sample of Whin Sill basalt, Potts *et al.* (1985) found that "For many elements coefficients of variation lie in the range 2-5% (1σ) which we consider to be excellent for INAA".

As mentioned above, all XRF and INAA analyses of unknowns were performed in duplicate and therefore the uncertainty is given by $\sigma/\sqrt{2}$. The calculated analytical precision (based on duplicate analyses) for the unknowns is listed in Table 3.2.

Table 3.2 Estimated precision (in percent) for the chilled margin analyses. Uncertainties in the major element XRF data have been calculated using multiple analyses of W-2. The precision of the elements obtained from INAA are from Table 2.1.

SiO ₂	0.3	Cs	4.8	Th	4.5
TiO ₂	0.8	La	1.2	U	4.5
Al ₂ O ₃	0.3	Ce	2.0	Hf	4.3
Fe ₂ O ₃	0.2	Nd	2.4	Sc	0.5
FeO	0.5	Sm	1.1	S	16.5
MnO	1.1	Eu	0.8		
MgO	0.6	Gd	4.5		
CaO	0.3	Tb	5.4		
Na ₂ O	1.3	Ho	4.0		
K ₂ O	0.9	Yb	3.9		
P ₂ O ₅	1.9	Lu	1.1		

3.1.2 Accuracy

The accuracy of the data produced by XRF and INAA in the Geology Department laboratory at ANU was expected to be high as a close check is maintained by frequent analysis of in-house standards (which have been carefully calibrated against well documented international standards such as BCR-1). Nevertheless, this has been quantified by multiple analysis of the international standard W-2.

The composition of W-2 may not be as well known as some other standards, and although it compares well with the original standard W-1 (Flanagan, 1984) many of the results for W-1 were produced prior to the improvements in instrumentation now available. Despite this, a recent compilation of data for W-2 by Flanagan (1984) includes data from up to 25 laboratories and is expected to provide a good estimate of W-2 with which to assess the accuracy of our results. The major advantage of using W-2 is that the chemical composition of this standard is close to that of the chilled margins analysed from Tasmania.

Eight separate preparations of W-2 were fused into glass discs for major element analysis (the oxidation state of Fe in W-2 was determined by 6 dissolutions of rock powder followed by titration, see Appendix 2). The trace elements Ba, Rb, Sr, Pb, Y, V, Cr, Mn, Ni, Cu, Zn and Ga have been obtained by triplicate analysis of pressed powder pellets.

The accuracy of the INAA data has been determined by including one W-2 sample in each of 6 different irradiation batches. Although these results could also have been used to determine

the analytical precision, the Mt. Wellington chilled margin sample 84-152 was included in each of six batches and has been used for this purpose since it is even closer to the average chilled margin in composition. The average composition for W-2 is presented in Table 3.3.

A number of important points are indicated from the results. The agreement between the compositions of W-2 obtained from this study, and the compilation of Flanagan (1984) is generally very close (Table 3.3); however, the few exceptions include Cs, Ho, Hf and V. Comparison with the original data used by Flanagan (1984) indicates that of the 9 laboratories reporting Cs data, 8 obtained values of $Cs \geq 0.9$ ppm. Similarly, the concentration of Hf in W-2 from each of the 10 sources listed (Flanagan, 1984) is greater than 2.35 ppm. It appears that our results for Hf, and possibly Cs, are genuinely in error (owing to unrecognised spectral interferences?).

Very little Ho data are available for W-2, and the three sources reported in Flanagan (1984) give mean values of 0.62, 0.71 and 1.26 ppm. The Ho value from this study is between the larger of these two values and is therefore considered to be in agreement.

The data compiled by Flanagan for V from different sources indicates some complication with the analysis of this element and a wide range in values has been obtained from different laboratories (128-440 ppm). The mean value estimated by Flanagan of 258.8 ppm is reasonable when the entire data set is considered in a broad sense. The 100 individual analyses listed have been divided into 10 ppm intervals and plotted against the frequency (Fig. 3.1). Clearly, the peak in the data at 250-260 ppm (indicated by the column labelled 255) is consistent with a mean value of ~ 259 ppm. However, if the data are divided into 5 ppm intervals, the bell-shaped distribution between ~ 255 and 285 ppm appears to split with the main peak at approximately 245-255 ppm and a second peak at ~ 260 -270 ppm (Fig. 3.1, inset). It may be fortuitous that few labs report data between these groups and that the true mean is close to ~ 259 ppm; however, the resolution of this problem requires further careful analysis.

The V content of ~ 240 ppm obtained in this study appears to be too low even if a bimodal distribution is assumed and the range ~ 245 -255 ppm preferred.

In addition, the difference in the Cr value is also surprisingly large (see later comparison between XRF and INAA results for this element).

Table 3.3 Average composition of W-2 obtained in this study compared with the compilation of Flanagan (1984). The $2\sigma_{\text{mean}}$ is listed for each elemental concentration. The number of individual analyses performed in this study (n) and the number of laboratory means used in the compilation of Flanagan (n) are also given. Major element oxides are in wt.% and trace elements are in ppm.

	W-2	$2\sigma_{\text{mean}}$	n	W-2	$2\sigma_{\text{mean}}$	n
	(this study)			(Flanagan, 1984)		
SiO ₂	53.21	0.166	8	52.68	0.58	18
TiO ₂	1.09	0.010	8	1.062	0.026	19
Al ₂ O ₃	15.48	0.054	8	15.45	0.32	17
Fe ₂ O ₃	1.57	0.022	8	1.53	0.174	6
FeO	8.32	0.036	8	8.34	0.186	6
MnO	0.16	0.002	8	0.167	0.008	20
MgO	6.53	0.043	8	6.37	0.116	13
CaO	11.11	0.037	8	10.86	0.156	11
Na ₂ O	2.17	0.035	8	2.20	0.074	17
K ₂ O	0.65	0.007	8	0.626	0.024	20
P ₂ O ₅	0.13	0.003	8	0.141	0.232	18
Cs	0.78	0.04	6	1.01	0.32	5
Ba	174.13		3	173.6	22.6	10
Rb	19.17		3	19.9	2.12	11
Sr	193.47		3	191.9	5.96	8
Pb	7.53		3	7.66		1
La	9.91	0.09	6	10.36	1.18	12
Ce	23.5	0.6	6	23.37	2.94	10
Nd	13.4	0.6	6	13.36	2.1	4
Sm	3.39	0.06	6	3.31	0.252	7
Eu	1.139	0.004	6	1.12	0.12	11
Gd	3.59	0.07	6	3.9		2
Tb	0.65	0.04	6	0.66	0.364	6
Ho	0.84	0.06	6	0.68	0.226	2
Yb	2.13	0.06	6	2.14	0.32	11
Lu	0.318	0.007	6	0.33	0.14	8
Y	19.75		3	23.4	3.26	7
Th	2.2	0.2	6	2.41	0.24	8
U	0.45	0.06	6	0.49	0.12	5
Zr				96.2	7.48	6
Hf	1.94	0.05	6	2.60	0.356	8
Nb				6.75	0.84	4
Sc	35.5	0.2	6	35.7	2.12	9
V	237.6		3	258.8	24.54	13
Cr	81.7		3	91.51	8.9	19
Mn	1296.3		3			
Ni	69.1		3	71.8	4.92	10
Cu	100.0		3	106.2	9.76	10
Zn	76.79		3	78.8	4.56	10
Ga	17.91		3	16.8	1.78	4

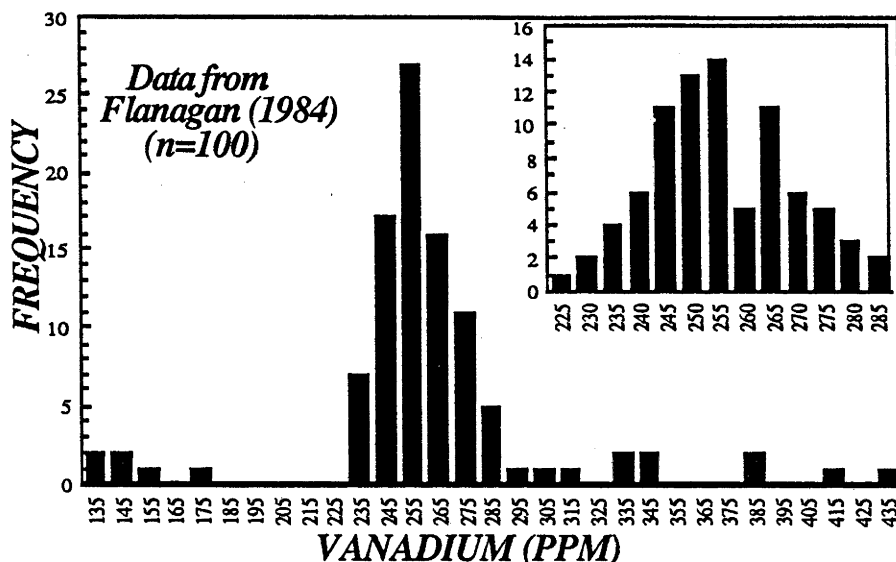


Fig. 3.1 Histogram illustrating the variation in V contents and their observed frequency for W-2 reported in Flanagan (1984). The data have been divided into 10 ppm intervals and indicate that at this scale the mean value lies in the interval 250-260 ppm. The inset (in which the concentrations are divided into 5 ppm intervals) illustrates that at a finer scale, the data may follow an almost bimodal distribution. In this case, the concentration of V is best represented by a mean either in the range 245-255 ppm, or 260-270 ppm.

3.1.3. Isotopic Data

An estimate of the precision and accuracy of the isotopic results has been achieved by several means. The errors associated with the Pb and O results are discussed in Appendix 2. This section deals with the Rb-Sr and Sm-Nd results which constitute the major proportion of the isotopic data. All but twelve of the sample dissolutions were performed on spiked samples and this section deals with the precision and accuracy of the concentration data as well as the isotopic ratios.

Analyses of various standard solutions during the period in which the data for the unknowns were obtained are listed in Table 3.4. These results were obtained jointly by all users of the MSZ mass spectrometer and do not represent my own work. The data are consistent with values reported in the literature. Analysis of these solutions is the predominant means by which a check of the accuracy of results is maintained. This is important if the running-conditions or hardware of the instrument are varied (e.g. McDonough et al., 1985).

Table 3.4 Compilation of analyses (of standard solutions) obtained during the collection of data on unknowns in this study. These results are a combination of analyses performed by the six operators of the MSZ in this period.

Standard	$^{87}\text{Sr}/^{86}\text{Sr}$	n	$^{143}\text{Nd}/^{144}\text{Nd}$	n
NBS 987	0.71028±3	18		
E&A	0.70801±1	11		
Ndα			0.511080±15	9
La Jolla			0.511025±07	4

In addition to the standard solutions, the international basalt standard BHVO-1 was dissolved and analysed in triplicate. Each dissolution was spiked to compare the reproducibility of the elemental concentrations as well as the isotopic compositions. The results are listed in Table 3.5 and are compared with values reported in the literature.

The isotopic compositions are identical to the results of White and Hofmann (1982) within the in-run analytical uncertainties. The concentration data (particularly the Sm and Nd results) are also close to reported values, although the Sr concentrations obtained in this study appear to be low. The range of values in the compilation of Abbey (1983) for Rb and Sr indicate that these lower values can be accommodated within the uncertainty. It will be shown in the next section that the concentration data obtained by isotope dilution is in excellent agreement with the results obtained using other techniques (shown previously to be accurate to within precision estimates for Rb, Sr, Sm and Nd by comparison with W-2).

Table 3.5 Compilation of 3 separate dissolutions of the basalt standard BHVO-1. Concentrations are in ppm, errors in individual analyses represent the 2σ in-run precision. The mean of the analyses is given together with the $2\sigma_{\text{mean}}$. Reported values for the elemental concentrations are from Abbey (1983) and the isotopic compositions are from White and Hofmann (1982). Note: The average Nd isotopic composition from this study has been recalculated normalised to $^{146}\text{Nd}/^{144}\text{Nd}=0.72190$ to facilitate the comparison.

Rb	Sr	$^{87}\text{Rb}/^{86}\text{Sr}$	$^{87}\text{Sr}/^{86}\text{Sr}$	Sm	Nd	$^{147}\text{Sm}/^{144}\text{Nd}$	$^{143}\text{Nd}/^{144}\text{Nd}$
8.8	380.1	0.067	0.70352 ± 2	6.14	24.75	0.1500	0.512155 ± 24
8.9	383.7	0.067	0.70347 ± 4	6.14	24.75	0.1499	0.512165 ± 42
9.4	393.7	0.069	0.70347 ± 4	6.24	25.28	0.1493	0.512128 ± 26
Average of three analyses ($\pm 2\sigma_{\text{mean}}$):							
9.0	385.8	0.068	0.70349	6.17	24.93	0.1497	0.512149
± 0.3	± 6.6	± 0.001	± 0.00003	± 0.05	± 0.29	± 0.0004	± 0.000018
(recalculated value = 0.512974)							
Reported values ($\pm 2\sigma$ in-run precision):							
10	420		0.70346 ± 2	6.1	24		0.512948 ± 13
			0.70345 ± 5				0.512990 ± 24

It is apparent that the reproducibility and accuracy of the isotopic compositions measured on the MSZ are within the 2σ in-run precision of individual analyses.

The Nd compositions for chilled margin dolerites analysed on the MSZ are considered to be the same within error, although it was thought that some analyses appear to be close to the limit of this uncertainty and may show subtle variation if the uncertainties were reduced. To examine this possibility, 12 unspiked Nd samples were run on the multiple-collector mass spectrometer to improve the precision. Each sample was analysed using the automatic data acquisition program, and the samples which were not exhausted were re-run.

The results are given in Table 3.6 and indicate that the 2σ in-run precision obtained for individual analyses is close to the reproducibility. The accuracy of the results was examined by

comparison of various standards (run by a number analysts) with the results obtained for the MSZ spectrometer. To make the comparison it was necessary to re-normalise the isotopic ratios because the correction for mass-fractionation used on the MSZ is $^{146}\text{Nd}/^{142}\text{Nd}=0.636151$, compared with $^{146}\text{Nd}/^{144}\text{Nd}=0.72190$ on the MAT. This was achieved by multiplying the MSZ data by a factor of 1.001596 (derived by dividing the average BCR-1 $^{143}\text{Nd}/^{144}\text{Nd}$ ratio obtained from the MAT by that from the MSZ). The results are compared in Table 3.7 and indicate that there is a systematic offset in the ratios obtained from each spectrometer, with the MAT giving values consistently higher than the MSZ. This off-set is illustrated by the ϵ_{Nd} values calculated for the standards BHVO-1 and La Jolla relative to the bulk-earth Nd composition (given by the mean BCR-1 value for each instrument). If there were random differences between the two mass spectrometers, the calculated ϵ_{Nd} values for BHVO-1 and La Jolla should not coincide; reference to Table 3.7 indicates that this is not the case, and the values for these standards are identical within the estimated error.

Table 3.6 Comparison between the reproducibility and 2σ in-run precision of Nd analyses obtained from the MAT 261 multiple collector mass spectrometer.

Sample Number ($\pm 2\sigma_{\text{mean}}$)	$^{143}\text{Nd}/^{144}\text{Nd}$ ($\pm 2\sigma$)	$^{143}\text{Nd}/^{144}\text{Nd}$ ($\pm 2\sigma$)	$^{143}\text{Nd}/^{144}\text{Nd}$
	RUN 1	RUN 2	MEAN
84-108	0.512325 ± 5	0.512313 ± 6	0.512319 ± 9
84-109	0.512324 ± 5	0.512310 ± 9	0.512317 ± 10
84-129	0.512322 ± 5	0.512335 ± 7	0.512329 ± 9
84-141	0.512320 ± 5	0.512312 ± 11	0.512316 ± 6
84-144	0.512330 ± 5	0.512324 ± 8	0.512327 ± 5
84-149	0.512331 ± 6		
84-169	0.512344 ± 7	0.512332 ± 6	0.512338 ± 9
84-170	0.512325 ± 8	0.512330 ± 6	0.512328 ± 4
84-203	0.512334 ± 8	0.512324 ± 5	0.512329 ± 7
84-205	0.512335 ± 8	0.512318 ± 5	0.512327 ± 12
84-221	0.512333 ± 5		
84-222	0.512328 ± 5		

Table 3.7 Comparison between the $^{143}\text{Nd}/^{144}\text{Nd}$ results for standards using the MAT and MSZ spectrometers. The number of analyses used for each mean is given in parenthesis. Ratios obtained using the MAT are systematically higher by 0.062-0.088 ‰. This deviation is removed by calculating the ϵ_{Nd} values using the mean BCR-1 value from each instrument as a measure of the present-day bulk earth Nd composition.

Standard	$^{143}\text{Nd}/^{144}\text{Nd} \pm 2\sigma_{\text{mean}}$ MAT	$^{143}\text{Nd}/^{144}\text{Nd} \pm 2\sigma_{\text{mean}}$ MSZ	Difference in ‰ MAT - MSZ
BCR-1	0.512653 \pm 5 (7)	0.512608 \pm 10 (7)	0.088
BHVO-1	0.512999 \pm 5 (1)	0.512967 \pm 17 (3)	0.062
La Jolla	0.511873 \pm 5 (17)	0.511841 \pm 7 (4)	0.063
Using the BCR-1 mean obtained from each instrument as a measure of present-day CHUR:			
	ϵ_{Nd}	ϵ_{Nd}	
BHVO-1	6.75 \pm 0.14	7.00 \pm 0.39	
La Jolla	-15.22 \pm 0.14	-14.96 \pm 0.24	

3.2 COMPARISON BETWEEN CONCENTRATION DATA OBTAINED BY DIFFERENT INSTRUMENTAL TECHNIQUES (XRF, INAA AND ID)

Various techniques have been employed to obtain a wide range of elemental and isotopic data and include XRF spectrometry, INAA, and isotope dilution mass spectrometry (ID). The concentrations of a few elements have been determined independently by more than one technique, providing a useful means for comparing the accuracy of different instrumental methods.

The following comparison is based on chilled margin dolerites and selected samples from differentiated intrusions, as these were analysed in the most detail. Most of the diagrams used in the discussion contain 67 data points; however, the plots using data obtained by ID contain less points owing to the smaller number of samples analysed by this technique. All figures include a 1:1 ratio reference line to illustrate the correlation expected in the ideal case where the values obtained by different methods are identical.

Figure 3.2 compares some of the major element data obtained by XRF and INAA. The Na_2O results from the two techniques show a linear relationship which is offset from the reference line. The values obtained by XRF analysis are consistently high relative to the INAA data. This is likely to be the result of a small difference in the calibration of the instruments as the offset is systematic.

In contrast, the CaO data from XRF and INAA agree closely and indicate that the results are likely to be accurate from both techniques. The third diagram in Figure 3.2 illustrates the correlation between the MnO contents of 67 samples measured by different XRF techniques (note that most points overlap each other giving the false impression that only 9 points are plotted). As well as measuring MnO as a major element, the Mn content is determined as a trace element by analysis of pressed powder pellets. The trace elemental abundances (in ppm) have been recalculated as oxides and the comparison illustrated indicates that excellent agreement is

obtained.

Other trace elements analysed by both XRF and INAA include Ce, La, Cr and Ba. The four diagrams of Figure 3.3 indicate that agreement is generally good, particularly in the case of La and Ce. Cr by INAA shows a small but consistent offset to higher values, whereas the Ba data from INAA are systematically low compared with the XRF results. This does not seriously effect the results below approximately 300 ppm, but does become an increasing problem at higher Ba contents.

Figure 3.4 compares the Sr and Rb data obtained by XRF and ID. The results from XRF and ID compare very well and plot close to the reference lines. It was shown previously that the Rb and Sr data from XRF are accurate (using W-2), and that the ID data from BHVO-1 are also close to reported values. The excellent agreement between the results shown in Figure 3.4 supports this conclusion. These results indicate that despite the increased precision obtained using ID, the accuracy is probably similar for both XRF and ID.

Finally, the Sm and Nd contents determined by INAA and ID are compared in Figure 3.5. Most of the data plot on or very close to the 1:1 reference lines with 3 points plotting one or two ppm away.

Comparisons such as those made in Figures 3.2 to 3.5 increase our confidence in the accuracy of the data obtained by independent analytical techniques, particularly where the instruments have been calibrated using different methods (e.g. the original calibration for Cr was performed using spec-pure chemicals for the XRF spectrometer, compared with W-1 for INAA). Concentration data obtained by ID has confirmed the accuracy of the XRF and INAA results. Conversely, the XRF and INAA results have enabled the constant monitoring of the spike solutions used in the isotopic measurements.

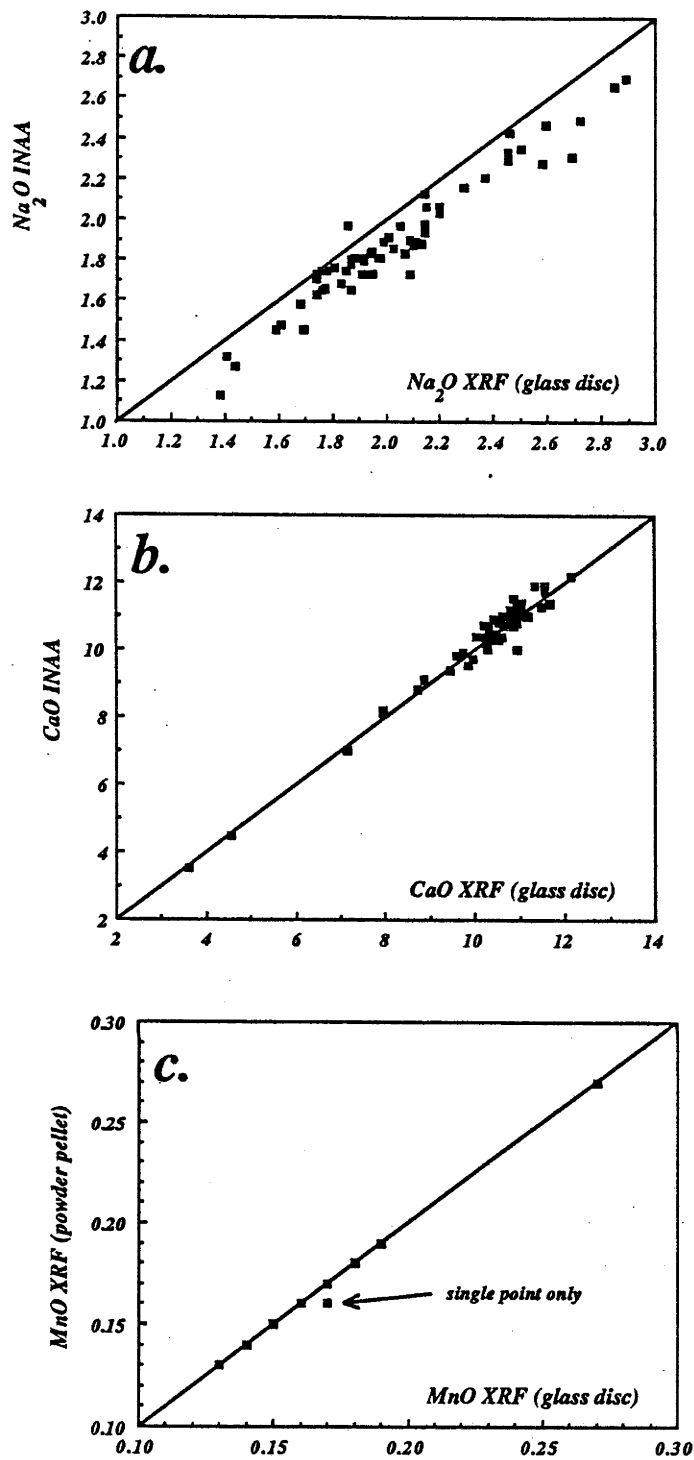


Fig. 3.2 Comparison between major element compositions obtained by different analytical techniques. a. Na_2O from INAA and XRF. b. CaO from INAA and XRF. c. MnO measured by XRF both as a major element oxide (glass disc) and trace element (as ppm Mn, recalculated for this comparison).

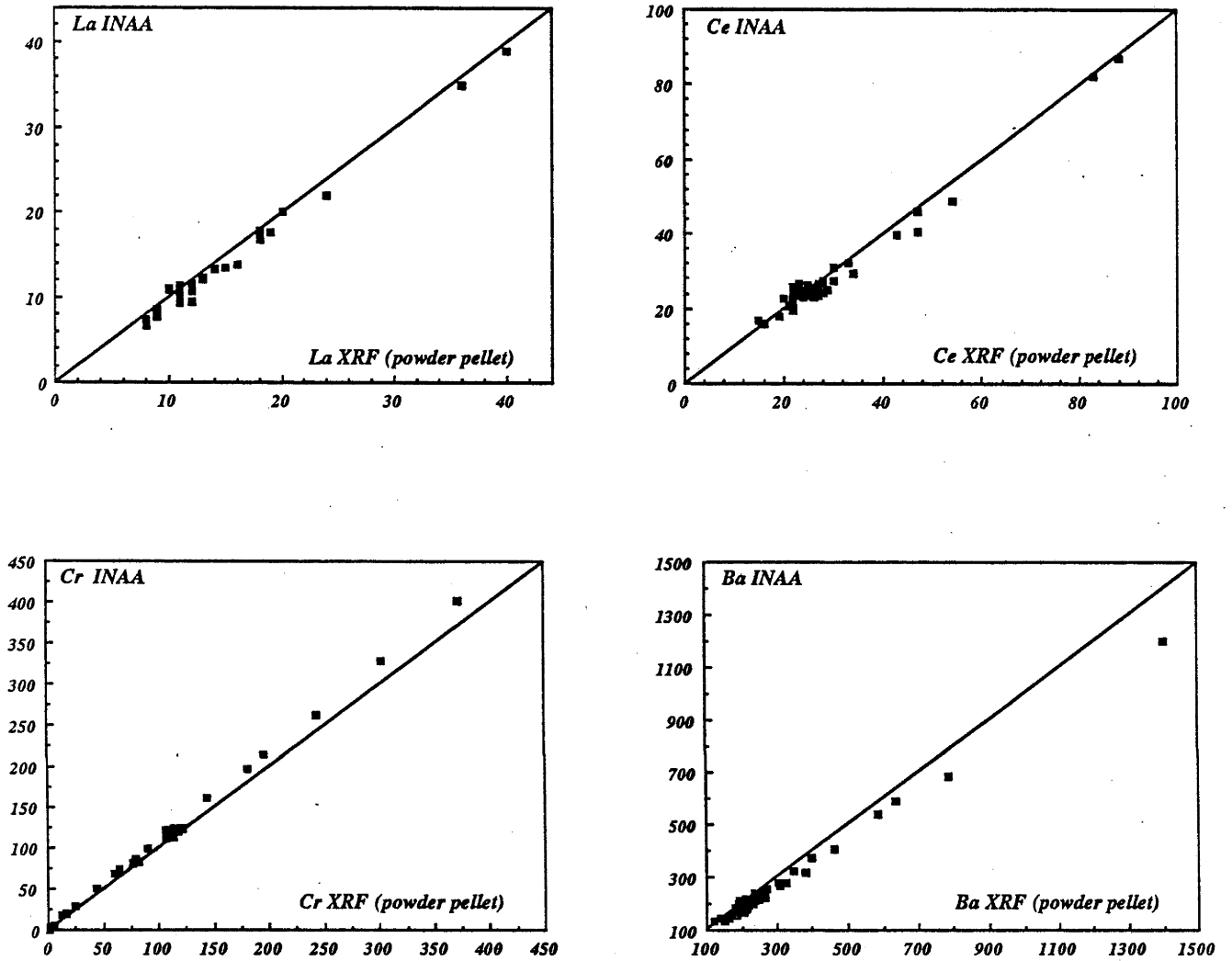


Fig. 3.3. Comparison between Ce, La, Cr and Ba data obtained by XRF and INAA.

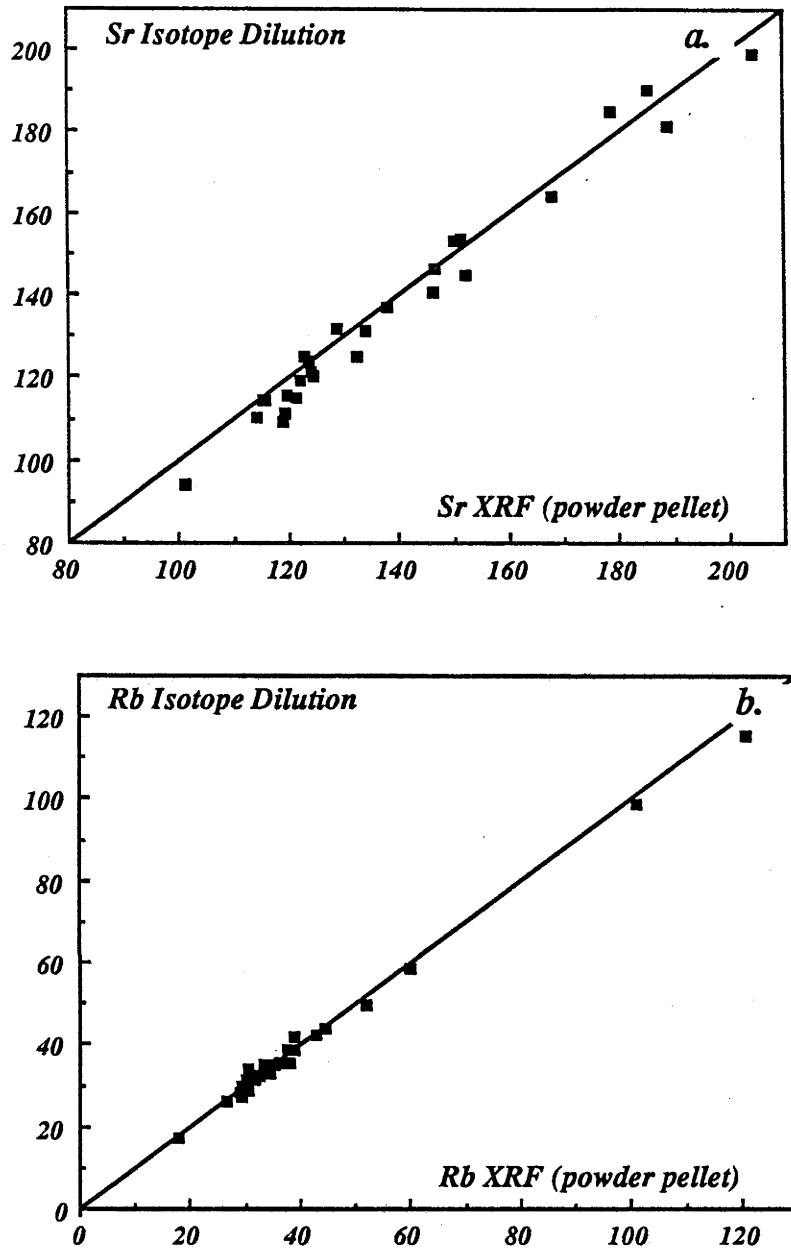


Fig. 3.4 Comparison between Sr (a.) and Rb (b.) data obtained using XRF and ID.

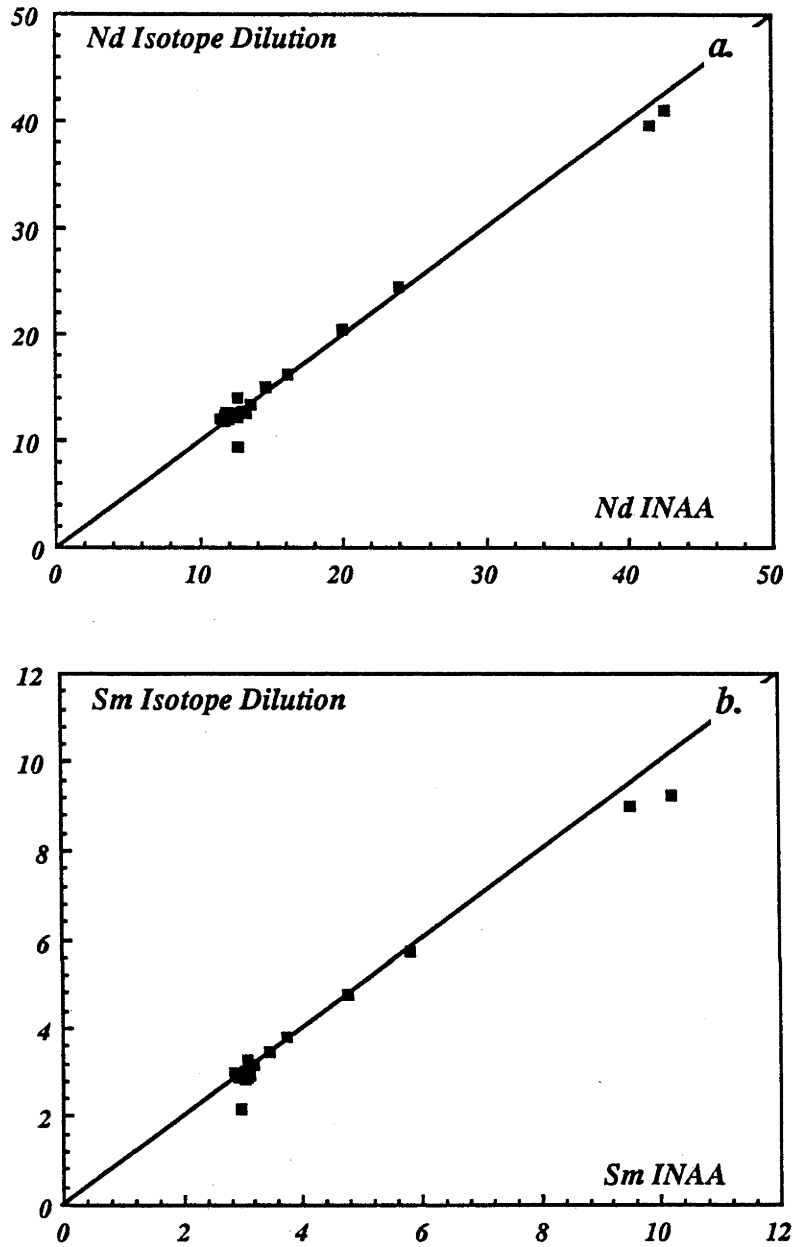


Fig. 3.5 Comparison between REE data obtained by ID and INAA. a. Nd from ID versus INAA. b. Sm from ID versus INAA.

APPENDIX 4.

GEOCHEMICAL DATABASE

The geochemical data compiled in this appendix are divided into three sections. The first section includes all of the geochemical data obtained on samples collected from Tasmania and constitutes the major proportion of the results. The second section of the appendix contains the analyses of samples from Portal Peak (Antarctica). Finally, the third part of the compilation lists the data obtained on samples related to the Tasmanian Dolerites occurring in western Victoria and Kangaroo Island. The powders remaining from the analytical work, as well as the thin-sections and rock specimens are retained in the A.N.U Research School of Earth Sciences collection.

4.1 TASMANIAN SAMPLES

The data from the Tasmanian specimens are subdivided into three sections. The major and trace element X-ray fluorescence spectrometric (XRF) data and wet-chemical results from Tasmanian samples are presented in order of increasing catalogue number in the first section of this part of the appendix. The results include analyses of sediments in direct contact with the chilled margin dolerites in a few localities. These samples were chosen as representative of the dominant country-rock material in contact with the dolerite and include arkosic sandstones and fossiliferous marls of the Parmeener Supergroup (Permo-Triassic). The dolerites include the 37 chilled margin samples and the many dolerites sampled from measured sections through intrusions.

Over 60 of the samples included in the first compilation were selected for more detailed analysis using instrumental neutron activation analysis (INAA). These additional results (combined with the data listed in the first compilation) are located at the end of this section. These more detailed data sets are divided into two parts; the first part contains the data for the chilled margin samples and provides the basis for the discussion in Chapter 4, the second part contains the data for measured sections through intrusive bodies used to investigate the differentiation history of the Tasmanian Dolerites (Chapter 3). Although the organisation of the data in this manner necessarily duplicates some of the results, this seems justified by the benefit of being able to refer to the particular group of results discussed in the text .

4.2 PORTAL PEAK SAMPLES

The XRF and wet-chemical data obtained on the suite of samples from Portal Peak are listed in order of increasing catalogue number (the original field numbers assigned by Prof. Gunter Faure and Teresa Mensing have also been included). Six of the samples were chosen for analysis by INAA and the results are located at the end of this section of the appendix.

4.3 VICTORIAN AND KANGAROO ISLAND SAMPLES

These 7 samples have been analysed for major elements by XRF and wet-chemical techniques, and trace elements by Spark Source Mass Spectrometry. The combined results are located in this section and are listed in order of increasing catalogue number.

4.1 DATABASE OF TASMANIAN SAMPLES

XRF AND WET-CHEMICAL ANALYSES OF CHILLED AND
DIFFERENTIATED DOLERITES AS WELL AS COUNTRY-ROCKS
SAMPLED FROM CONTACTS

	84 101	84 102	84 103	84 104	84 105	84 106	84 107
SiO ₂	54.42	53.66	62.91	67.29	72.15	54.00	53.73
TiO ₂	0.63	0.56	1.26	0.83	0.30	0.63	0.63
Al ₂ O ₃	14.54	16.76	10.86	11.09	13.80	14.69	14.60
Fe ₂ O ₃	0.86	1.88	5.90	3.51	1.47	1.03	1.47
FeO	7.99	7.45	6.22	5.32	0.45	7.83	7.53
MnO	0.17	0.17	0.18	0.14	0.02	0.19	0.19
MgO	6.68	4.41	0.53	0.20	0.53	6.71	6.78
CaO	11.05	10.70	4.54	3.60	0.99	10.69	10.59
Na ₂ O	1.99	2.13	2.59	2.89	7.88	2.15	2.09
K ₂ O	0.60	0.80	2.67	3.08	1.79	0.85	0.83
P ₂ O ₅	0.09	0.09	0.38	0.19	0.11	0.09	0.09
S	0.04	0.03	0.06	0.05	0.02	0.03	0.03
H ₂ O+	0.93	1.08	1.18	0.97	0.52	0.62	0.80
H ₂ O-	0.26	0.26	0.47	0.70	0.29	0.53	0.68
CO ₂	0.20	0.08	0.46	0.18	0.11	0.16	0.15
rest	0.15	0.13	0.18	0.19	0.09	0.15	0.15
	100.60	100.19	100.39	100.23	100.52	100.35	100.34
O=S	0.02	0.02	0.03	0.02		0.01	0.01
Total	100.58	100.17	100.36	100.21	100.52	100.34	100.33
Ba	190	195	585	635	165	205	200
Rb	17.8	29.5	101.0	121.0	33.5	30.6	29.3
Sr	132.2	154	114	101	102	121.4	119.0
Pb	7	5	16	17	14	5	6
La	11	10	36	40	18	11	10
Ce	24	23	83	88	39	23	24
Y	21.1	19	59	59	10	20.3	20.7
Th	4.0	2.0	13.0	14.0	9.0	4.0	3.0
U	3.0	1.0	6.0	2.0	1.0	2.0	1.0
Zr	94	84	273	334	179	94	96
Nb	4.0	3.0	14.5	16.0	4.5	5.0	4.5
V	223	215	1	1	41	231	221
Cr	113	3	1	1	34	121	112
Mn	1340	1300	1370	1060	155	1450	1470
Ni	79	40	3	1	9	83	80
Cu	78	153	37	19	3	79	75
Zn	82	78	128	111	27	80	78
Ga	16.0	18.0	19.5	18.0	14.5	15.5	16.0

	84 108	84 109	84 110	84 111	84 112	84 113	84 114
SiO ₂	54.02	54.43	54.40	53.99	55.24	54.51	54.17
TiO ₂	0.63	0.66	0.66	0.64	0.83	0.63	0.61
Al ₂ O ₃	14.66	14.42	14.41	14.66	13.40	14.84	14.67
Fe ₂ O ₃	1.19	1.59	1.57	1.22	1.96	1.61	1.25
FeO	7.70	7.60	7.51	7.59	8.83	7.31	7.47
MnO	0.18	0.18	0.17	0.17	0.19	0.17	0.17
MgO	6.78	6.46	6.35	6.83	5.24	6.66	6.83
CaO	10.53	10.04	10.23	10.32	9.63	10.66	10.71
Na ₂ O	2.14	2.14	1.87	1.89	2.08	1.89	2.14
K ₂ O	0.84	0.98	1.07	1.02	1.06	0.95	0.81
P ₂ O ₅	0.09	0.10	0.10	0.09	0.12	0.09	0.09
S	0.07	0.05	0.05	0.05	0.02	0.05	0.06
H ₂ O+	0.93	1.35	1.29	1.10	1.41	0.89	0.85
H ₂ O-	0.58	0.49	0.38	0.44	0.41	0.49	0.28
CO ₂	0.22	0.14	0.17	0.14	0.09	0.16	0.17
rest	0.15	0.15	0.15	0.16	0.15	0.15	0.14
	100.71	100.78	100.38	100.31	100.66	101.06	100.42
O=S	0.03	0.02	0.02	0.02		0.02	0.03
Total	100.68	100.76	100.36	100.29	100.66	101.04	100.39
Ba	225	195	200	305	235	215	185
Rb	29.1	33.2	44.0	33.7	38.5	36.0	27.5
Sr	124.4	178.6	166	155.5	132	128	118
Pb	5	5	6	6	7	6	5
La	11	11	12	11	15	11	12
Ce	24	25	27	22	31	25	24
Y	20.4	20.8	21	20.5	26	20	19
Th	4.0	3.0	4.0	4.0	5.0	5.0	3.0
U	1.0	2.0	1.0	1.0	4.0	1.0	1.0
Zr	95	100	101	95	126	95	91
Nb	5.0	4.5	4.5	5.0	6.0	4.5	5.0
V	222	238	229	225	239	215	221
Cr	110	82	78	114	15	106	124
Mn	1410	1390	1350	1320	1460	1300	1300
Ni	78	71	69	80	50	79	79
Cu	74	73	74	77	110	81	73
Zn	79	81	82	79	90	74	74
Ga	15.5	17.0	16.0	17.0	18.5	16.5	15.0

	84 115	84 116	84 117	84 118	84 119	84 120	84 121
SiO ₂	53.94	53.49	53.12	53.70	54.50	52.92	52.33
TiO ₂	0.62	0.49	0.49	0.62	0.64	0.47	0.41
Al ₂ O ₃	14.45	12.65	15.34	14.60	14.85	13.59	14.03
Fe ₂ O ₃	1.36	1.53	1.81	1.84	1.44	1.98	1.98
FeO	7.49	7.43	5.68	7.01	7.46	6.57	5.90
MnO	0.17	0.18	0.15	0.17	0.16	0.18	0.17
MgO	6.76	9.03	7.96	6.71	6.23	8.85	9.20
CaO	10.30	11.88	11.95	10.20	10.73	12.14	12.49
Na ₂ O	2.14	1.38	1.41	2.19	2.01	1.36	1.33
K ₂ O	0.93	0.62	0.66	0.98	0.91	0.57	0.49
P ₂ O ₅	0.09	0.07	0.07	0.09	0.08	0.07	0.06
S	0.09	0.02	0.02	0.03	0.02	0.04	0.02
H ₂ O+	0.86	0.84	0.45	0.78	0.86	1.01	0.99
H ₂ O-	0.36	0.22	0.24	0.63	0.51	0.25	0.21
CO ₂	0.20	0.07		0.11	0.11	0.10	0.12
rest	0.15	0.13	0.14	0.15	0.13	0.13	0.13
	99.91	100.03	99.49	99.81	100.64	100.23	99.86
O=S	0.04			0.01	0.01	0.02	
Total	99.87	100.03	99.49	99.80	100.63	100.21	99.86
Ba	200	155	150	215	205	140	125
Rb	35.2	23.0	24.0	37.0	33.0	20.0	18.0
Sr	142.0	107	118	155	118	98	105
Pb	6	5	4	6	5	6	3
La	11	8	8	11	11	9	6
Ce	24	19	20	24	25	17	13
Y	20.2	16	15	20	20	15	13
Th	3.0	1.0	2.0	4.0	3.0	2.0	3.0
U	2.0	1.0	1.0	1.0	1.0	1.0	1.0
Zr	94	69	71	93	96	64	55
Nb	4.5	3.5	3.5	4.5	4.5	2.5	2.0
V	222	242	205	214	222	237	226
Cr	114	78	199	115	27	99	124
Mn	1300	1390	1200	1300	1240	1390	1310
Ni	79	99	100	77	69	102	110
Cu	74	51	48	73	79	50	45
Zn	77	66	60	76	70	66	60
Ga	16.5	14.5	14.5	16.0	16.5	14.0	13.5

	84 122	84 123	84 124	84 125	84 126	84 127	84 128
SiO ₂	53.80	52.55	55.92	54.62	55.29	54.19	54.21
TiO ₂	0.56	0.36	1.04	1.50	0.82	0.75	0.64
Al ₂ O ₃	14.72	18.16	15.61	13.96	17.15	17.64	14.66
Fe ₂ O ₃	0.97	1.49	2.72	3.97	1.68	1.52	0.76
FeO	7.26	5.04	7.96	9.02	7.48	7.24	8.02
MnO	0.16	0.14	0.15	0.18	0.13	0.14	0.17
MgO	7.81	7.24	2.38	2.89	2.52	3.10	6.75
CaO	10.81	13.16	8.56	8.31	9.42	10.25	10.99
Na ₂ O	2.35	1.55	2.68	2.45	2.35	2.29	1.74
K ₂ O	0.68	0.51	1.33	1.30	1.21	1.05	0.68
P ₂ O ₅	0.08	0.05	0.14	0.15	0.13	0.11	0.09
S	0.02	0.03	0.03	0.03	0.06	0.03	0.04
H ₂ O+	0.82	0.64	1.23	1.17	1.29	1.62	1.13
H ₂ O-	0.35	0.18	0.33	0.14	0.26	0.26	0.22
CO ₂	0.15	0.07	0.15	0.10	0.19	0.12	0.16
rest	0.15	0.11	0.16	0.18	0.14	0.14	0.15
	100.69	101.28	100.39	99.97	100.12	100.45	100.41
O=S	0.01	0.01	0.01	0.01	0.03	0.01	0.02
Total	100.68	101.27	100.38	99.96	100.09	100.44	100.39
Ba	175	115	315	305	285	260	190
Rb	22.0	18.0	49.5	49.0	43.5	38.0	25.4
Sr	130	129	145	145	179	189	121.2
Pb	5	4	9	8	7	7	6
La	10	6	17	18	17	14	12
Ce	23	14	40	41	37	34	25
Y	17	12	28	29	25	22	20.1
Th	4.0	1.0	7.0	6.0	6.0	3.0	3.0
U	1.0	1.0	1.0	1.0	1.0	3.0	3.0
Zr	82	48	144	140	124	108	94
Nb	4.0	2.0	7.0	8.0	6.0	4.5	4.5
V	210	175	214	324	161	176	219
Cr	181	104	1	1	1	1	116
Mn	1250	1090	1190	1380	1040	1050	1340
Ni	94	87	14	18	19	27	79
Cu	63	41	126	147	119	109	80
Zn	68	51	92	106	78	75	79
Ga	15.0	15.5	20.5	19.5	20.5	19.5	16.0

	84 129	84 130	84 131	84 132	84 133	84 134	84 135
SiO ₂	54.15	53.70	55.59	54.63	55.10	56.78	63.96
TiO ₂	0.61	0.54	0.90	0.94	1.57	1.59	1.22
Al ₂ O ₃	14.66	15.08	16.73	16.99	14.83	14.39	11.16
Fe ₂ O ₃	0.57	1.32	1.84	1.99	3.52	4.17	6.45
FeO	8.16	7.25	7.83	7.93	8.77	7.91	4.82
MnO	0.17	0.16	0.15	0.14	0.16	0.15	0.16
MgO	6.61	6.97	2.37	2.52	2.09	1.43	0.52
CaO	10.97	11.47	9.22	9.45	7.95	7.14	3.99
Na ₂ O	1.88	1.67	2.69	2.45	2.69	2.72	2.76
K ₂ O	0.75	0.69	1.21	1.13	1.40	1.65	2.88
P ₂ O ₅	0.09	0.08	0.13	0.12	0.15	0.18	0.32
S	0.04	0.05	0.07	0.05	0.07	0.05	0.03
H ₂ O+	0.74	0.64	1.17	1.18	1.20	1.43	1.18
H ₂ O-	0.25	0.31	0.28	0.22	0.34	0.38	0.76
CO ₂	0.13	0.17	0.18	0.17	0.16	0.16	0.16
rest	0.14	0.13	0.15	0.15	0.19	0.16	0.18
	99.92	100.23	100.51	100.06	100.19	100.29	100.55
O=S	0.02	0.02	0.03	0.02	0.03	0.02	0.01
Total	99.90	100.21	100.48	100.04	100.16	100.27	100.54
Ba	180	175	285	265	325	380	590
Rb	26.4	24.5	43.0	39.0	52.0	60.0	108.0
Sr	122.1	126	149	152	146	134	112
Pb	5	6	7	7	9	11	16
La	11	11	15	16	19	24	36
Ce	24	21	36	33	47	54	85
Y	19.8	17	25	23	31	37	54
Th	4.0	4.0	6.0	6.0	9.0	6.0	12.0
U	1.0	1.0	3.0	2.0	2.0	2.0	6.0
Zr	94	79	125	115	152	176	297
Nb	4.5	4.0	6.5	6.0	7.0	8.0	14.0
V	212	211	193	220	336	149	1
Cr	113	96	1	1	1	1	1
Mn	1310	1270	1130	1100	1240	1180	1210
Ni	78	80	17	16	11	3	2
Cu	64	67	124	135	157	103	26
Zn	78	71	86	81	97	102	118
Ga	16.5	16.0	20.0	21.0	21.0	21.5	19.0

	84 136	84 137	84 138	84 139	84 140	84 141	84 142
SiO ₂	62.21	55.17	54.61	64.90	53.83	54.41	69.55
TiO ₂	0.98	0.85	0.64	0.54	0.63	0.63	0.73
Al ₂ O ₃	13.24	16.01	14.86	10.52	14.57	14.70	19.05
Fe ₂ O ₃	2.98	2.04	0.42	0.12	1.12	1.20	0.77
FeO	6.66	7.64	8.42	2.86	7.71	7.71	3.94
MnO	0.14	0.15	0.17	0.02	0.17	0.17	0.04
MgO	0.58	3.59	6.66	1.12	6.65	6.84	0.71
CaO	5.41	9.66	10.91	13.12	10.74	10.79	0.92
Na ₂ O	3.05	2.31	1.87	2.37	1.78	1.95	1.14
K ₂ O	2.37	1.16	0.86	2.10	0.82	0.83	1.14
P ₂ O ₅	0.29	0.12	0.09	0.05	0.09	0.09	0.03
S	0.06	0.06	0.05	0.30	0.05	0.04	0.02
H ₂ O+	1.61	1.08	0.43	0.95	0.71	0.76	1.17
H ₂ O-	0.43	0.07	0.23	0.17	0.52	0.29	0.25
CO ₂	0.16	0.18	0.15	1.81	0.15	0.21	0.02
rest	0.16	0.14	0.15	0.14	0.14	0.15	0.17
	100.33	100.23	100.52	101.09	99.68	100.77	99.65
O=S	0.03	0.03	0.02		0.02	0.02	
Total	100.30	100.20	100.50	101.09	99.66	100.75	99.65
Ba	510	270	190	210	195	310	545
Rb	90.0	42.5	31.7	74.0	29.8	30.0	65.0
Sr	116	137	123.9	366	125.8	115.8	132
Pb	14	8	6	5	5	6	32
La	31	16	10	19	11	12	18
Ce	70	38	25	43	24	26	44
Y	45	24	20	23	19	19	25
Th	12.0	6.0	5.0	9.0	5.0	3.0	15.0
U	4.0	1.0	1.0	1.0	2.0	1.0	4.0
Zr	245	125	95	158	94	93	284
Nb	12.5	6.0	4.0	5.5	4.5	4.0	10.5
V	12	187	223	69	214	214	51
Cr	1	3	115	27	107	107	20
Mn	1050	1160	1330	150	1290	1290	280
Ni	2	31	80	24	78	75	11
Cu	43	106	76	8	76	72	7
Zn	103	80	78	97	80	75	110
Ga	20.0	19.5	16.5	13.5	16.5	16.5	23.5

	84 143	84 144	84 145	84 146	84 147	84 148	84 149
SiO ₂	54.81	54.08	54.29	58.06	54.22	54.65	53.79
TiO ₂	0.66	0.63	0.64	0.40	0.63	0.64	0.63
Al ₂ O ₃	14.45	14.66	14.74	8.91	14.65	14.86	14.61
Fe ₂ O ₃	1.53	0.97	0.96	0.09	0.71	0.32	0.65
FeO	7.70	7.92	8.02	2.30	8.13	8.52	8.15
MnO	0.18	0.17	0.17	0.15	0.17	0.17	0.18
MgO	6.34	6.78	6.58	1.11	6.64	6.77	6.64
CaO	9.73	10.84	10.76	23.87	10.87	10.95	10.93
Na ₂ O	2.05	2.11	2.02	2.10	1.86	1.81	1.76
K ₂ O	0.98	0.84	0.96	1.50	0.81	0.85	0.83
P ₂ O ₅	0.10	0.09	0.09	0.09	0.09	0.10	0.09
S	0.03	0.04	0.05	0.28	0.05	0.11	0.05
H ₂ O+	0.92	0.54	0.67	0.97	0.44	0.39	0.61
H ₂ O-	0.43	0.35	0.25	0.17	0.29	0.23	0.35
CO ₂	0.14	0.18	0.19	1.10	0.16	0.29	0.19
rest	0.15	0.14	0.15	0.12	0.14	0.15	0.15
	100.20	100.34	100.54	101.22	99.86	100.81	99.61
O=S	0.01	0.02	0.02	0.14	0.02	0.05	0.02
Total	100.19	100.32	100.52	101.08	99.84	100.76	99.59
Ba	235	200	210	215	190	200	190
Rb	35.6	34.6	39.0	45.0	30.5	30.2	30.4
Sr	123.8	119.3	128	373	120.3	123.7	128.6
Pb	6	6	6	8	5	6	5
La	11	12	11	15	11	12	11
Ce	27	23	23	36	25	24	26
Y	21.6	20.1	20	27	19.9	20.2	20.9
Th	4.0	5.0	4.0	7.0	3.0	4.0	4.0
U	2.0	1.0	1.0	1.0	1.0	1.0	2.0
Zr	102	91	93	153	93	95	94
Nb	5.5	4.5	3.5	3.5	4.5	4.5	4.0
V	232	220	222	44	220	224	223
Cr	78	115	100	11	113	117	117
Mn	1390	1320	1310	1150	1330	1340	1370
Ni	71	79	75	14	79	80	79
Cu	80	62	78	4	65	77	65
Zn	85	76	77	63	77	77	79
Ga	17.5	15.0	16.0	11.5	16.0	17.0	16.5

	84 150	84 151	84 152	84 153	84 154	84 155	84 156
SiO ₂	54.17	68.10	54.30	54.06	54.11	53.70	53.75
TiO ₂	0.63	0.46	0.64	0.62	0.62	0.58	0.51
Al ₂ O ₃	14.63	10.55	14.62	14.52	14.56	14.56	14.78
Fe ₂ O ₃	0.95	0.34	1.65	1.25	1.39	1.28	0.98
FeO	7.90	2.95	7.29	7.62	7.44	7.26	7.17
MnO	0.17	0.02	0.17	0.17	0.17	0.16	0.16
MgO	6.75	1.39	6.60	6.68	6.82	7.25	8.08
CaO	10.92	9.26	10.92	10.64	10.94	11.15	11.65
Na ₂ O	1.85	2.04	1.68	1.77	1.74	1.59	1.69
K ₂ O	0.84	2.42	0.85	0.92	0.83	0.76	0.67
P ₂ O ₅	0.09	0.10	0.09	0.09	0.09	0.08	0.07
S	0.05	0.72	0.05	0.05	0.04	0.04	0.04
H ₂ O+	0.67	1.51	0.78	0.75	0.66	0.58	0.57
H ₂ O-	0.38	0.31	0.32	0.31	0.27	0.25	0.33
CO ₂	0.21	1.67	0.12	0.13	0.11	0.14	0.09
rest	0.15	0.17	0.15	0.15	0.14	0.14	0.14
	100.36	102.01	100.23	99.73	99.93	99.52	100.68
O=S	0.02	0.36	0.02	0.02	0.02	0.02	0.02
Total	100.34	101.65	100.21	99.71	99.91	99.50	100.66
Ba	190	325	250	265	195	185	165
Rb	29.7	76.0	29.9	33.5	29.5	27.5	24.0
Sr	128.5	500	123.6	137	118	116	110
Pb	6	17	5	5	5	5	4
La	12	15	11	11	11	11	9
Ce	24	32	26	27	25	21	22
Y	19.9	20	20.7	19	19	18	16
Th	4.0	7.0	4.0	5.0	3.0	5.0	3.0
U	2.0	1.0	1.0	1.0	1.0	1.0	2.0
Zr	92	142	95	92	90	82	72
Nb	5.0	5.0	4.5	4.5	4.0	4.0	3.0
V	223	70	218	209	217	205	207
Cr	115	27	111	111	121	144	195
Mn	1330	120	1310	1280	1280	1230	1250
Ni	79	25	76	77	82	83	95
Cu	71	10	75	72	72	67	60
Zn	77	89	76	74	75	70	65
Ga	16.0	13.5	16.5	16.0	16.5	16.0	15.0

	84 157	84 158	84 159	84 160	84 161	84 162	84 163
SiO ₂	53.52	53.73	53.34	53.32	54.93	54.93	55.97
TiO ₂	0.49	0.50	0.56	0.69	0.71	0.99	0.98
Al ₂ O ₃	13.78	13.95	15.12	17.56	15.03	16.98	16.02
Fe ₂ O ₃	0.93	1.26	1.16	1.49	1.74	2.04	2.20
FeO	7.58	7.12	7.27	7.00	8.56	7.52	7.93
MnO	0.17	0.17	0.16	0.15	0.18	0.14	0.15
MgO	9.08	8.75	6.78	4.08	4.32	2.18	2.31
CaO	11.56	11.64	11.50	11.20	9.83	9.09	8.70
Na ₂ O	1.44	1.38	1.61	1.98	2.13	2.55	2.58
K ₂ O	0.63	0.67	0.74	0.85	1.06	1.28	1.37
P ₂ O ₅	0.07	0.09	0.08	0.10	0.12	0.13	0.15
S	0.04	0.04	0.04	0.06	0.08	0.08	0.08
H ₂ O+	0.55	0.67	0.80	0.88	1.10	1.33	1.42
H ₂ O-	0.30	0.29	0.33	0.29	0.42	0.28	0.41
CO ₂	0.13	0.19	0.11	0.48	0.15	0.19	0.20
rest	0.15	0.15	0.13	0.12	0.15	0.15	0.15
	100.42	100.60	99.73	100.25	100.51	99.86	100.62
O=S	0.02	0.02	0.02	0.03	0.04	0.04	0.04
Total	100.40	100.58	99.71	100.22	100.47	99.82	100.58
Ba	150	160	175	200	250	290	305
Rb	23.0	23.5	27.0	30.5	39.0	48.5	51.0
Sr	105	106	118	141	130	158	147
Pb	3	5	6	6	6	8	9
La	9	9	11	12	15	17	18
Ce	19	16	22	29	30	36	43
Y	15	17	17	19	24	26	29
Th	2.0	5.0	5.0	4.0	7.0	6.0	7.0
U	3.0	2.0	1.0	3.0	1.0	2.0	1.0
Zr	67	71	82	96	118	137	149
Nb	3.5	3.5	4.0	5.0	5.0	6.0	7.0
V	222	223	209	191	215	209	192
Cr	244	218	90	6	1	1	1
Mn	1320	1300	1250	1140	1360	1120	1180
Ni	114	109	79	41	38	16	16
Cu	62	59	69	85	144	118	120
Zn	70	67	69	70	91	88	94
Ga	14.0	15.0	17.0	18.5	18.5	19.5	20.5

	84 164	84 165	84 166	84 167	84 167b	84 168	84 169
SiO ₂	56.00	55.79	57.67	54.19	54.19	51.04	53.70
TiO ₂	0.89	1.02	0.90	0.63	0.64	0.64	0.62
Al ₂ O ₃	16.74	15.81	16.48	14.74	14.67	15.06	14.50
Fe ₂ O ₃	2.07	1.99	1.48	0.61	0.75	2.70	1.80
FeO	7.44	8.32	7.88	8.30	8.12	5.36	7.01
MnO	0.14	0.15	0.13	0.18	0.17	0.27	0.18
MgO	2.11	2.17	1.57	6.68	6.62	5.29	6.88
CaO	8.89	8.84	7.94	10.89	10.92	12.15	9.94
Na ₂ O	2.60	2.46	2.85	1.94	1.83	1.74	2.01
K ₂ O	1.32	1.37	1.61	0.66	0.75	0.81	1.01
P ₂ O ₅	0.14	0.15	0.17	0.09	0.09	0.10	0.09
S	0.07	0.08	0.08	0.04	0.04	0.05	0.05
H ₂ O+	1.44	1.44	1.65	0.57	0.45	1.75	1.53
H ₂ O-	0.32	0.28	0.22	0.24	0.29	1.25	0.49
CO ₂	0.16	0.16	0.22	0.20	0.11	2.31	0.29
rest	0.15	0.15	0.15	0.14	0.14	0.18	0.21
	100.48	100.18	101.00	100.10	99.78	100.70	100.31
O=S	0.03	0.04	0.04	0.02	0.02	0.02	0.02
Total	100.45	100.14	100.96	100.08	99.76	100.68	100.29
Ba	305	310	345	190	190	460	785
Rb	48.5	50.0	60.0	28.0	32.5	29.3	38.8
Sr	149	141	151	115.7	115.1	146.4	137.7
Pb	9	8	9	6	5	5	5
La	17	18	20	11	11	12	11
Ce	39	43	4.7	27	24	26	26
Y	28	29	30	20.2	19.8	21.6	19.6
Th	5.0	5.0	6.0	3.0	3.0	5.0	4.0
U	1.0	2.0	2.0	1.0	1.0	2.0	1.0
Zr	146	150	168	93	93	95	90
Nb	7.0	7.0	7.5	5.0	4.0	4.5	4.5
V	177	183	132	227	221	239	226
Cr	1	1	1	116	114	122	117
Mn	1120	1170	1000	1380	1340	2110	1420
Ni	13	14	8	80	78	82	79
Cu	116	121	86	64	70	74	72
Zn	89	93	81	80	77	81	77
Ga	20.0	19.5	20.5	15.5	16.5	16.0	15.0

	84 170	84 171	84 172	84 173	84 174	84 175	84 176
SiO ₂	53.91	54.01	53.99	53.98	54.17	53.56	53.53
TiO ₂	0.62	0.63	0.62	0.60	0.59	0.53	0.52
Al ₂ O ₃	14.57	14.57	14.54	14.40	14.29	14.19	14.39
Fe ₂ O ₃	1.72	1.99	1.47	1.54	1.34	1.27	1.12
FeO	7.17	6.95	7.41	7.23	7.32	7.18	7.03
MnO	0.18	0.17	0.17	0.17	0.17	0.17	0.17
MgO	6.89	6.54	6.83	7.24	7.40	8.11	8.28
CaO	9.67	10.33	10.59	10.98	10.83	11.19	11.31
Na ₂ O	2.20	2.10	2.02	1.73	1.80	1.67	1.53
K ₂ O	1.18	0.97	0.90	0.74	0.76	0.70	0.66
P ₂ O ₅	0.09	0.09	0.09	0.09	0.09	0.08	0.07
S	0.04	0.05	0.04	0.05	0.04		
H ₂ O+	1.40	1.10	0.83	1.69	0.71	0.68	0.92
H ₂ O-	0.50	0.43	0.40	0.42	0.39	0.38	0.30
CO ₂	0.15	0.07	0.01	0.27	0.16	0.12	0.07
rest	0.29	0.17	0.15	0.15	0.15	0.15	0.15
	100.58	100.17	100.06	101.28	100.21	99.98	100.05
O=S	0.02	0.02	0.02	0.02	0.02		
Total	100.56	100.15	100.04	101.26	100.19	99.98	100.05
Ba	1400	330	210	180	175	165	150
Rb	44.5	36.5	33.0	26.0	27.0	24.5	24.0
Sr	167.8	164	139	111	112	110	107
Pb	6	6	5	6	5	5	4
La	12	11	11	11	14	9	9
Ce	28	24	24	22	25	19	21
Y	20.0	20	20	19	19	16	16
Th	5.0	6.0	5.0	5.0	3.0	3.0	4.0
U	1.0	1.0	1.0	1.0	1.0	1.0	3.0
Zr	88	94	91	90	86	78	75
Nb	4.5	4.0	4.5	4.0	4.0	3.0	3.5
V	220	228	225	227	224	223	216
Cr	115	110	121	147	166	213	228
Mn	1400	1320	1330	1330	1350	1320	1280
Ni	79	78	83	88	92	102	103
Cu	76	76	74	70	68	63	58
Zn	79	79	77	74	75	71	66
Ga	16.5	15.5	16.0	16.5	16.0	15.5	15.0

	84 177	84 178	84 179	84 180	84 181	84 182	84 183
SiO ₂	53.24	53.57	53.36	53.34	53.22	52.83	53.73
TiO ₂	0.48	0.47	0.49	0.53	0.53	0.52	0.54
Al ₂ O ₃	14.12	13.42	14.07	14.39	15.60	15.95	12.41
Fe ₂ O ₃	0.98	1.11	1.51	1.22	1.28	1.23	1.13
FeO	6.98	6.91	6.66	7.21	6.72	6.65	8.10
MnO	0.17	0.17	0.17	0.17	0.16	0.16	0.19
MgO	8.85	9.41	8.53	7.79	6.63	6.22	8.91
CaO	11.65	11.60	11.73	11.70	11.78	11.59	11.47
Na ₂ O	1.37	1.59	1.44	1.60	1.69	1.71	1.45
K ₂ O	0.60	0.63	0.63	0.67	0.73	0.70	0.65
P ₂ O ₅	0.07	0.07	0.07	0.07	0.15	0.08	0.07
S	0.02				0.04	0.03	0.02
H ₂ O+	1.01	0.83	0.41	1.08	0.99	1.59	1.28
H ₂ O-	0.30	0.85	0.39	0.30	0.33	0.36	0.31
CO ₂	0.23	0.13	0.13	0.10	0.14	0.17	0.11
rest	0.15	0.16	0.15	0.14	0.13	0.13	0.14
	100.22	100.92	99.74	100.31	100.12	99.92	100.51
O=S	0.01				0.02	0.01	0.01
Total	100.21	100.92	99.74	100.31	100.10	99.91	100.50
Ba	140	140	150	150	170	160	155
Rb	21.0	22.5	22.5	24.5	25.5	26.0	23.0
Sr	105	100	102	111	118	124	96
Pb	4	4	3	4	5	5	4
La	9	7	9	9	10	9	10
Ce	16	15	20	21	20	21	19
Y	15	14	15	16	17	17	17
Th	2.0	3.0	3.0	3.0	3.0	3.0	1.0
U	1.0	1.0	1.0	1.0	2.0	1.0	1.0
Zr	69	67	70	75	80	84	75
Nb	3.5	3.0	3.0	3.5	4.0	3.0	3.5
V	223	233	226	232	212	208	252
Cr	259	288	227	153	98	70	127
Mn	1290	1320	1300	1340	1250	1210	1490
Ni	118	126	107	98	82	75	107
Cu	52	51	55	61	60	60	67
Zn	67	67	67	69	68	67	75
Ga	14.0	14.0	14.5	15.0	16.5	15.5	14.0

	84 184	84 185	84 186	84 187	84 188	84 189	84 190
SiO ₂	53.60	53.70	53.48	53.76	53.67	54.13	54.20
TiO ₂	0.60	0.62	0.61	0.61	0.61	0.70	0.70
Al ₂ O ₃	15.20	15.04	16.16	13.02	14.54	15.76	16.05
Fe ₂ O ₃	1.24	1.66	1.40	1.65	1.56	1.63	1.80
FeO	7.35	7.03	7.08	8.11	7.61	7.24	7.10
MnO	0.17	0.17	0.16	0.19	0.18	0.16	0.16
MgO	6.49	6.23	5.44	7.46	6.46	4.86	4.51
CaO	11.24	10.90	11.18	10.87	10.89	10.92	10.49
Na ₂ O	1.80	1.94	1.86	1.77	1.90	1.97	2.03
K ₂ O	0.76	0.84	0.83	0.81	0.86	0.96	1.00
P ₂ O ₅	0.08	0.09	0.08	0.08	0.09	0.10	0.09
S	0.05	0.04	0.04	0.06	0.05	0.03	0.05
H ₂ O+	1.10	1.05	1.13	1.02	0.96	1.19	1.16
H ₂ O-	0.29	0.30	0.42	0.35	0.41	0.46	0.46
CO ₂	0.13	0.14	0.12	0.15	0.14	0.08	0.15
rest	0.13	0.14	0.13	0.14	0.14	0.13	0.13
	100.23	99.89	100.12	100.05	100.07	100.32	100.08
O=S	0.02	0.02	0.02	0.03	0.02	0.01	0.02
Total	100.21	99.87	100.10	100.02	100.05	100.31	100.06
Ba	185	195	190	190	195	210	230
Rb	28.0	31.5	30.5	28.5	31.5	35.0	36.5
Sr	117	121	127	105	120	133	132
Pb	5	5	6	5	6	5	6
La	11	11	11	11	11	13	14
Ce	22	24	25	22	23	29	28
Y	18	19	19	19	19	21	21
Th	3.0	2.0	2.0	4.0	4.0	5.0	6.0
U	2.0	3.0	1.0	1.0	2.0	1.0	1.0
Zr	86	92	93	87	91	104	108
Nb	4.5	4.5	4.5	4.0	4.0	4.5	5.0
V	219	221	201	243	229	218	202
Cr	74	59	39	62	45	17	14
Mn	1300	1300	1220	1470	1360	1260	1230
Ni	76	73	59	88	73	51	46
Cu	69	73	74	74	77	72	82
Zn	72	75	72	80	79	75	78
Ga	15.5	16.5	17.0	15.0	16.5	17.5	18.0

	84 191	84 192	84 193	84 194	84 195	84 196	84 197
SiO ₂	53.91	53.84	65.91	55.39	55.29	53.99	54.77
TiO ₂	0.73	0.67	0.85	0.76	0.75	0.64	0.64
Al ₂ O ₃	16.77	15.58	18.84	14.27	14.43	14.62	14.69
Fe ₂ O ₃	2.10	2.03	0.19	1.64	1.04	1.71	1.15
FeO	6.87	7.46	4.88	8.21	8.32	7.10	7.76
MnO	0.15	0.17	0.06	0.17	0.19	0.17	0.17
MgO	3.92	5.10	1.51	5.09	5.47	6.68	6.50
CaO	10.47	10.56	0.85	9.10	10.08	10.20	10.71
Na ₂ O	2.07	1.95	1.32	2.37	2.06	2.10	1.83
K ₂ O	1.00	0.95	3.34	1.23	0.77	0.93	0.80
P ₂ O ₅	0.11	0.10	0.20	0.11	0.12	0.09	0.10
S	0.04	0.03		0.09	0.07	0.04	0.04
H ₂ O+	1.10	1.10	1.03	0.98	0.89	1.07	0.95
H ₂ O-	0.35	0.40	0.22	0.50	0.35	0.52	0.34
CO ₂	0.07	0.04	0.01	0.19	0.74	0.10	0.10
rest	0.13	0.13	0.20	0.16	0.15	0.16	0.15
	99.79	100.11	99.41	100.26	100.72	100.12	100.70
O=S	0.02	0.01		0.04	0.03	0.02	0.02
Total	99.77	100.10	99.41	100.22	100.69	100.10	100.68
Ba	230	230	510	270	255	270	210
Rb	36.5	35.0	151.0	44.0	35.5	36.2	29.5
Sr	140	139	179	197	143	151.5	121
Pb	6	6	27	7	8	6	6
La	15	12	50	15	15	12	12
Ce	31	25	105	33	30	25	25
Y	22	21	36	25	24	20.7	20
Th	5.0	3.0	23.0	6.0	5.0	1.0	5.0
U	2.0	2.0	4.0	1.0	2.0	1.0	1.0
Zr	110	101	219	124	120	95	95
Nb	5.5	5.0	14.5	5.5	5.5	4.0	4.0
V	194	213	98	224	231	220	225
Cr	6	11	68	39	48	110	101
Mn	1180	1280	485	1310	1500	1300	1350
Ni	37	49	35	51	54	78	72
Cu	91	87	31	87	88	74	73
Zn	76	76	111	88	87	81	78
Ga	18.5	17.5	25.0	17.5	17.0	16.5	16.5

	84 198	84 199	84 200	84 201	84 202	84 203	84 204
SiO ₂	55.04	54.54	54.51	54.17	Not	54.44	53.79
TiO ₂	0.82	0.63	0.63	0.63	Analysed	0.63	0.62
Al ₂ O ₃	13.64	14.78	14.69	14.65		14.70	14.58
Fe ₂ O ₃	2.49	1.24	0.91	1.83		0.92	2.06
FeO	8.09	7.61	8.00	7.25		7.96	6.82
MnO	0.17	0.18	0.17	0.18		0.18	0.17
MgO	5.13	6.79	6.77	6.72		6.63	6.66
CaO	9.31	10.55	10.61	10.04		10.83	10.90
Na ₂ O	2.17	2.20	1.95	2.14		1.68	1.61
K ₂ O	1.19	0.83	0.98	1.09		0.90	0.88
P ₂ O ₅	0.12	0.09	0.09	0.10		0.09	0.09
S	0.03	0.05	0.04	0.04		0.04	0.03
H ₂ O+	0.85	0.65	0.71	1.13		0.45	0.63
H ₂ O-	0.49	0.28	0.28	0.45		0.41	0.67
CO ₂	0.06	0.13	0.13	0.10		0.08	0.20
rest	0.15	0.15	0.15	0.17		0.15	0.15
	99.75	100.70	100.62	100.69		100.09	99.86
O=S	0.01	0.02	0.02	0.02		0.02	0.01
Total	99.74	100.68	100.60	100.67		100.07	99.85
Ba	250	210	220	310		220	200
Rb	45.5	31.1	37.6	39.5		35.4	29.0
Sr	183	130.4	134.8	199		119.5	124
Pb	7	5	6	6		6	5
La	15	11	11	11		12	11
Ce	31	24	26	26		25	27
Y	25	20.2	20.0	20		20.3	20
Th	6.0	3.0	6.0	3.0		2.0	5.0
U	1.0	1.0	2.0	1.0		4.0	2.0
Zr	126	94	93	95		93	94
Nb	6.0	4.0	4.5	4.0		4.0	4.5
V	3203	233	219	219		231	224
Cr	13	115	109	110		114	114
Mn	1340	1370	1320	1360		1380	1330
Ni	46	81	78	78		78	79
Cu	81	77	76	75		78	74
Zn	77	79	77	79		78	78
Ga	17.5	16.5	16.5	16.0		15.5	16.5

	84 205	84 206	84 207	84 208	84 209	84 210	84 211
SiO ₂	53.94	53.85	54.06	54.09	54.40	53.65	53.77
TiO ₂	0.64	0.63	0.63	0.64	0.65	0.64	0.65
Al ₂ O ₃	14.86	14.61	14.65	14.59	14.61	14.63	14.75
Fe ₂ O ₃	2.08	0.82	0.69	1.05	1.18	1.36	1.38
FeO	6.66	7.94	8.07	7.77	7.82	7.68	7.57
MnO	0.17	0.18	0.17	0.17	0.17	0.17	0.17
MgO	6.13	6.63	6.70	6.65	6.65	6.68	6.83
CaO	10.27	10.85	11.03	10.92	10.96	10.93	10.69
Na ₂ O	1.92	1.91	1.80	1.76	1.76	1.82	2.04
K ₂ O	0.88	0.89	0.76	0.82	0.69	0.79	1.05
P ₂ O ₅	0.09	0.10	0.09	0.09	0.09	0.09	0.10
S	0.03	0.05	0.04	0.05	0.03	0.02	0.02
H ₂ O+	1.08	1.04	0.66	0.71	0.92	0.98	1.03
H ₂ O-	1.10	0.41	0.29	0.26	0.33	0.42	0.43
CO ₂	0.04	0.41	0.18	0.32	0.28	0.21	0.20
rest	0.15	0.15	0.14	0.14	0.14	0.14	0.15
	100.04	100.47	99.96	100.03	100.68	100.21	100.83
O=S	0.01	0.02	0.02	0.02	0.01	0.01	0.01
Total	100.03	100.45	99.94	100.01	100.67	100.20	100.82
Ba	215	205	180	200	185	190	200
Rb	31.8	34.4	23.0	27.5	24.0	27.5	46.0
Sr	123.0	125.0	119	119	120	121	117
Pb	12	7	6	6	6	5	6
La	11	11	11	11	11	12	11
Ce	26	22	26	26	24	25	22
Y	21.1	20.1	19	19	20	19	19
Th	3.0	3.0	4.0	5.0	2.0	4.0	5.0
U	1.0	2.0	1.0	2.0	2.0	1.0	2.0
Zr	96	93	92	94	86	88	85
Nb	5.0	4.5	4.0	4.0	5.0	4.0	4.5
V	232	224	227	221	224	222	225
Cr	108	114	116	110	112	109	110
Mn	1240	1370	1320	1290	1320	1300	1320
Ni	83	79	80	75	77	77	79
Cu	77	76	76	73	72	77	72
Zn	98	81	78	76	78	79	79
Ga	17.0	16.0	16.0	16.5	15.5	17.5	16.0

	84 212	84 213	84 214	84 215	84 216	84 217	84 218
SiO ₂	54.01	54.86	55.05	55.04	54.69	55.18	54.65
TiO ₂	0.64	0.68	0.67	0.67	0.65	0.66	0.66
Al ₂ O ₃	14.84	14.60	14.49	14.48	14.48	14.68	14.51
Fe ₂ O ₃	1.13	0.80	0.61	0.72	1.12	1.02	1.06
FeO	7.87	8.35	8.56	8.45	8.46	8.06	8.18
MnO	0.17	0.19	0.18	0.18	0.18	0.17	0.17
MgO	6.94	6.08	6.21	6.26	6.31	6.29	6.39
CaO	11.28	10.40	10.38	10.41	10.60	10.65	10.62
Na ₂ O	1.74	1.92	2.03	2.02	1.96	2.08	1.94
K ₂ O	0.71	0.76	1.05	0.98	0.89	0.91	0.90
P ₂ O ₅	0.10	0.11	0.11	0.11	0.11	0.11	0.10
S	0.02	0.07	0.05	0.05	0.06	0.05	0.05
H ₂ O+	0.84	0.99	0.55	0.57	0.48	0.61	0.58
H ₂ O-	0.44	0.29	0.20	0.20	0.23	0.20	0.19
CO ₂	0.10	0.11	0.07	0.06	0.03	0.11	0.17
rest	0.14	0.14	0.15	0.15	0.15	0.15	0.15
	100.97	100.35	100.36	100.35	100.40	100.93	100.32
O=S	0.01	0.03	0.02	0.02	0.03	0.02	0.02
Total	100.96	100.32	100.34	100.33	100.37	100.91	100.30
Ba	180	175	245	230	210	205	210
Rb	22.0	27.9	40.0	35.0	30.0	30.5	31.5
Sr	117	130.8	143.8	140	138	139	133
Pb	5	6	7	6	6	6	7
La	10	12	12	12	11	12	11
Ce	22	26	23	23	26	27	24
Y	18	21.2	21.0	20	20	20	20
Th	3.0	3.0	5.0	4.0	4.0	2.0	3.0
U	1.0	2.0	2.0	1.0	1.0	2.0	1.0
Zr	81	100	101	99	93	96	96
Nb	4.0	5.0	4.5	4.5	4.5	4.5	4.5
V	222	235	235	234	230	232	231
Cr	108	80	76	77	83	77	86
Mn	1300	1480	1380	1380	1360	1340	1330
Ni	78	70	69	68	65	68	71
Cu	77	76	74	74	83	76	75
Zn	77	79	78	78	80	78	78
Ga	16.0	16.5	17.0	17.0	16.5	17.0	16.5

	84 219	84 220	84 221	84 222	84 223	84 224	84 225
SiO ₂	56.61	54.53	53.41	52.89	54.06	54.37	55.98
TiO ₂	1.03	0.63	0.62	0.64	0.63	0.70	0.79
Al ₂ O ₃	13.14	14.62	14.69	14.68	14.65	14.40	14.14
Fe ₂ O ₃	2.50	1.30	1.73	1.81	1.61	1.77	1.60
FeO	8.90	7.84	7.09	7.28	7.45	7.66	7.29
MnO	0.18	0.17	0.18	0.18	0.17	0.17	0.16
MgO	3.93	6.51	6.76	6.49	6.66	5.87	5.70
CaO	8.56	10.76	9.58	10.16	10.80	10.29	9.62
Na ₂ O	2.19	1.94	2.50	2.45	1.90	1.94	2.08
K ₂ O	1.31	0.86	1.09	1.01	0.90	1.00	1.21
P ₂ O ₅	0.16	0.09	0.09	0.09	0.09	0.10	0.11
S	0.07	0.05	0.02	0.02	0.03	0.05	0.04
H ₂ O+	0.86	0.63	2.11	2.24	1.05	1.06	1.12
H ₂ O-	0.20	0.22	0.56	0.73	0.46	0.46	0.55
CO ₂	0.20	0.21	0.14	0.12	0.12	0.13	0.14
rest	0.17	0.15	0.16	0.16	0.15	0.14	0.15
	100.01	100.51	100.73	100.95	100.73	100.11	100.68
O=S	0.03	0.02	0.01	0.01	0.01	0.02	0.02
Total	99.98	100.49	100.72	100.94	100.72	100.09	100.66
Ba	315	200	235	255	205	230	255
Rb	46.0	31.5	42.9	37.6	34.5	36.5	45.0
Sr	138	134	204.6	185.6	141	134	128
Pb	9	5	5	5	5	5	7
La	19	11	11	11	11	13	16
Ce	44	25	24	22	25	29	32
Y	30	20	19.8	20.5	19	21	23
Th	7.0	3.0	5.0	5.0	4.0	5.0	7.0
U	3.0	1.0	1.0	1.0	2.0	3.0	2.0
Zr	146	91	94	95	92	105	129
Nb	7.0	4.0	4.0	4.0	4.5	5.0	6.0
V	237	223	225	232	220	220	213
Cr	5	104	116	118	103	45	46
Mn	1370	1330	1370	1360	1290	1330	1270
Ni	35	73	79	81	77	62	61
Cu	116	75	72	75	78	80	85
Zn	100	77	79	81	75	79	76
Ga	19.5	16.5	16.5	17.0	17.0	16.5	17.0

	84 226	84 227	84 228	84 229	84 230	84 231	84 232
SiO ₂	53.82	52.91	53.38	53.58	53.22	53.55	53.13
TiO ₂	0.59	0.53	0.52	0.52	0.51	0.49	0.45
Al ₂ O ₃	14.96	14.97	13.34	12.80	13.72	13.43	13.60
Fe ₂ O ₃	1.73	1.15	1.65	1.21	1.62	1.41	1.17
FeO	7.08	7.38	7.07	7.29	6.98	7.14	7.07
MnO	0.16	0.17	0.18	0.18	0.17	0.18	0.18
MgO	6.62	7.31	9.02	9.51	8.72	9.09	9.30
CaO	11.24	11.83	11.53	11.63	11.36	11.43	11.83
Na ₂ O	1.84	1.60	1.43	1.37	1.69	1.57	1.46
K ₂ O	0.77	0.66	0.66	0.66	0.72	0.66	0.58
P ₂ O ₅	0.08	0.07	0.07	0.07	0.08	0.07	0.06
S	0.03	0.04	0.04	0.03	0.04		
H ₂ O+	0.77	0.57	0.55	0.53	0.76	0.78	0.74
H ₂ O-	0.59	0.48	0.58	0.51	0.41	0.38	0.33
CO ₂	0.15	0.14	0.16	0.16	0.11	0.11	0.20
rest	0.13	0.14	0.15	0.16	0.15	0.15	0.15
	100.56	99.95	100.33	100.21	100.26	100.44	100.25
O=S	0.01	0.02	0.02	0.01	0.02		
Total	100.55	99.93	100.31	100.20	100.24	100.44	100.25
Ba	185	155	160	145	170	150	145
Rb	27.0	24.0	23.0	23.0	25.5	24.5	21.0
Sr	121	117	102	98	112	109	116
Pb	5	5	4	4	4	5	4
La	11	9	9	8	9	9	8
Ce	24	22	18	18	20	19	14
Y	18	16	16	16	16	15	14
Th	1.0	2.0	2.0	2.0	3.0	3.0	2.0
U	1.0	1.0	1.0	1.0	1.0	1.0	1.0
Zr	83	74	72	73	70	70	61
Nb	4.0	3.0	3.5	3.0	3.5	3.0	3.0
V	215	223	234	245	216	231	225
Cr	58	128	227	241	199	194	228
Mn	1250	1290	1360	1370	1280	1390	1360
Ni	70	83	112	119	104	108	115
Cu	73	66	61	57	57	56	50
Zn	71	69	69	68	66	68	64
Ga	16.0	15.0	15.0	13.5	15.0	13.5	13.5

	84 233	84 234	84 235	84 236	84 237	84 238	84 239
SiO ₂	53.61	53.21	52.87	53.33	53.20	51.63	53.41
TiO ₂	0.58	0.48	0.43	0.41	0.40	0.42	0.44
Al ₂ O ₃	13.52	13.04	12.67	13.28	12.95	12.91	12.62
Fe ₂ O ₃	1.75	1.18	1.19	1.40	0.75	1.38	1.77
FeO	7.24	7.28	7.01	6.70	7.24	6.87	6.64
MnO	0.18	0.18	0.18	0.17	0.18	0.18	0.18
MgO	8.15	9.72	10.54	10.27	10.76	10.91	10.51
CaO	11.01	11.71	11.97	12.11	12.09	12.28	11.77
Na ₂ O	1.68	1.39	1.33	1.33	1.30	1.12	1.38
K ₂ O	0.73	0.60	0.52	0.55	0.47	0.48	0.58
P ₂ O ₅	0.08	0.06	0.06	0.06	0.05	0.06	0.06
S		0.02		0.03			0.02
H ₂ O+	1.30	0.69	0.68	0.53	0.70	1.53	0.69
H ₂ O-	0.51	0.38	0.33	1.29	0.34	0.56	0.34
CO ₂	0.12	0.10	0.12	0.12	0.05	0.07	0.19
rest	0.15	0.15	0.16	0.15	0.16	0.16	0.16
	100.61	100.19	100.06	101.73	100.64	100.56	100.76
O=S		0.01		0.01			0.01
Total	100.61	100.18	100.06	101.72	100.64	100.56	100.75
Ba	165	145	125	130	115	125	130
Rb	26.5	20.5	18.5	20.0	17.0	17.5	20.5
Sr	121	101	92	100	98	106	97
Pb	5	4	2	4	2	3	4
La	10	8	7	7	7	7	7
Ce	24	19	15	16	15	15	19
Y	18	15	13	13	12	13	14
Th	4.0	2.0	3.0	3.0	2.0	3.0	3.0
U	2.0	1.0	1.0	1.0	1.0	1.0	1.0
Zr	83	65	57	56	54	55	60
Nb	4.0	2.5	2.0	2.5	2.5	2.5	2.5
V	232	235	236	231	234	238	237
Cr	156	253	292	280	314	306	294
Mn	1380	1380	1390	1340	1370	1390	1380
Ni	97	125	135	131	137	138	133
Cu	66	54	45	45	43	44	47
Zn	73	65	62	60	61	62	63
Ga	14.0	14.0	12.5	13.0	12.5	12.5	13.5

	84 240	84 241	84 242	84 243	84 244	84 245	84 246
SiO ₂	52.94	53.07	53.31	53.04	53.35	53.35	53.20
TiO ₂	0.41	0.42	0.44	0.40	0.42	0.47	0.41
Al ₂ O ₃	13.21	12.99	12.71	12.39	13.14	12.88	13.53
Fe ₂ O ₃	1.43	1.45	1.17	1.18	1.17	1.51	1.39
FeO	6.56	6.73	7.04	6.95	6.67	6.52	6.39
MnO	0.17	0.18	0.18	0.18	0.17	0.18	0.17
MgO	10.22	10.35	10.53	11.06	10.50	10.81	10.49
CaO	12.23	12.08	11.86	11.89	11.99	12.08	12.13
Na ₂ O	1.26	1.37	1.34	1.31	1.36	1.30	1.25
K ₂ O	0.49	0.51	0.59	0.50	0.52	0.49	0.52
P ₂ O ₅	0.06	0.06	0.06	0.05	0.06	0.06	0.06
S	0.03	0.02			0.02		0.03
H ₂ O+	0.59	0.81	0.62	0.83	0.59	0.76	0.60
H ₂ O-	0.32	0.24	0.35	0.25	0.32	0.25	0.31
CO ₂	0.13	0.10	0.16	0.18	0.11	0.13	0.58
rest	0.15	0.15	0.16	0.16	0.16	0.16	0.16
	100.20	100.53	100.52	100.37	100.55	100.95	101.22
O=S	0.01	0.01			0.01		0.01
Total	100.19	100.52	100.52	100.37	100.54	100.95	101.21
Ba	120	120	140	115	125	115	125
Rb	17.0	18.0	21.0	19.0	18.5	17.5	18.5
Sr	97	100	98	97	98	96	100
Pb	3	3	4	3	3	4	4
La	6	7	7	6	7	7	6
Ce	16	19	18	14	17	14	17
Y	13	13	14	13	13	12	13
Th	3.0	3.0	2.0	3.0	2.0	1.0	1.0
U	1.0	1.0	1.0	3.0	3.0	1.0	2.0
Zr	57	58	61	55	57	53	55
Nb	1.5	2.5	2.5	3.0	2.5	3.0	3.0
V	229	232	236	236	233	235	227
Cr	273	276	291	326	315	322	305
Mn	1330	1360	1370	1390	1340	1360	1310
Ni	131	133	136	141	136	139	133
Cu	45	46	47	44	42	44	41
Zn	60	61	63	63	61	62	59
Ga	13.0	13.0	12.5	13.0	13.5	12.5	13.5

	84 247	84 248	84 249	84 250	84 251	84 252	84 253
SiO ₂	53.00	53.01	53.05	53.15	53.55	53.09	53.26
TiO ₂	0.42	0.43	0.41	0.43	0.45	0.44	0.42
Al ₂ O ₃	11.97	13.32	12.41	14.00	11.56	13.43	12.81
Fe ₂ O ₃	1.46	1.19	1.13	1.30	1.10	1.13	1.23
FeO	7.05	6.99	6.98	6.60	7.52	6.83	6.80
MnO	0.18	0.17	0.18	0.16	0.19	0.17	0.17
MgO	11.01	10.03	10.98	9.46	10.93	9.77	10.62
CaO	11.74	11.91	11.83	12.10	11.76	12.26	11.92
Na ₂ O	1.33	1.24	1.22	1.38	1.31	1.37	1.28
K ₂ O	0.49	0.59	0.51	0.57	0.54	0.53	0.51
P ₂ O ₅	0.05	0.06	0.06	0.06	0.06	0.06	0.06
S		0.02					0.03
H ₂ O+	0.64	0.64	0.76	0.61	0.69	0.86	0.60
H ₂ O-	0.37	0.31	0.44	0.29	0.30	0.24	0.25
CO ₂	0.22	0.12	0.12	0.05	0.10	0.08	0.07
rest	0.16	0.15	0.16	0.14	0.16	0.14	0.16
	100.09	100.18	100.24	100.30	100.22	100.40	100.19
O=S		0.01					0.01
Total	100.09	100.17	100.24	100.30	100.22	100.40	100.18
Ba	125	140	125	140	140	130	125
Rb	17.0	20.5	18.5	20.5	19.5	19.0	19.0
Sr	91	104	93	106	90	97	98
Pb	3	4	3	3	3	4	3
La	6	8	7	7	8	7	7
Ce	13	14	13	16	15	15	16
Y	13	14	13	14	14	14	13
Th	1.0	2.0	3.0	4.0	1.0	1.0	1.0
U	3.0	1.0	1.0	1.0	1.0	1.0	1.0
Zr	55	60	56	60	59	59	56
Nb	3.0	2.5	2.5	3.0	2.5	2.5	2.5
V	238	229	236	214	254	231	232
Cr	311	266	306	233	257	172	308
Mn	1390	1320	1370	1260	1450	1340	1340
Ni	142	126	140	116	141	125	135
Cu	50	45	43	46	50	42	44
Zn	65	63	63	60	66	61	62
Ga	13.0	12.5	12.5	14.5	12.5	13.5	13.0

	84 254	84 255	84 256	84 257	84 258	84 259	84 260
SiO ₂	53.15	52.49	52.68	53.61	52.76	53.22	52.85
TiO ₂	0.43	0.41	0.44	0.40	0.42	0.43	0.43
Al ₂ O ₃	13.87	13.40	12.13	11.03	13.25	12.60	12.75
Fe ₂ O ₃	1.32	1.47	1.54	1.60	1.08	1.17	1.15
FeO	6.43	6.95	7.10	7.03	7.03	7.17	7.15
MnO	0.16	0.17	0.18	0.19	0.17	0.18	0.17
MgO	9.59	10.23	10.90	11.94	10.01	10.56	10.32
CaO	12.12	12.29	11.89	11.92	12.15	11.92	11.86
Na ₂ O	1.38	1.39	1.21	1.21	1.36	1.32	1.33
K ₂ O	0.49	0.50	0.51	0.46	0.52	0.53	0.54
P ₂ O ₅	0.06	0.06	0.06	0.06	0.06	0.06	0.06
S	0.02	0.03	0.03		0.02		
H ₂ O+	0.67	0.80	0.59	0.56	0.69	0.69	0.67
H ₂ O-	0.27	0.24	0.25	0.28	0.30	0.29	0.23
CO ₂	0.12	0.09	0.11	0.11	0.09	0.09	0.08
rest	0.15	0.15	0.16	0.16	0.15	0.15	0.16
	100.23	100.67	99.78	100.56	100.06	100.38	99.75
O=S	0.01	0.01	0.01		0.01		
Total	100.22	100.66	99.77	100.56	100.05	100.38	99.75
Ba	125	115	120	115	125	125	130
Rb	17.0	17.5	18.0	16.5	18.5	19.5	20.5
Sr	102	98	90	81	96	92	95
Pb	5	4	4	3	4	3	4
La	8	7	7	6	7	7	7
Ce	15	13	15	14	15	16	14
Y	14	13	14	13	14	13	14
Th	3.0	2.0	3.0	2.0	2.0	1.0	1.0
U	1.0	1.0	1.0	1.0	1.0	1.0	1.0
Zr	61	56	58	52	58	58	60
Nb	2.5	2.0	3.0	2.0	2.5	2.0	3.0
V	214	228	239	251	230	235	238
Cr	255	285	301	333	260	283	287
Mn	1270	1350	1390	1440	1320	1360	1350
Ni	119	129	139	151	128	133	135
Cu	46	45	49	46	48	46	49
Zn	61	63	65	64	63	62	64
Ga	14.0	13.0	12.5	11.5	14.0	13.0	12.5

	84 261	84 262	84 263	84 264	84 265	84 266	84 267
SiO ₂	52.90	53.54	53.46	53.82	53.98	54.99	54.81
TiO ₂	0.42	0.42	0.42	0.66	0.64	0.79	0.71
Al ₂ O ₃	12.41	12.72	10.94	15.99	16.18	16.45	14.86
Fe ₂ O ₃	1.05	1.03	1.18	1.85	1.59	1.58	1.71
FeO	7.47	7.13	7.52	7.27	7.28	7.78	8.12
MnO	0.18	0.18	0.19	0.18	0.15	0.16	0.18
MgO	10.78	10.72	12.00	5.07	5.05	3.66	5.38
CaO	12.05	12.05	11.65	10.88	11.43	10.39	10.51
Na ₂ O	1.24	1.28	1.36	1.92	1.75	2.09	1.94
K ₂ O	0.49	0.50	0.50	1.02	0.53	0.98	0.98
P ₂ O ₅	0.06	0.06	0.06	0.11	0.10	0.12	0.11
S				0.04	0.04	0.06	0.06
H ₂ O+	0.60	0.60	0.61	0.69	1.16	0.75	0.48
H ₂ O-	0.27	0.26	0.27	0.58	0.30	0.47	0.38
CO ₂	0.10	0.10	0.12	0.07	0.08	0.06	0.10
rest	0.15	0.16	0.17	0.13	0.13	0.14	0.14
	100.17	100.75	100.45	100.28	100.39	100.47	100.47
O=S				0.02	0.02	0.03	0.03
Total	100.17	100.75	100.45	100.26	100.37	100.44	100.44
Ba	120	120	115	230	205	250	225
Rb	16.5	17.5	17.5	35.5	12.5	26.5	36.0
Sr	91	92	79	143	170	148	124
Pb	5	3	4	7	7	7	6
La	6	6	7	13	12	14	14
Ce	15	14	14	28	23	33	28
Y	13	14	13	21	20	24	22
Th	2.0	2.0	2.0	4.0	4.0	6.0	1.0
U	2.0	2.0	1.0	1.0	1.0	1.0	2.0
Zr	55	58	56	104	96	122	108
Nb	3.0	2.5	2.5	4.5	4.0	5.5	5.0
V	234	236	257	200	209	200	225
Cr	287	325	375	17	14	1	11
Mn	1390	1360	1450	1380	1190	1210	1360
Ni	138	143	157	49	50	32	52
Cu	50	45	48	86	87	103	87
Zn	68	63	67	75	75	83	83
Ga	13.0	12.5	12.5	18.0	16.5	19.0	17.5

	84 268	84 269	84 270	84 271	84 272	84 273	84 274
SiO ₂	54.14	54.45	53.75	53.52	54.22	53.67	54.19
TiO ₂	0.70	0.65	0.66	0.69	0.69	0.64	0.61
Al ₂ O ₃	15.45	15.70	16.23	15.30	15.55	15.77	15.13
Fe ₂ O ₃	2.05	1.96	2.25	2.47	1.83	2.21	1.78
FeO	7.28	7.18	6.68	6.93	7.32	6.78	7.11
MnO	0.16	0.16	0.16	0.16	0.16	0.16	0.16
MgO	5.12	5.61	5.29	5.78	5.60	5.62	6.10
CaO	10.80	11.12	11.25	11.12	10.31	10.65	10.93
Na ₂ O	1.97	2.03	1.97	1.86	2.37	2.35	1.85
K ₂ O	0.95	0.87	0.86	0.88	0.99	0.86	0.87
P ₂ O ₅	0.12	0.10	0.13	0.11	0.11	0.10	0.10
S	0.05	0.05	0.02	0.07	0.06	0.06	
H ₂ O+	0.49	0.49	0.54	0.74	0.82	1.05	1.26
H ₂ O-	0.42	0.44	0.42	0.66	0.41	0.33	0.34
CO ₂	0.11	0.13	0.03	0.16	0.12	0.13	0.11
rest	0.13	0.13	0.13	0.14	0.13	0.13	0.13
	99.94	101.07	100.37	100.59	100.69	100.51	100.67
O=S	0.02	0.02	0.01	0.03	0.03	0.03	
Total	99.92	101.05	100.36	100.56	100.66	100.48	100.67
Ba	210	205	205	205	230	210	210
Rb	34.0	31.0	30.5	31.5	36.0	31.5	31.0
Sr	126	125	127	130	123	118	119
Pb	6	6	6	5	6	5	6
La	13	11	11	13	12	11	11
Ce	30	26	25	26	28	24	26
Y	21	20	20	20	21	19	20
Th	8.0	4.0	4.0	4.0	5.0	4.0	4.0
U	2.0	1.0	1.0	3.0	1.0	2.0	1.0
Zr	105	94	96	96	100	93	96
Nb	5.5	4.0	5.0	4.0	4.0	4.5	5.0
V	217	211	209	231	211	202	206
Cr	12	19	20	24	25	29	39
Mn	1270	1260	1250	1270	1270	1220	1260
Ni	50	53	53	60	58	57	63
Cu	84	81	78	82	82	77	71
Zn	78	74	74	75	73	70	73
Ga	17.0	17.5	18.0	17.0	17.5	16.0	16.5

	84 275	84 276	84 277	84 278	84 279	84 280	84 281
SiO ₂	54.18	53.76	53.52	54.18	54.20	53.84	53.96
TiO ₂	0.67	0.66	0.64	0.65	0.66	0.67	0.67
Al ₂ O ₃	15.47	14.69	14.32	13.25	13.88	14.45	14.20
Fe ₂ O ₃	2.14	1.99	1.94	1.66	1.60	2.07	1.85
FeO	6.78	7.24	7.20	7.83	7.63	7.31	7.46
MnO	0.16	0.17	0.17	0.18	0.17	0.17	0.17
MgO	5.75	6.33	6.78	7.23	6.82	6.26	6.40
CaO	11.06	10.77	11.20	10.69	10.95	10.90	10.55
Na ₂ O	1.90	1.77	1.66	1.81	1.87	1.77	2.19
K ₂ O	0.88	0.86	0.79	0.87	0.87	0.85	0.90
P ₂ O ₅	0.10	0.09	0.09	0.09	0.10	0.11	0.10
S	0.03					0.02	0.04
H ₂ O+	1.28	1.37	1.36	1.19	1.18	1.55	1.18
H ₂ O-	0.30	0.36	0.38	0.26	0.26	0.34	0.28
CO ₂	0.13	0.11	0.11	0.07	0.10	0.07	0.12
rest	0.13	0.13	0.13	0.14	0.14	0.13	0.14
	100.96	100.30	100.29	100.10	100.43	100.51	100.21
O=S	0.01					0.01	0.02
Total	100.95	100.30	100.29	100.10	100.43	100.50	100.19
Ba	210	210	190	215	215	195	215
Rb	32.0	31.5	28.5	31.5	31.0	31.5	32.5
Sr	113	107	102	101	104	106	113
Pb	6	6	5	6	6	6	5
La	12	11	11	11	12	13	12
Ce	24	23	24	25	25	26	27
Y	20	20	19	20	20	20	20
Th	3.0	3.0	3.0	4.0	4.0	3.0	6.0
U	1.0	1.0	1.0	1.0	2.0	3.0	1.0
Zr	97	97	91	93	95	99	97
Nb	4.0	4.5	5.0	4.5	4.5	5.0	4.0
V	206	213	222	230	221	218	213
Cr	37	44	60	68	64	53	55
Mn	1230	1310	1310	1400	1350	1310	1320
Ni	59	66	77	81	74	68	71
Cu	72	77	71	74	73	75	79
Zn	73	77	74	79	74	76	75
Ga	17.0	17.5	16.0	15.0	16.5	17.0	16.0

	84 282	84 283	84 284	84 285	84 286	84 287	84 288
SiO ₂	53.51	53.57	53.52	53.39	53.12	53.2	53.47
TiO ₂	0.65	0.63	0.60	0.55	0.48	0.45	0.42
Al ₂ O ₃	15.13	14.88	15.17	15.25	13.07	12.56	12.46
Fe ₂ O ₃	2.56	2.22	2.10	1.74	1.69	1.99	1.64
FeO	6.47	6.79	6.58	6.46	6.62	6.29	6.49
MnO	0.15	0.17	0.16	0.16	0.17	0.18	0.17
MgO	5.93	6.17	6.56	7.18	9.55	10.40	11.04
CaO	10.96	11.03	11.39	11.62	11.95	11.83	11.54
Na ₂ O	2.19	2.09	2.09	1.66	1.62	1.26	1.38
K ₂ O	0.89	0.87	0.81	0.73	0.59	0.56	0.51
P ₂ O ₅	0.15	0.12	0.11	0.09	0.08	0.07	0.06
S	0.02	0.05	0.05	0.03	0.03		0.03
H ₂ O+	0.82	0.70	0.83	0.75	0.55	0.57	0.56
H ₂ O-	0.44	0.33	0.40	0.43	0.39	0.39	0.57
CO ₂	0.10	0.12	0.15	0.09	0.11	0.23	0.14
rest	0.13	0.14	0.13	0.13	0.15	0.16	0.17
	100.10	99.88	100.65	100.26	100.17	100.14	100.65
O=S	0.01	0.02	0.02	0.01	0.01		0.01
Total	100.09	99.86	100.63	100.25	100.16	100.14	100.64
Ba	210	205	195	175	140	130	120
Rb	32.0	31.0	29.0	26.0	21.0	20.0	18.0
Sr	117	119	116	111	93	91	87
Pb	6	6	6	5	4	3	3
La	12	11	10	10	8	8	8
Ce	26	22	23	21	18	16	16
Y	20	20	18	17	15	14	13
Th	4.0	2.0	3.0	3.0	5.0	3.0	4.0
U	2.0	1.0	1.0	1.0	1.0	1.0	1.0
Zr	94	95	86	78	66	62	59
Nb	4.0	4.5	4.0	4.0	3.0	2.5	2.0
V	203	210	207	206	233	231	224
Cr	51	60	81	120	255	317	372
Mn	1200	1290	1250	1250	1330	1360	1340
Ni	63	70	75	86	120	134	143
Cu	73	77	73	64	53	50	49
Zn	73	74	69	64	65	63	61
Ga	17.5	16.5	16.0	16.0	14.0	14.0	13.5

	84 289	84 290	84 291	84 292	84 293	84 294	84 295
SiO ₂	53.29	53.17	52.86	53.27	53.39	53.41	54.00
TiO ₂	0.43	0.42	0.45	0.48	0.51	0.55	0.56
Al ₂ O ₃	12.60	13.60	14.24	14.24	14.48	14.31	14.52
Fe ₂ O ₃	1.47	1.48	1.57	1.13	1.08	1.13	1.04
FeO	6.69	6.46	6.36	6.94	7.06	7.32	7.37
MnO	0.17	0.16	0.16	0.16	0.16	0.16	0.16
MgO	10.90	10.64	9.58	9.21	8.55	7.88	7.76
CaO	11.30	11.18	11.34	11.19	11.18	11.17	11.03
Na ₂ O	1.26	1.40	1.41	1.39	1.53	1.77	1.74
K ₂ O	0.53	0.52	0.60	0.64	0.68	0.74	0.75
P ₂ O ₅	0.06	0.09	0.08	0.08	0.08	0.17	0.10
S			0.03	0.03	0.03	0.04	0.04
H ₂ O+	0.58	0.65	0.66	0.63	0.72	1.15	0.84
H ₂ O-	0.42	0.53	0.53	0.48	0.44	0.34	0.29
CO ₂	0.28	0.08	0.18	0.14	0.18	0.15	0.15
rest	0.17	0.16	0.15	0.15	0.15	0.15	0.15
	100.15	100.54	100.20	100.16	100.22	100.44	100.50
O=S			0.01	0.01	0.01	0.02	0.02
Total	100.15	100.54	100.19	100.15	100.21	100.42	100.48
Ba	125	130	140	145	155	175	180
Rb	18.5	17.0	21.5	22.5	24.0	25.5	27.5
Sr	99	119	101	99	103	99	107
Pb	4	3	3	4	4	4	4
La	7	6	8	9	9	10	12
Ce	13	14	15	17	20	22	20
Y	14	13	14	15	16	17	17
Th	2.0	3.0	3.0	3.0	4.0	2.0	5.0
U	1.0	1.0	2.0	1.0	1.0	2.0	1.0
Zr	59	57	65	68	74	80	80
Nb	3.0	2.5	3.0	3.0	3.0	4.0	4.0
V	223	202	203	211	208	211	208
Cr	366	349	302	274	241	190	180
Mn	1330	1250	1250	1260	1260	1270	1270
Ni	143	135	122	116	107	94	93
Cu	48	43	50	54	61	64	65
Zn	63	61	62	64	67	67	68
Ga	13.5	13.5	14.5	14.0	15.5	15.5	15.5

	84 296	84 297	84 298	84 299	84 300	84 301	84 302
SiO ₂	54.05	54.14	54.20	54.01	44.57	54.08	54.58
TiO ₂	0.58	0.60	0.61	0.63	0.67	0.63	0.66
Al ₂ O ₃	14.14	14.36	14.30	14.49	15.64	14.64	14.43
Fe ₂ O ₃	1.67	2.03	1.88	2.04	3.01	2.44	1.51
FeO	7.13	6.83	7.06	6.99	5.29	6.55	7.93
MnO	0.17	0.17	0.17	0.17	0.24	0.17	0.17
MgO	7.77	7.29	7.13	6.78	3.91	6.48	6.12
CaO	10.51	10.63	10.85	10.41	13.24	10.26	10.51
Na ₂ O	1.99	1.98	1.86	2.07	2.14	1.76	1.86
K ₂ O	0.81	0.80	0.84	0.94	0.84	0.86	0.72
P ₂ O ₅	0.12	0.13	0.13	0.14	0.09	0.09	0.10
S	0.04		0.04		0.02	0.03	0.05
H ₂ O+	0.94	0.90	0.73	0.92	3.04	1.20	1.08
H ₂ O-	0.46	0.41	0.53	0.64	1.30	0.85	0.42
CO ₂	0.22	0.18	0.19	0.15	6.56	0.25	0.49
rest	0.15	0.15	0.15	0.17	0.17	0.15	0.14
	100.75	100.60	100.67	100.55	100.73	100.44	100.77
O=S	0.02		0.02		0.01	0.01	0.02
Total	100.73	100.60	100.65	100.55	100.72	100.43	100.75
Ba	185	190	215	400	225	230	205
Rb	30.0	30.0	27.0	33.0	28.0	30.7	24.5
Sr	109	137	132	147	199	150.3	134
Pb	6	5	6	6	6	6	6
La	11	10	12	12	11	11	12
Ce	22	22	23	23	26	26	24
Y	18	19	19	20	21	20.2	21
Th	2.0	5.0	3.0	3.0	4.0	4.0	3.0
U	1.0	1.0	1.0	1.0	1.0	1.0	1.0
Zr	85	89	91	94	102	97	98
Nb	4.0	4.0	4.5	4.0	4.5	4.5	4.5
V	222	220	218	222	270	227	234
Cr	154	134	131	110	129	107	76
Mn	1290	1300	1290	1340	1870	1320	1330
Ni	90	83	83	77	82	76	71
Cu	66	69	73	75	76	75	74
Zn	71	72	73	76	80	77	81
Ga	16.0	16.0	15.0	16.5	17.0	16.5	17.5

	84 303	84 304	84 305	84 306	84 307	84 308	84 309
SiO ₂	55.27	55.02	54.95	54.24	53.95	54.48	53.84
TiO ₂	0.71	0.66	0.68	0.62	0.70	0.64	0.60
Al ₂ O ₃	15.14	14.42	14.74	14.76	17.44	14.71	15.57
Fe ₂ O ₃	1.96	1.52	1.99	1.47	2.44	1.39	1.73
FeO	6.25	7.74	6.45	7.65	5.46	7.93	6.73
MnO	0.13	0.17	0.14	0.16	0.12	0.17	0.16
MgO	5.49	6.20	5.75	6.17	4.50	6.48	6.33
CaO	10.43	10.50	10.53	10.79	10.84	10.71	11.28
Na ₂ O	1.81	1.77	1.94	1.86	2.06	1.80	1.88
K ₂ O	0.70	0.88	0.76	0.81	0.96	0.69	0.69
P ₂ O ₅	0.10	0.09	0.10	0.09	0.12	0.09	0.09
S	0.05	0.04	0.04	0.04	0.05	0.05	0.05
H ₂ O+	1.26	0.75	1.27	0.62	1.11	0.75	0.67
H ₂ O-	0.87	0.32	0.70	0.27	0.69	0.35	0.26
CO ₂	0.56	0.19	0.27	0.17	0.16	0.14	0.16
rest	0.15	0.15	0.15	0.14	0.14	0.14	0.13
	100.88	100.42	100.46	99.86	100.74	100.52	100.17
O=S	0.02	0.02	0.02	0.02	0.02	0.02	0.02
Total	100.86	100.40	100.44	99.84	100.72	100.50	100.15
Ba	225	215	215	205	220	190	185
Rb	26.5	30.5	30.0	29.5	35.0	28.5	27.0
Sr	151	136	146	136	151	127	134
Pb	6	5	6	5	6	6	5
La	12	11	12	11	12	11	9
Ce	26	25	25	18	27	25	20
Y	19	20	20	19	20	19	17
Th	5.0	5.0	4.0	3.0	4.0	4.0	2.0
U	1.0	1.0	1.0	1.0	1.0	1.0	2.0
Zr	100	100	106	91	101	93	84
Nb	4.5	5.0	5.0	4.5	4.5	4.0	4.0
V	245	229	246	228	213	225	217
Cr	78	76	79	78	30	103	68
Mn	1040	1300	1090	1270	960	1280	1210
Ni	70	67	70	68	47	73	73
Cu	76	75	78	81	77	77	64
Zn	86	79	85	79	76	77	71
Ga	17.5	16.5	17.5	17.0	17.5	16.5	17.5

	84 310	84 311	84 312	84 313	84 314	84 315	84 316
SiO ₂	55.65	54.12	53.27	54.65	53.37	54.24	54.23
TiO ₂	0.77	0.67	0.57	0.62	0.52	0.57	0.58
Al ₂ O ₃	15.77	13.87	14.88	11.58	15.91	13.76	14.93
Fe ₂ O ₃	1.68	1.18	1.03	1.18	0.78	0.97	0.94
FeO	7.98	8.65	8.09	8.75	7.17	7.97	7.63
MnO	0.16	0.18	0.17	0.19	0.15	0.18	0.17
MgO	3.93	6.58	6.70	8.99	6.87	8.16	7.21
CaO	9.81	10.45	11.08	10.66	11.28	11.11	11.05
Na ₂ O	2.42	1.92	1.86	1.41	1.98	1.59	1.63
K ₂ O	0.94	0.92	0.73	0.73	0.74	0.68	0.75
P ₂ O ₅	0.14	0.11	0.08	0.09	0.07	0.09	0.09
S	0.06	0.05	0.04	0.04	0.04	0.04	0.06
H ₂ O+	0.97	0.94	0.70	0.77	0.69	0.69	0.61
H ₂ O-	0.29	0.28	0.29	0.40	0.22	0.28	0.25
CO ₂	0.15	0.14	0.12	0.14	0.19	0.16	0.17
rest	0.14	0.14	0.13	0.16	0.13	0.15	0.14
	100.86	100.20	99.74	100.36	100.11	100.64	100.44
O=S	0.03	0.02	0.02	0.02	0.02	0.02	0.03
Total	100.83	100.18	99.72	100.34	100.09	100.62	100.41
Ba	255	205	185	170	175	165	175
Rb	31.5	35.5	26.0	24.0	30.5	22.5	25.0
Sr	143	120	124	92	132	108	118
Pb	7	6	5	5	4	5	5
La	15	12	10	10	10	9	10
Ce	35	26	23	22	18	20	20
Y	24	20	17	19	15	17	18
Th	3.0	5.0	1.0	3.0	5.0	3.0	3.0
U	1.0	1.0	1.0	3.0	1.0	1.0	1.0
Zr	120	93	77	88	73	78	84
Nb	5.5	4.5	3.5	3.5	2.5	3.5	4.5
V	203	239	220	259	201	232	224
Cr	11	47	78	178	120	166	137
Mn	1250	1420	1300	1490	1200	1380	1300
Ni	37	69	73	109	79	98	86
Cu	112	83	73	72	59	66	64
Zn	87	82	76	82	65	74	72
Ga	18.0	16.5	17.0	14.5	17.5	15.5	17.0

	84 317	84 318	84 319	84 320	84 321	84 322	84 323
SiO ₂	54.15	54.54	54.45	54.92	54.94	55.08	55.17
TiO ₂	0.56	0.60	0.61	0.64	0.66	0.67	0.67
Al ₂ O ₃	14.34	14.30	14.18	14.43	14.41	14.52	14.48
Fe ₂ O ₃	0.94	0.97	0.77	1.00	1.02	0.99	1.19
FeO	7.65	7.82	8.09	8.00	8.08	8.08	7.81
MnO	0.17	0.17	0.17	0.17	0.17	0.17	0.17
MgO	7.80	7.55	7.36	6.57	6.38	6.30	6.26
CaO	11.06	10.90	10.68	10.57	10.45	10.46	10.44
Na ₂ O	1.60	1.80	1.73	1.89	1.80	1.64	1.87
K ₂ O	0.71	0.73	0.81	0.88	0.96	0.93	0.92
P ₂ O ₅	0.08	0.09	0.09	0.09	0.10	0.10	0.10
S	0.05	0.05	0.05	0.05	0.05	0.05	0.05
H ₂ O+	0.60	0.60	0.68	0.60	0.86	0.72	0.78
H ₂ O-	0.26	0.25	0.17	0.23	0.26	0.25	0.25
CO ₂	0.11	0.14	0.13	0.20	0.21	0.25	0.26
rest	0.15	0.15	0.15	0.15	0.15	0.15	0.15
	100.23	100.66	100.12	100.39	100.50	100.36	100.57
O=S	0.02	0.02	0.02	0.02	0.02	0.02	0.02
Total	100.21	100.64	100.10	100.37	100.48	100.34	100.55
Ba	170	180	200	210	220	240	230
Rb	23.0	24.5	30.0	31.0	34.5	33.0	32.7
Sr	113	117	123	129	137	130	128.9
Pb	4	5	5	6	5	5	6
La	10	10	10	10	11	12	11
Ce	20	21	25	25	24	26	27
Y	17	18	19	19	20	20	20.4
Th	3.0	2.0	4.0	3.0	4.0	8.0	6.0
U	1.0	2.0	1.0	1.0	2.0	1.0	1.0
Zr	81	85	88	95	98	100	101
Nb	3.0	3.5	4.0	4.0	4.5	4.5	4.5
V	226	231	225	232	226	230	235
Cr	169	163	148	99	86	79	80
Mn	1300	1310	1310	1290	1300	1290	1290
Ni	97	91	86	73	72	69	71
Cu	64	68	71	72	74	72	72
Zn	72	74	74	77	78	78	79
Ga	15.5	16.0	16.0	16.5	17.0	16.0	17.0

	84 324
SiO ₂	54.87
TiO ₂	0.66
Al ₂ O ₃	14.41
Fe ₂ O ₃	1.15
FeO	7.71
MnO	0.17
MgO	6.16
CaO	10.29
Na ₂ O	1.81
K ₂ O	0.88
P ₂ O ₅	0.10
S	0.05
H ₂ O+	0.91
H ₂ O-	0.40
CO ₂	0.66
rest	0.15
	100.38
O=S	0.02
Total	100.36

Ba	235
Rb	32.4
Sr	129.2
Pb	6
La	11
Ce	25
Y	20.8
Th	5.0
U	1.0
Zr	101
Nb	4.5
V	233
Cr	79
Mn	1280
Ni	68
Cu	73
Zn	80
Ga	17.0

XRF, INAA AND WET-CHEMICAL ANALYSES OF CHILLED MARGIN
DOLERITES

	84 101	84 106	84 107	84 108	84 109	84 111	84 115
SiO ₂	54.42	54.00	53.73	54.02	54.43	53.99	53.94
TiO ₂	0.63	0.63	0.63	0.63	0.66	0.64	0.62
Al ₂ O ₃	14.54	14.69	14.60	14.66	14.42	14.66	14.45
Fe ₂ O ₃	0.86	1.03	1.47	1.19	1.59	1.22	1.36
FeO	7.99	7.83	7.53	7.70	7.60	7.59	7.49
MnO	0.17	0.19	0.19	0.18	0.18	0.17	0.17
MgO	6.68	6.71	6.78	6.78	6.46	6.83	6.76
CaO	11.05	10.69	10.59	10.53	10.04	10.32	10.30
CaO _{INAA}	11.1	10.9	11.0	10.3	10.4	10.5	10.5
Na ₂ O	1.99	2.15	2.09	2.14	2.14	1.89	2.14
Na ₂ O _{INAA}	1.89	2.06	1.90	1.94	1.98	1.81	2.13
K ₂ O	0.60	0.85	0.83	0.84	0.98	1.02	0.93
P ₂ O ₅	0.09	0.09	0.09	0.09	0.10	0.09	0.09
S	0.04	0.03	0.03	0.07	0.05	0.05	0.09
H ₂ O +	0.93	0.62	0.80	0.93	1.35	1.10	0.86
H ₂ O -	0.26	0.53	0.68	0.58	0.49	0.44	0.36
CO ₂	0.20	0.16	0.15	0.22	0.14	0.14	0.20
rest	0.15	0.15	0.15	0.15	0.15	0.16	0.15
	100.60	100.35	100.34	100.71	100.78	100.31	99.91
O=S	0.02	0.01	0.01	0.03	0.02	0.02	0.04
Total	100.58	100.34	100.33	100.68	100.76	100.29	99.87
Q	6.59	5.03	5.59	5.40	6.77	6.27	5.67
Or	3.55	5.02	4.91	4.96	5.79	6.03	5.50
Ab	16.84	18.19	17.69	18.11	18.11	15.99	18.11
An	28.97	27.92	28.00	27.91	26.85	28.5	27.08
Di	20.74	20.11	19.59	19.44	18.28	18.09	19.17
Hy	19.68	19.68	19.22	19.81	19.02	20.34	19.37
Mt	1.25	1.49	2.13	1.73	2.31	1.77	1.97
Il	1.20	1.20	1.20	1.20	1.25	1.22	1.18
Ap	0.21	0.21	0.21	0.21	0.23	0.21	0.21
rest	1.39	1.31	1.63	1.73	1.98	1.68	1.42
Total	100.42	100.16	100.17	100.5	100.59	100.10	99.68
Mg#	61.0	61.1	61.1	61.3	59.5	61.7	61.4

	84 101	84 106	84 107	84 108	84 109	84 111	84 115
Cs _{INAA}	1.01	1.03	1.01	1.03	0.63	0.88	0.89
Ba	190	205	200	225	195	305	200
Ba _{INAA}	182	187	187	198	211	276	182
Rb	17.8	30.6	29.3	39.1	33.2	33.7	35.2
Rb _{INAA}	17	24	27	27	31	30	31
Rb _{ID}	17.0	28.9	27.3	28.3	35.0		
Sr	132.2	121.4	119.0	124.4	178.6	155.5	142.0
Sr _{ID}	125.0	115.1	109.2	120.1	185.1		
Pb	7	5	6	5	5	6	6
La	11	11	10	11	11	11	11
La _{INAA}	10.9	10.6	10.9	10.9	11.4	10.8	10.8
Ce	24	23	24	24	25	22	24
Ce _{INAA}	24.4	23.4	24.3	23.2	25.6	23.2	24.2
Nd _{INAA}	12.6	11.8	11.8	11.5	13.5	11.5	11.7
Nd _{ID}	9.50	11.94	12.40	12.12	13.44		
Sm _{INAA}	3.02	3.02	3.03	3.03	3.20	3.06	3.05
Sm _{ID}	2.92	2.83	2.93	2.86	3.17		
Eu _{INAA}	0.830	0.806	0.819	0.806	0.876	0.794	0.821
Gd _{INAA}	3.1	3.0	3.1	3.1	3.2	2.9	3.1
Tb _{INAA}	0.59	0.57	0.59	0.59	0.60	0.56	0.56
Ho _{INAA}	0.79	0.81	0.79	0.74	0.80	0.85	0.81
Yb _{INAA}	2.38	2.40	2.42	2.38	2.49	2.34	2.40
Lu _{INAA}	0.368	0.358	0.365	0.368	0.373	0.357	0.360
Y	21.1	20.3	20.7	20.4	20.8	20.5	20.2
Th	4	4	3	4	3	4	3
Th _{INAA}	3.5	3.5	3.4	3.3	3.5	3.3	3.6
U _{INAA}	1.1	1.1	1.2	1.1	1.0	1.0	1.0
Zr	94	94	96	95	100	95	94
Hf _{INAA}	1.9	1.8	1.9	1.8	2.0	1.8	1.9
Nb	4.0	5.0	4.5	5.0	4.5	5.0	4.5
Sc _{INAA}	40.6	41.0	41.2	40.9	42.3	40.5	40.9
V	223	231	221	222	238	225	222
Cr	113	121	112	110	82	114	114
Cr _{INAA}	120	124	119	119	83	114	123
Mn	1340	1450	1470	1410	1390	1320	1300
Co	49	49	48	49	49	49	47
Ni	79	83	80	78	71	80	79
Cu	78	79	75	74	73	77	74
Zn	82	80	78	79	81	79	77
Ga	16.0	15.5	16.0	15.5	17.0	17.0	16.5

	84 128	84 129	84 138	84 140	84 141	84 143	84 144
SiO ₂	54.21	54.15	54.61	53.83	54.41	54.81	54.08
TiO ₂	0.64	0.61	0.64	0.63	0.63	0.66	0.63
Al ₂ O ₃	14.66	14.66	14.86	14.57	14.70	14.45	14.66
Fe ₂ O ₃	0.76	0.57	0.42	1.12	1.20	1.53	0.97
FeO	8.02	8.16	8.42	7.71	7.71	7.70	7.92
MnO	0.17	0.17	0.17	0.17	0.17	0.18	0.17
MgO	6.75	6.61	6.66	6.65	6.84	6.34	6.78
CaO	10.99	10.97	10.91	10.74	10.79	9.73	10.84
CaO _{INAA}	11.3	11.4	11.0	10.9	10.8	9.9	10.8
Na ₂ O	1.74	1.88	1.87	1.78	1.95	2.05	2.11
Na ₂ O _{INAA}	1.73	1.81	1.80	1.75	1.73	1.97	1.89
K ₂ O	0.68	0.75	0.86	0.82	0.83	0.98	0.84
P ₂ O ₅	0.09	0.09	0.09	0.09	0.09	0.10	0.09
S	0.04	0.04	0.05	0.05	0.04	0.03	0.04
H ₂ O +	1.13	0.74	0.43	0.71	0.76	0.92	0.54
H ₂ O -	0.22	0.25	0.23	0.52	0.29	0.43	0.35
CO ₂	0.16	0.13	0.15	0.15	0.21	0.14	0.18
rest	0.15	0.14	0.15	0.14	0.15	0.15	0.14
	100.41	99.92	100.52	99.68	100.77	100.20	100.34
O=S	0.02	0.02	0.02	0.02	0.02	0.01	0.02
Total	100.39	99.90	100.50	99.66	100.75	100.19	100.32
Q	7.17	6.23	6.01	7.01	6.35	7.98	5.01
Or	4.02	4.43	5.08	4.85	4.91	5.79	4.96
Ab	14.72	15.91	15.82	15.06	16.5	17.35	17.85
An	30.18	29.35	29.61	29.34	28.91	27.33	28.05
Di	19.50	20.14	19.70	19.14	19.68	16.64	20.62
Hy	20.59	20.37	21.25	19.68	19.83	19.74	19.78
Mt	1.10	0.83	0.61	1.62	1.74	2.22	1.41
Il	1.22	1.16	1.22	1.20	1.20	1.25	1.20
Ap	0.21	0.21	0.21	0.21	0.21	0.23	0.21
rest	1.51	1.12	0.81	1.38	1.26	1.49	1.07
Total	100.22	99.75	100.32	99.49	100.59	100.02	100.16
Mg#	61.4	61.0	60.8	61.0	61.5	58.9	61.2

	84 128	84 129	84 138	84 140	84 141	84 143	84 144
Cs _{INAA}	3.77	4.96	1.01	1.02	1.27	1.68	1.08
Ba	190	180	190	195	310	235	200
Ba _{INAA}	178	184	181	171	264	239	178
Rb	25.4	26.4	31.7	29.8	30.0	35.6	34.6
Rb _{INAA}	23	24	28	27	29	33	25
Rb _{ID}		26.3	31.2		31.5		32.8
Sr	121.2	122.1	123.9	125.8	115.8	123.8	119.3
Sr _{ID}		119.3	121.0		114.7		111.1
Pb	6	5	6	5	6	6	6
La	12	11	10	11	12	11	12
La _{INAA}	10.9	10.6	11.0	10.9	11.0	11.4	10.6
Ce	25	24	25	24	26	27	23
Ce _{INAA}	23.6	24.0	23.9	24.2	24.1	25.9	24.2
Nd _{INAA}	11.7	12.6	11.9	12.5	12.0	13.1	12.6
Nd _{ID}		12.38	12.56		12.13		12.55
Sm _{INAA}	3.05	3.07	3.09	3.10	3.02	3.19	3.04
Sm _{ID}		2.91	2.96		3.05		2.96
Eu _{INAA}	0.813	0.803	0.822	0.829	0.825	0.878	0.817
Gd _{INAA}	3.1	2.9	3.0	3.0	3.1	3.3	3.0
Tb _{INAA}	0.56	0.55	0.57	0.60	0.58	0.59	0.55
Ho _{INAA}	0.78	0.77	0.83	0.76	0.80	0.80	0.86
Yb _{INAA}	2.34	2.41	2.37	2.38	2.37	2.42	2.26
Lu _{INAA}	0.356	0.354	0.366	0.367	0.359	0.387	0.361
Y	20.1	19.8	20.6	19.9	19.7	21.6	20.1
Th	3	4	5	5	3	4	5
Th _{INAA}	3.5	3.4	3.4	3.5	3.3	3.8	3.4
U _{INAA}	0.9	1.1	1.1	1.1	1.0	1.3	0.9
Zr	94	94	95	94	93	102	91
Hf _{INAA}	1.8	1.9	1.9	1.9	1.9	2.0	1.8
Nb	4.5	4.5	4.0	4.5	4.0	5.5	4.5
Sc _{INAA}	40.7	40.6	41.2	40.9	41.4	42.6	41.3
V	219	212	223	214	214	232	220
Cr	116	113	115	107	107	78	115
Cr _{INAA}	121	119	122	120	121	83	121
Mn	1340	1310	1330	1290	1290	1390	1320
Co	47	49	49	48	47	52	48
Ni	79	78	80	78	75	71	79
Cu	80	64	76	76	72	80	62
Zn	79	78	78	80	75	85	76
Ga	16.0	16.5	16.5	16.5	16.5	17.5	15.0

	84 147	84 148	84 149	84 150	84 152	84 167	84 167b
SiO ₂	54.22	54.65	53.79	54.17	54.30	54.19	54.19
TiO ₂	0.63	0.64	0.63	0.63	0.64	0.63	0.64
Al ₂ O ₃	14.65	14.86	14.61	14.63	14.62	14.74	14.67
Fe ₂ O ₃	0.71	0.32	0.65	0.95	1.65	0.61	0.75
FeO	8.13	8.52	8.15	7.90	7.29	8.30	8.12
MnO	0.17	0.17	0.18	0.17	0.17	0.18	0.17
MgO	6.64	6.77	6.64	6.75	6.60	6.68	6.62
CaO	10.87	10.95	10.93	10.92	10.92	10.89	10.92
CaO _{INAA}	11.2	10.9	10.9	10.8	10.9	11.1	10.0
Na ₂ O	1.86	1.81	1.76	1.85	1.68	1.94	1.83
Na ₂ O _{INAA}	1.97	1.76	1.75	1.75	1.58	1.83	1.68
K ₂ O	0.81	0.85	0.83	0.84	0.85	0.66	0.75
P ₂ O ₅	0.09	0.10	0.09	0.09	0.09	0.09	0.09
S	0.05	0.11	0.05	0.05	0.05	0.04	0.04
H ₂ O +	0.44	0.39	0.61	0.67	0.78	0.57	0.45
H ₂ O -	0.29	0.23	0.35	0.38	0.32	0.24	0.29
CO ₂	0.16	0.29	0.19	0.21	0.12	0.20	0.11
rest	0.14	0.15	0.15	0.15	0.15	0.14	0.14
	99.86	100.81	99.61	100.36	100.23	100.10	99.78
O=S	0.02	0.05	0.02	0.02	0.02	0.02	0.02
Total	99.84	100.76	99.59	100.34	100.21	100.08	99.76
Q	6.37	6.06	6.27	6.35	8.28	6.11	6.67
Or	4.79	5.02	4.91	4.96	5.02	3.90	4.43
Ab	15.74	15.32	14.89	15.65	14.22	16.42	15.48
An	29.23	29.91	29.51	29.13	29.84	29.56	29.60
Di	19.82	19.56	19.84	20.06	19.41	19.64	19.72
Hy	20.41	21.86	20.50	19.96	18.23	21.00	20.34
Mt	1.03	0.46	0.94	1.38	2.39	0.88	1.09
Il	1.20	1.22	1.20	1.20	1.22	1.20	1.22
Ap	0.21	0.23	0.21	0.21	0.21	0.21	0.21
rest	0.89	0.91	1.15	1.26	1.22	1.01	0.85
Total	99.69	100.55	99.42	100.16	100.04	99.93	99.61
Mg#	60.8	61.2	60.9	61.2	60.7	60.7	60.7

	84 147	84 148	84 149	84 150	84 152	84 167	84 167b
Cs _{INAA}	1.18	0.98	0.95	0.92	1.34	1.06	0.92
Ba	190	200	190	190	250	190	190
Ba _{INAA}	185	198	185	200	235	199	176
Rb	30.5	30.2	30.4	29.7	29.9	28.0	32.5
Rb _{INAA}	26	24	22	25	25	24	28
Rb _{ID}			34.0		30.0		32.4
Sr	120.3	123.7	128.6	128.5	123.6	115.7	115.1
Sr _{ID}			131.9		123.9		114.6
Pb	5	6	5	6	5	6	5
La	11	12	11	12	11	11	11
La _{INAA}	10.9	10.8	10.8	10.9	10.8	10.5	10.3
Ce	25	24	26	24	26	27	24
Ce _{INAA}	24.3	24.5	24.0	24.5	24.7	24.4	23.0
Nd _{INAA}	12.4	11.8	12.6	12.5	12.6	12.2	12.3
Nd _{ID}			12.71		13.96		12.63
Sm _{INAA}	3.01	3.04	3.09	3.05	3.07	3.04	2.87
Sm _{ID}			3.02		3.30		2.97
Eu _{INAA}	0.815	0.824	0.831	0.825	0.819	0.830	0.783
Gd _{INAA}	3.0	3.1	3.2	3.2	3.2	3.1	2.9
Tb _{INAA}	0.57	0.58	0.55	0.55	0.58	0.58	0.56
Ho _{INAA}	0.65	0.87	0.90	0.90	0.83	0.80	0.83
Yb _{INAA}	2.34	2.35	2.31	2.29	2.33	2.20	2.20
Lu _{INAA}	0.361	0.363	0.362	0.353	0.361	0.368	0.335
Y	19.9	20.2	20.9	19.9	20.7	20.2	19.8
Th	3	4	4	4	4	3	3
Th _{INAA}	3.3	3.3	3.4	3.4	3.6	3.6	3.7
U _{INAA}	0.9	0.9	0.9	1.0	1.2	1.3	1.0
Zr	93	95	94	92	95	93	93
Hf _{INAA}	1.9	1.9	1.7	1.9	1.9	1.9	1.8
Nb	4.5	4.5	4.0	5.0	4.5	5.0	4.0
Sc _{INAA}	40.9	41.5	40.9	41.1	40.7	41.1	38.8
V	220	224	223	223	218	227	221
Cr	113	117	117	115	111	116	114
Cr _{INAA}	118	121	120	120	120	119	112
Mn	1330	1340	1370	1330	1310	1380	1340
Co	49	51	51	50	48	51	48
Ni	79	80	79	79	76	80	78
Cu	65	77	65	71	75	64	70
Zn	77	77	79	77	76	80	77
Ga	16.0	17.0	16.5	16.0	16.5	15.5	16.5

	84 168	84 169	84 170	84 196	84 199	84 200	84 203
SiO ₂	51.04	53.70	53.91	53.99	54.54	54.51	54.44
TiO ₂	0.64	0.62	0.62	0.64	0.63	0.63	0.63
Al ₂ O ₃	15.06	14.50	14.57	14.62	14.78	14.69	14.70
Fe ₂ O ₃	2.70	1.80	1.72	1.71	1.24	0.91	0.92
FeO	5.36	7.01	7.17	7.10	7.61	8.00	7.96
MnO	0.27	0.18	0.18	0.17	0.18	0.17	0.18
MgO	5.29	6.88	6.89	6.68	6.79	6.77	6.63
CaO	12.15	9.94	9.67	10.20	10.55	10.61	10.83
CaO _{INAA}	12.2	9.7	9.8	10.7	10.8	10.4	6.5
Na ₂ O	1.74	2.01	2.20	2.10	2.20	1.95	1.68
Na ₂ O _{INAA}	1.71	1.91	2.06	1.87	2.03	1.84	1.58
K ₂ O	0.81	1.01	1.18	0.93	0.83	0.98	0.90
P ₂ O ₅	0.10	0.09	0.09	0.09	0.09	0.09	0.09
S	0.05	0.05	0.04	0.04	0.05	0.04	0.04
H ₂ O +	1.75	1.53	1.40	1.07	0.65	0.71	0.45
H ₂ O -	1.25	0.49	0.50	0.52	0.28	0.28	0.41
CO ₂	2.31	0.29	0.15	0.10	0.13	0.13	0.08
rest	0.18	0.21	0.29	0.16	0.15	0.15	0.15
	100.70	100.31	100.58	100.12	100.70	100.62	100.09
O=S	0.02	0.02	0.02	0.02	0.02	0.02	0.02
Total	100.68	100.29	100.56	100.10	100.68	100.60	100.07
Q	7.17	6.54	5.36	6.51	5.64	5.92	7.41
Or	4.79	5.97	6.97	5.50	4.91	5.79	5.32
Ab	14.72	17.01	18.62	17.77	18.62	16.50	14.22
An	30.89	27.56	26.40	27.72	28.00	28.44	29.91
Di	23.31	17.25	17.12	18.19	19.44	19.38	19.07
Hy	8.92	19.42	19.86	18.64	19.63	20.56	20.29
Mt	3.91	2.61	2.49	2.48	1.80	1.32	1.33
Il	1.22	1.18	1.18	1.22	1.20	1.20	1.20
Ap	0.23	0.21	0.21	0.21	0.21	0.21	0.21
rest	5.31	2.31	2.05	1.69	1.06	1.12	0.94
Total	100.47	100.06	100.26	99.93	100.51	100.44	99.90
Mg#	58.2	62.0	61.8	61.3	61.5	61.1	60.7

	84 168	84 169	84 170	84 196	84 199	84 200	84 203
Cs _{INAA}	1.17	2.14	2.62	2.96	1.94	1.28	0.74
Ba	460	785	1400	270	210	220	220
Ba _{INAA}	408	685	1198	258	175	194	203
Rb	29.3	38.8	44.5	36.2	31.1	37.6	35.4
Rb _{INAA}	24	29	37	31	28	30	29
Rb _{ID}	29.8	38.8	44.1	35.4			35.0
Sr	146.4	137.7	167.8	151.5	130.4	134.8	119.5
Sr _{ID}	146.5	136.9	164.3	153.5			115.5
Pb	5	5	6	6	5	6	6
La	12	11	12	12	11	11	12
La _{INAA}	11.2	10.5	10.8	10.6	10.9	10.9	10.8
Ce	26	26	28	25	24	26	25
Ce _{INAA}	24.4	23.2	24.2	24.6	24.1	24.2	24.4
Nd _{INAA}	12.9	12.0	12.6	12.6	11.9	12.6	12.3
Nd _{ID}	12.82	12.23	12.44	12.47			12.46
Sm _{INAA}	3.11	2.92	3.03	3.09	3.02	3.05	3.12
Sm _{ID}	3.05	2.89	2.93	2.96			2.94
Eu _{INAA}	0.884	0.792	0.814	0.833	0.815	0.832	0.819
Gd _{INAA}	3.4	3.0	3.1	3.0	3.3	3.2	3.0
Tb _{INAA}	0.57	0.53	0.55	0.55	0.56	0.53	0.56
Ho _{INAA}	0.88	0.78	0.81	0.81	0.77	0.81	0.78
Yb _{INAA}	2.40	2.20	2.22	2.26	2.32	2.35	2.34
Lu _{INAA}	0.377	0.352	0.362	0.366	0.357	0.355	0.359
Y	21.6	19.6	20.0	20.7	20.2	20.0	20.3
Th	5	4	5	1	3	6	2
Th _{INAA}	3.4	3.2	3.4	3.3	3.3	3.3	3.5
U _{INAA}	0.8	1.0	0.8	0.9	0.9	1.0	1.0
Zr	95	90	88	95	94	93	93
Hf _{INAA}	1.9	1.8	1.9	2.0	1.9	1.8	1.9
Nb	4.5	4.5	4.5	4.0	4.0	4.5	4.0
Sc _{INAA}	42.4	40.9	41.4	40.8	41.0	41.6	41.3
V	239	226	220	220	233	219	231
Cr	122	117	115	110	115	109	114
Cr _{INAA}	123	119	120	118	121	116	116
Mn	2110	1420	1400	1300	1370	1320	1380
Co	50	50	49	49	51	48	49
Ni	82	79	79	78	81	78	78
Cu	74	72	76	74	77	76	78
Zn	81	77	79	81	79	77	78
Ga	16.0	15.0	16.5	16.5	16.5	16.5	15.5

	84 205	84 206	84 213	84 214	84 221	84 222	84 301
SiO ₂	53.94	53.85	54.86	55.05	53.41	52.89	54.08
TiO ₂	0.64	0.63	0.68	0.67	0.62	0.64	0.63
Al ₂ O ₃	14.86	14.61	14.60	14.49	14.69	14.68	14.64
Fe ₂ O ₃	2.08	0.82	0.80	0.61	1.73	1.81	2.44
FeO	6.66	7.94	8.35	8.56	7.09	7.28	6.55
MnO	0.17	0.18	0.19	0.18	0.18	0.18	0.17
MgO	6.13	6.63	6.08	6.21	6.76	6.49	6.48
CaO	10.27	10.85	10.40	10.38	9.58	10.16	10.26
CaO _{INAA}	10.0	10.7	10.3	10.3	9.8	10.4	10.2
Na ₂ O	1.92	1.91	1.92	2.03	2.50	2.45	1.76
Na ₂ O _{INAA}	1.81	1.73	1.80	1.86	2.35	2.29	1.65
K ₂ O	0.88	0.89	0.76	1.05	1.09	1.01	0.86
P ₂ O ₅	0.09	0.10	0.11	0.11	0.09	0.09	0.09
S	0.03	0.05	0.07	0.05	0.02	0.02	0.03
H ₂ O +	1.08	1.04	0.99	0.55	2.11	2.24	1.20
H ₂ O -	1.10	0.41	0.29	0.20	0.56	0.73	0.85
CO ₂	0.04	0.41	0.11	0.07	0.14	0.12	0.25
rest	0.15	0.15	0.14	0.15	0.16	0.16	0.15
	100.04	100.47	100.35	100.36	100.73	100.95	100.44
O=S	0.01	0.02	0.03	0.02	0.01	0.01	0.01
Total	100.03	100.45	100.32	100.34	100.72	100.94	100.43
Q	8.61	5.77	8.14	6.52	3.98	3.63	9.42
Or	5.20	5.26	4.49	6.21	6.44	5.97	5.08
Ab	16.25	16.16	16.25	17.18	21.15	20.73	14.89
An	29.33	28.66	28.97	27.32	25.64	26.08	29.51
Di	17.15	20.13	18.07	19.36	17.37	19.41	16.91
Hy	16.67	19.82	20.12	20.34	19.26	17.82	17.20
Mt	3.02	1.19	1.16	0.88	2.51	2.62	3.54
Il	1.22	1.20	1.29	1.27	1.18	1.22	1.20
Ap	0.21	0.23	0.26	0.26	0.21	0.21	0.21
rest	2.22	1.86	1.39	0.82	2.81	3.09	2.30
Total	99.88	100.28	100.14	100.16	100.55	100.78	100.26
Mg#	59.6	61.0	57.9	58.3	61.6	59.9	60.3

	84 205	84 206	84 213	84 214	84 221	84 222	84 301
Cs _{INAA}	0.98	1.57	1.67	1.69	1.83	1.18	1.01
Ba	215	205	175	245	235	255	230
Ba _{INAA}	189	166	167	233	212	235	199
Rb	31.8	34.4	27.9	40.0	42.9	37.6	30.7
Rb _{INAA}	26	25	27	35	35	31	27
Rb _{ID}	32.4				42.1	38.7	31.5
Sr	123.0	125.0	130.8	143.8	204.6	185.6	150.3
Sr _{ID}	125.0				199.2	190.3	153.3
Pb	12	7	6	7	5	5	6
La	11	11	12	12	11	11	11
La _{INAA}	10.6	10.7	11.5	11.5	10.7	11.1	11.1
Ce	26	22	26	23	24	22	26
Ce _{INAA}	24.5	22.6	25.5	26.4	24.2	25.9	25.0
Nd _{INAA}	12.6	11.6	13.7	13.4	12.7	13.2	12.6
Nd _{ID}	12.62				12.28	12.73	12.73
Sm _{INAA}	3.12	3.01	3.31	3.26	2.96	3.11	3.08
Sm _{ID}	3.00				2.17	3.00	2.99
Eu _{INAA}	0.844	0.799	0.867	0.885	0.823	0.866	0.848
Gd _{INAA}	3.2	3.1	3.1	2.3	3.0	3.1	3.1
Tb _{INAA}	0.56	0.52	0.63	0.60	0.59	0.60	0.57
Ho _{INAA}	0.94	0.72	0.93	0.87	0.79	0.90	0.79
Yb _{INAA}	2.43	2.31	2.46	2.52	2.28	2.38	2.48
Lu _{INAA}	0.384	0.336	0.379	0.383	0.356	0.372	0.366
Y	21.1	20.1	21.2	21.0	19.8	20.5	20.2
Th	3	3	3	5	5	5	4
Th _{INAA}	3.5	3.3	3.8	3.7	3.4	3.8	3.6
U _{INAA}	1.1	1.4	1.0	1.4	1.1	0.8	1.6
Zr	96	93	100	101	94	95	97
Hf _{INAA}	1.9	1.7	2.1	2.2	1.9	2.0	1.9
Nb	5.0	4.5	5.0	4.5	4.0	4.0	4.5
Sc _{INAA}	42.2	40.6	43.1	43.0	41.0	42.0	41.3
V	232	224	235	235	225	232	227
Cr	108	114	80	76	116	118	107
Cr _{INAA}	115	117	83	80	119	120	111
Mn	1240	1370	1480	1380	1370	1360	1320
Co	49	48	48	49	51	50	49
Ni	83	79	70	69	79	81	76
Cu	77	76	76	74	72	75	75
Zn	98	81	79	78	79	81	77
Ga	17.0	16.0	16.5	17.0	16.5	17.0	16.5

	84 323	84 324
SiO ₂	55.17	54.87
TiO ₂	0.67	0.66
Al ₂ O ₃	14.48	14.41
Fe ₂ O ₃	1.19	1.15
FeO	7.81	7.71
MnO	0.17	0.17
MgO	6.26	6.16
CaO	10.44	10.29
CaO _{INAA}	10.5	10.2
Na ₂ O	1.87	1.81
Na ₂ O _{INAA}	1.78	1.76
K ₂ O	0.92	0.88
P ₂ O ₅	0.10	0.10
S	0.05	0.05
H ₂ O +	0.78	0.91
H ₂ O -	0.25	0.40
CO ₂	0.26	0.66
rest	0.15	0.15
	100.57	100.38
O=S	0.02	0.02
Total	100.55	100.36
Q	8.54	9.07
Or	5.44	5.20
Ab	15.82	15.32
An	28.40	28.60
Di	18.68	17.91
Hy	18.97	18.97
Mt	1.73	1.67
Il	1.27	1.25
Ap	0.23	0.23
rest	1.29	1.97
Total	100.37	100.19
Mg#	59.1	59.1

	84 323	84 324
Cs _{INAA}	1.71	2.16
Ba	230	235
Ba _{INAA}	205	209
Rb	32.7	32.4
Rb _{INAA}	28	28
Rb _{ID}		
Sr	128.9	129.2
Sr _{ID}		
Pb	6	6
La	11	11
La _{INAA}	11.2	11.1
Ce	27	25
Ce _{INAA}	26.4	26.3
Nd _{INAA}	13.1	13.6
Nd _{ID}		
Sm _{INAA}	3.27	3.16
Sm _{ID}		
Eu _{INAA}	0.882	0.882
Gd _{INAA}	3.4	3.4
Tb _{INAA}	0.64	0.63
Ho _{INAA}	0.92	0.96
Yb _{INAA}	2.46	2.45
Lu _{INAA}	0.377	0.379
Y	20.4	20.8
Th	6	5
Th _{INAA}	3.5	3.5
U _{INAA}	1.0	1.0
Zr	101	101
Hf _{INAA}	2.1	2.2
Nb	4.5	4.5
Sc _{INAA}	43.1	42.8
V	235	233
Cr	80	79
Cr _{INAA}	85	84
Mn	1290	1280
Co	48	49
Ni	71	68
Cu	72	73
Zn	79	80
Ga	17.0	17.0

XRF, INAA AND WET CHEMICAL ANALYSES OF DIFFERENTIATED
DOLERITES

	84 103	84 104	84 127	84 132	84 133	84 134	84 153
SiO ₂	62.91	67.29	54.19	54.63	55.10	56.78	54.06
TiO ₂	1.26	0.83	0.75	0.94	1.57	1.59	0.62
Al ₂ O ₃	10.86	11.09	17.64	16.99	14.83	14.39	14.52
Fe ₂ O ₃	5.90	3.51	1.52	1.99	3.52	4.17	1.25
FeO	6.22	5.32	7.24	7.93	8.77	7.91	7.62
MnO	0.18	0.14	0.14	0.14	0.16	0.15	0.17
MgO	0.53	0.20	3.10	2.52	2.09	1.43	6.68
CaO	4.54	3.60	10.25	9.45	7.95	7.14	10.64
Na ₂ O	2.59	2.89	2.29	2.45	2.69	2.72	1.77
Na ₂ O _{INAA}	2.47	2.70	2.16	2.33	2.31	2.49	1.66
K ₂ O	2.67	3.08	1.05	1.13	1.40	1.65	0.92
P ₂ O ₅	0.38	0.19	0.11	0.12	0.15	0.18	0.09
S	0.06	0.05	0.03	0.05	0.07	0.05	0.05
H ₂ O+	1.18	0.97	1.62	1.18	1.20	1.43	0.75
H ₂ O-	0.47	0.70	0.26	0.22	0.34	0.38	0.31
CO ₂	0.46	0.18	0.12	0.17	0.16	0.16	0.13
rest	0.18	0.19	0.14	0.15	0.19	0.16	0.15
	100.39	100.23	100.45	100.06	100.19	100.29	99.73
O=S	0.03	0.02	0.01	0.02	0.03	0.02	0.02
Total	100.36	100.21	100.44	100.04	100.16	100.27	99.71
Q	28.14	30.41	8.86	10.13	12.47	16.35	7.18
Or	15.78	18.20	6.21	6.68	8.27	9.75	5.44
Ab	21.92	24.45	19.38	20.73	22.76	23.02	14.98
An	10.12	8.19	34.75	32.02	24.26	22.18	28.96
Di	8.55	7.42	12.88	11.89	12.05	10.28	19.04
Hy	1.72	2.36	12.32	11.89	9.99	7.05	19.56
Mt	8.55	5.09	2.20	2.89	5.10	6.05	1.81
Il	2.39	1.58	1.42	1.79	2.98	3.02	1.18
Ap	0.88	0.44	0.26	0.28	0.35	0.42	0.21
rest	2.11	1.85	2.00	1.57	1.70	1.97	1.19
Total	100.16	99.99	100.28	99.87	99.93	100.09	99.55
Mg#	8.6	4.6	42.5	34.7	26.4	20.1	61.0

	84 103	84 104	84 127	84 132	84 133	84 134	84 153
Cs _{INAA}	2.8	5.0	1.5	1.1	1.7	1.5	1.8
Ba	585	635	260	265	325	380	265
Rb	101	121	38	39	52	60	34
Sr	114	101	189	152	146	134	137
Pb	16	17	7	7	9	11	5
La	36	40	14	16	19	24	11
La _{INAA}	35.0	39.0	13.2	13.8	17.6	22.0	10.4
Ce	83	88	34	33	47	54	27
Ce _{INAA}	82.0	87.0	29.5	32.0	40.5	49.0	23.5
Nd _{INAA}	41.5	42.5	14.6	16.2	20.0	24.0	11.8
Sm _{INAA}	9.5	10.2	3.45	3.75	4.75	5.8	2.95
Eu _{INAA}	1.86	1.75	1.01	1.09	1.23	1.45	0.8
Gd _{INAA}	9.5	9.7	3.6	3.8	4.8	6.2	2.3
Tb _{INAA}	1.69	1.74	0.66	0.73	0.90	1.06	0.57
Ho _{INAA}	2.5	2.6	0.85	1	1.3	1.5	0.75
Yb _{INAA}	6.6	7	2.7	2.85	3.55	4.4	2.25
Lu _{INAA}	0.99	1.03	0.41	0.44	0.54	0.67	0.36
Y	59	59	22	23	31	37	19
Th	13	14	3	6	9	6	5
Th _{INAA}	9.7	11.2	4.3	4.4	5.5	6.1	3.4
U	6	2	3	2	2	2	1
U _{INAA}	5.2	3.1	1	0.9	1.2	1.5	0.9
Zr	273	334	108	115	152	176	92
Hf _{INAA}	6.6	7.7	2.4	2.6	3.4	3.9	1.9
Nb	14.5	16	4.5	6	7	8	4.5
Sc _{INAA}	23	17.8	33	35	38	35	41
V	1	1	176	220	336	149	209
Cr	1	1	1	1	1	1	111
Mn	1370	1060	1050	1100	1240	1180	1280
Ni	3	1	27	16	11	3	77
Cu	37	19	109	135	157	103	72
Zn	128	111	75	81	97	102	74
Ga	19.5	18.0	19.5	21.0	21.0	21.5	16.0

	84 155	84 156	84 157	84 159	84 160	84 161	84 163
SiO ₂	53.70	53.75	53.52	53.34	53.32	54.93	55.97
TiO ₂	0.58	0.51	0.49	0.56	0.69	0.71	0.98
Al ₂ O ₃	14.56	14.78	13.78	15.12	17.56	15.03	16.02
Fe ₂ O ₃	1.28	0.98	0.93	1.16	1.49	1.74	2.20
FeO	7.26	7.17	7.58	7.27	7.00	8.56	7.93
MnO	0.16	0.16	0.17	0.16	0.15	0.18	0.15
MgO	7.25	8.08	9.08	6.78	4.08	4.32	2.31
CaO	11.15	11.65	11.56	11.50	11.20	9.83	8.70
Na ₂ O	1.59	1.69	1.44	1.61	1.98	2.13	2.58
Na ₂ O _{INAA}	1.45	1.45	1.27	1.48	1.81	1.88	2.28
K ₂ O	0.76	0.67	0.63	0.74	0.85	1.06	1.37
P ₂ O ₅	0.08	0.07	0.07	0.08	0.10	0.12	0.15
S	0.04	0.04	0.04	0.04	0.06	0.08	0.08
H ₂ O +	0.58	0.57	0.55	0.80	0.88	1.10	1.42
H ₂ O -	0.25	0.33	0.30	0.33	0.29	0.42	0.41
CO ₂	0.14	0.09	0.13	0.11	0.48	0.15	0.20
rest	0.14	0.14	0.15	0.13	0.12	0.15	0.15
	99.52	100.68	100.42	99.73	100.25	100.51	100.62
O=S	0.02	0.02	0.02	0.02	0.03	0.04	0.04
Total	99.50	100.66	100.40	99.71	100.22	100.47	100.58
Q	7.06	4.90	4.82	6.59	7.82	9.44	11.91
Or	4.49	3.96	3.72	4.37	5.02	6.26	8.10
Ab	13.45	14.30	12.18	13.62	16.75	18.02	21.83
An	30.35	30.76	29.28	31.84	36.52	28.32	28.08
Di	19.94	21.62	22.43	20.18	15.25	16.39	11.93
Hy	19.94	21.42	24.37	18.77	13.36	16.02	11.11
Mt	1.86	1.42	1.35	1.68	2.16	2.52	3.19
Il	1.10	0.97	0.93	1.06	1.31	1.35	1.86
Ap	0.19	0.16	0.16	0.19	0.23	0.28	0.35
rest	0.97	0.99	0.98	1.24	1.65	1.67	2.03
Total	99.35	100.50	100.22	99.54	100.07	100.27	100.39
Mg#	63.9	67.3	68.9	62.6	50.1	46.7	32.3

	84 155	84 156	84 157	84 159	84 160	84 161	84 163
Cs _{INAA}	1.2	0.9	0.9	0.5	1.4	0.7	2.3
Ba	185	165	150	175	200	250	305
Rb	28	24	23	27	31	39	51
Sr	116	110	105	118	141	130	147
Pb	5	4	3	6	6	6	9
La	11	9	9	11	12	15	18
La _{INAA}	9.4	8.5	7.7	9.2	11.0	13.4	16.8
Ce	21	22	19	22	29	30	43
Ce _{INAA}	20.5	19.4	18.0	20.5	25.0	31.0	39.5
Nd _{INAA}	10.2	9.2	8.8	10.2	12.2	15.0	19.0
Sm _{INAA}	2.65	2.4	2.3	2.5	3.05	3.7	4.65
Eu _{INAA}	0.73	0.68	0.66	0.75	0.9	0.98	1.17
Gd _{INAA}	2.7	2.5	2.3	2.5	3.1	3.7	4.6
Tb _{INAA}	0.49	0.46	0.47	0.49	0.55	0.69	0.87
Ho _{INAA}	0.7	0.65	0.6	0.65	0.75	0.95	1.25
Yb _{INAA}	2.05	1.88	1.86	2	2.3	2.85	3.35
Lu _{INAA}	0.33	0.3	0.3	0.32	0.35	0.44	0.52
Y	18	16	15	17	19	24	29
Th	5	3	2	5	4	7	7
Th _{INAA}	3.2	2.9	2.7	3.1	3.5	4.1	5.3
U	1	1	3	1	3	1	1
U _{INAA}	0.8	0.8	1.1	1.2	0.8	1.2	1.5
Zr	82	72	67	82	96	118	149
Hf _{INAA}	1.6	1.5	1.4	1.7	2	2.5	3.2
Nb	4	3	3.5	4	5	5	7
Sc _{INAA}	40	40	43	40	35	42	34
V	205	207	222	209	191	215	192
Cr	144	195	244	90	6	1	1
Mn	1230	1250	1320	1250	1140	1360	1180
Ni	83	95	114	79	41	38	16
Cu	67	60	62	69	85	144	120
Zn	70	65	70	69	70	91	94
Ga	16	15	14	17	18.5	18.5	20.5

	84 165	84 166	84 264	84 268	84 272	84 276	84 279
SiO ₂	55.79	57.67	53.82	54.14	54.22	53.76	54.20
TiO ₂	1.02	0.90	0.66	0.70	0.69	0.66	0.66
Al ₂ O ₃	15.81	16.48	15.99	15.45	15.55	14.69	13.88
Fe ₂ O ₃	1.99	1.48	1.85	2.05	1.83	1.99	1.60
FeO	8.32	7.88	7.27	7.28	7.32	7.24	7.63
MnO	0.15	0.13	0.18	0.16	0.16	0.17	0.17
MgO	2.17	1.57	5.07	5.12	5.60	6.33	6.82
CaO	8.84	7.94	10.88	10.80	10.31	10.77	10.95
Na ₂ O	2.46	2.85	1.92	1.97	2.37	1.77	1.87
Na ₂ O _{INAA}	2.43	2.66	1.79	1.81	2.21	1.66	1.65
K ₂ O	1.37	1.61	1.02	0.95	0.99	0.86	0.87
P ₂ O ₅	0.15	0.17	0.11	0.12	0.11	0.09	0.10
S	0.08	0.08	0.04	0.05	0.06		
H ₂ O +	1.44	1.65	0.69	0.49	0.82	1.37	1.18
H ₂ O -	0.28	0.22	0.58	0.42	0.41	0.36	0.26
CO ₂	0.16	0.22	0.07	0.11	0.12	0.11	0.10
rest	0.15	0.15	0.13	0.13	0.13	0.13	0.14
	100.18	101.00	100.28	99.94	100.69	100.30	100.43
O=S	0.04	0.04	0.02	0.02	0.03		
Total	100.14	100.96	100.26	99.92	100.66	100.30	100.43
Q	12.12	12.93	7.74	8.50	6.12	7.98	7.00
Or	8.10	9.51	6.03	5.61	5.85	5.08	5.14
Ab	20.82	24.12	16.25	16.67	20.05	14.98	15.82
An	28.05	27.42	32.00	30.51	28.87	29.60	26.91
Di	12.60	9.32	17.51	18.34	17.70	19.00	21.89
Hy	11.22	11.13	15.06	14.54	16.33	17.35	18.19
Mt	2.89	2.15	2.68	2.97	2.65	2.89	2.32
Il	1.94	1.71	1.25	1.33	1.31	1.25	1.25
Ap	0.35	0.40	0.26	0.28	0.26	0.21	0.23
rest	1.88	2.09	1.34	1.02	1.35	1.84	1.54
Total	99.97	100.78	100.12	99.77	100.49	100.18	100.29
Mg#	30.6	25.9	53.8	53.5	56.1	59.0	60.7

	84 165	84 166	84 264	84 268	84 272	84 276	84 279
Cs _{INAA}	2.6	2.7	0.9	0.8	2.2	1.2	1.0
Ba	310	345	230	210	230	210	215
Rb	50	60	36	34	36	32	31
Sr	141	151	143	126	1223	107	104
Pb	8	9	7	6	6	6	6
La	18	20	13	13	12	11	12
La _{INAA}	17.8	20.0	12.0	12.2	11.6	11.0	10.8
Ce	43	47	28	30	28	23	25
Ce _{INAA}	39.5	46	27.5	27.5	26.5	24.5	24
Nd _{INAA}	20.5	23	13	13	13	12	11.8
Sm _{INAA}	4.65	5.3	2.65	3.45	3.25	3.1	3.1
Eu _{INAA}	1.23	1.33	0.96	0.93	0.91	0.86	0.82
Gd _{INAA}	4.6	5.1	3.5	3.4	3.4	3.2	3.3
Tb _{INAA}	0.82	0.92	0.62	0.62	0.61	0.58	0.57
Ho _{INAA}	1.1	1.2	0.7	0.95	0.9	0.85	0.85
Yb _{INAA}	3.5	3.7	2.4	2.55	2.45	2.3	2.25
Lu _{INAA}	0.53	0.56	0.4	0.41	0.39	0.39	0.39
Y	29	30	21	21	21	20	20
Th	5	6	4	8	5	3	4
Th _{INAA}	5.5	6	4.5	4.4	4.3	3.6	4.2
U	2	2	1	2	1	1	2
U _{INAA}	1	1.2	1.1	0.9	1.1	1	1.3
Zr	150	168	104	105	100	97	95
Hf _{INAA}	3.2	3.8	2.2	2.2	2.1	2	1.9
Nb	7	7.5	4.5	5.5	4	4.5	4.5
Sc _{INAA}	36	27	40	40	41	41	43
V	183	132	200	217	211	213	221
Cr	1	1	17	12	25	44	64
Mn	1170	1000	1380	1270	1270	1310	1350
Ni	14	8	49	50	58	66	74
Cu	121	86	86	84	82	77	73
Zn	93	81	75	78	73	77	74
Ga	19.5	20.5	18.0	17.0	17.5	17.5	16.5

	84 283	84 288	84 291	84 295	84 299
SiO ₂	53.57	53.47	52.86	54.00	54.01
TiO ₂	0.63	0.42	0.45	0.56	0.63
Al ₂ O ₃	14.88	12.46	14.24	14.52	14.49
Fe ₂ O ₃	2.22	1.64	1.57	1.04	2.04
FeO	6.79	6.49	6.36	7.37	6.99
MnO	0.17	0.17	0.16	0.16	0.17
MgO	6.17	11.04	9.58	7.76	6.78
CaO	11.03	11.54	11.34	11.03	10.41
Na ₂ O	2.09	1.38	1.41	1.74	2.07
Na ₂ O _{INAA}	1.73	1.13	1.32	1.63	1.83
K ₂ O	0.87	0.51	0.60	0.75	0.94
P ₂ O ₅	0.12	0.06	0.08	0.10	0.14
S	0.05	0.03	0.03	0.04	
H ₂ O +	0.70	0.56	0.66	0.84	0.92
H ₂ O -	0.33	0.57	0.53	0.29	0.64
CO ₂	0.12	0.14	0.18	0.15	0.15
rest	0.14	0.17	0.15	0.15	0.17
	99.88	100.65	100.20	100.50	100.55
O=S	0.02	0.01	0.01	0.02	
Total	99.86	100.64	100.19	100.48	100.55
Q	6.54	4.44	4.88	5.89	6.63
Or	5.14	3.01	3.55	4.43	5.56
Ab	17.69	11.68	11.93	14.72	17.52
An	28.65	26.30	30.75	29.59	27.47
Di	20.63	24.55	20.13	19.94	18.95
Hy	15.21	25.89	24.10	21.65	18.08
Mt	3.22	2.38	2.28	1.51	2.96
Il	1.20	0.80	0.85	1.06	1.20
Ap	0.28	0.14	0.19	0.23	0.33
rest	1.15	1.27	1.37	1.28	1.71
Total	99.71	100.46	100.03	100.30	100.41
Mg#	59.0	74.0	71.6	65.7	61.2

	84 283	84 288	84 291	84 295	84 299
Cs _{INAA}	0.7	1.1	0.5	1.2	1.4
Ba	205	120	140	180	400
Rb	31	18	22	28	33
Sr	119	87	101	107	147
Pb	6	3	3	4	6
La	11	8	8	12	12
La _{INAA}	11.0	6.6	7.4	9.5	11.0
Ce	22	16	15	20	23
Ce _{INAA}	24.5	15.8	16.8	22.5	25.5
Nd _{INAA}	12.4	8.1	8.4	11	12.4
Sm _{INAA}	3.15	1.95	2.15	2.75	3.1
Eu _{INAA}	0.83	0.57	0.63	0.75	0.85
Gd _{INAA}	3.1	2.1	2.3	2.8	3.1
Tb _{INAA}	0.57	0.39	0.39	0.52	0.57
Ho _{INAA}	0.85	0.55	0.6	0.8	0.9
Yb _{INAA}	2.35	1.5	1.65	2.05	2.3
Lu _{INAA}	0.38	0.27	0.28	0.35	0.38
Y	20	13	14	17	20
Th	2	4	3	5	3
Th _{INAA}	3.9	2.3	2.6	3.6	4
U	1	1	2	1	1
U _{INAA}	1	0.7	0.6	1	1.2
Zr	95	59	65	80	94
Hf _{INAA}	2	1.2	1.3	1.8	2.1
Nb	4.5	2	3	4	4
Sc _{INAA}	41	44	39	41	42
V	210	224	203	208	222
Cr	60	372	302	180	110
Mn	1290	1340	1250	1270	1340
Ni	70	143	122	93	77
Cu	77	49	50	65	75
Zn	74	61	62	68	76
Ga	16.5	13.5	14.5	15.5	16.5

4.2 DATABASE OF FERRAR SAMPLES

	P-1-2	P-1-3	P-1-4	P-1-5	P-1-6	P-1-7	P-1-8
	87 85	87 86	87 87	87 88	87 89	87 90	87 91
SiO ₂	82.48	50.80	53.49	53.21	53.66	53.43	53.78
TiO ₂	0.46	1.12	0.77	0.74	0.74	0.73	0.73
Al ₂ O ₃	7.99	21.45	14.41	14.23	14.35	14.30	14.38
Fe ₂ O ₃	0.50	1.13	2.12	2.04	2.06	2.18	2.32
FeO	2.11	8.37	7.28	7.45	7.56	7.36	7.20
MnO	0.05	0.07	0.13	0.15	0.16	0.15	0.16
MgO	1.20	4.09	6.02	6.04	6.19	6.29	6.35
CaO	0.82	0.70	10.52	10.57	10.63	10.51	10.56
Na ₂ O	2.98	0.82	2.10	2.07	2.12	2.01	2.06
K ₂ O	0.27	5.44	0.74	0.78	0.83	0.82	0.86
P ₂ O ₅	0.18	0.23	0.13	0.12	0.12	0.12	0.12
S	0.01	0.01	0.02	0.01	0.01	0.01	0.01
H ₂ O+	1.30	4.93	1.51	1.35	1.17	1.10	1.07
H ₂ O-	0.11	0.51	0.56	0.50	0.39	0.29	0.35
CO ₂	0.09	0.11	0.09	0.08	0.11	0.10	0.10
Total	100.55	99.78	99.89	99.34	100.10	99.40	100.05
Ba	66	1150	205	210	215	215	220
Rb	15.0	307.0	20.0	23.5	25.5	27.0	28.5
Sr	232	355	142	145	142	145	153
Pb	20	38	5	5	6	4	5
Y	28	37	22	22	21	19	20
Zr	214	112	105	101	99	98	98
Nb	8.0	20.0	5.0	5.0	5.0	4.5	5.0
La	23	84	12	11	12	11	11
Ce	50	186	26	25	27	26	26
V	33	171	233	224	226	215	218
Cr	32	150	106	112	120	120	132
Mn	455	560	1100	1200	1250	1190	1210
Ni	20	68	86	92	88	91	93
Cu	30	55	88	91	87	94	90
Zn	40	109	72	71	72	69	65
Ga	7.0	32.0	16.5	17.0	16.0	16.5	16.5

	P-1-10	P-1-11	P-1-12	P-1-13	P-1-14	P-1-15	P-1-16
	87 92	87 93	87 94	87 95	87 96	87 97	87 98
SiO ₂	54.40	54.41	53.57	54.13	53.49	54.13	53.95
TiO ₂	0.71	0.71	0.70	0.69	0.69	0.68	0.68
Al ₂ O ₃	14.58	14.75	14.69	14.79	14.54	14.82	15.01
Fe ₂ O ₃	2.30	2.49	2.08	1.80	2.11	2.00	1.91
FeO	7.13	6.89	7.31	7.39	7.16	7.12	7.13
MnO	0.16	0.16	0.16	0.16	0.16	0.15	0.16
MgO	6.32	6.47	6.50	6.59	6.62	6.60	6.66
CaO	10.49	10.63	10.77	10.95	10.81	10.88	11.10
Na ₂ O	2.10	2.15	2.05	2.03	2.09	2.12	2.01
K ₂ O	0.91	0.85	0.83	0.82	0.76	0.78	0.75
P ₂ O ₅	0.12	0.11	0.12	0.11	0.12	0.11	0.12
S	0.00	0.03	0.02	0.02	0.04	0.01	0.01
H ₂ O+	1.06	0.92	1.13	0.96	1.05	1.03	0.96
H ₂ O-	0.35	0.27	0.36	0.34	0.35	0.31	0.34
CO ₂	0.12	0.10	0.14	0.19	0.18	0.08	0.13
Total	100.75	100.94	100.43	100.97	100.17	100.82	100.92
Ba	225	230	205	210	200	205	195
Rb	30.0	27.5	29.0	26.5	24.0	24.0	24.0
Sr	144	152	146	148	149	144	143
Pb	4	4	5	5	5	4	5
Y	20	19	20	19	19	19	19
Zr	97	92	86	92	87	88	86
Nb	5.0	5.0	5.0	4.0	5.0	4.0	4.5
La	11	12	10	12	10	11	11
Ce	23	28	23	25	23	24	25
V	216	205	212	208	214	210	216
Cr	140	129	143	147	151	146	153
Mn	1250	1130	1200	1200	1210	1190	1200
Ni	92	86	98	92	101	102	97
Cu	85	84	86	88	84	80	84
Zn	64	63	64	63	62	66	64
Ga	16.0	16.5	16.0	16.0	16.0	16.0	16.5

	P-1-17	P-1-18	P-1-19	P-1-20	P-1-21	P-1-22	P-1-23
	87 99	87 100	87 101	87 102	87 103	87 104	87 105
SiO ₂	53.38	54.00	53.41	53.35	53.81	53.18	52.97
TiO ₂	0.69	0.70	0.66	0.69	0.75	0.74	0.72
Al ₂ O ₃	14.93	14.67	14.83	14.89	14.21	14.79	14.88
Fe ₂ O ₃	1.96	2.07	1.72	1.77	1.88	2.01	1.99
FeO	7.24	7.34	7.49	7.47	7.84	7.48	7.50
MnO	0.16	0.16	0.15	0.15	0.16	0.16	0.16
MgO	6.53	6.45	6.56	6.28	6.65	6.23	6.26
CaO	11.05	10.77	10.95	10.90	10.66	10.65	10.80
Na ₂ O	2.03	2.03	2.10	2.12	2.00	2.17	2.14
K ₂ O	0.76	0.86	0.72	0.75	0.81	0.78	0.75
P ₂ O ₅	0.11	0.11	0.11	0.13	0.15	0.14	0.12
S	0.03	0.02	0.01	0.02	0.01		
H ₂ O+	0.96	1.11	0.85	0.85	0.97	0.95	1.02
H ₂ O-	0.40	0.39	0.33	0.31	0.35	0.31	0.39
CO ₂	0.11	0.16	0.09	0.13	0.10	0.07	0.10
Total	100.34	100.84	99.98	99.81	100.35	99.66	99.80
Ba	190	230	195	195	215	205	195
Rb	25.0	29.5	21.5	21.5	26.0	24.0	22.5
Sr	145	148	145	145	144	150	150
Pb	4	4	4	4	5	4	5
Y	19	19	18	19	22	20	19
Zr	86	92	82	89	96	94	89
Nb	4.0	4.0	3.5	5.0	4.5	4.0	4.5
La	10	12	10	10	11	10	10
Ce	24	25	22	24	28	24	22
V	217	194	201	208	215	218	221
Cr	139	100	127	118	108	96	101
Mn	1210	1070	1170	1200	1220	1210	1240
Ni	95	80	94	90	94	87	90
Cu	83	84	74	84	80	84	83
Zn	63	61	62	66	65	68	69
Ga	16.0	16.5	17.0	17.0	16.5	16.5	17.0

	P-1-24	P-1-25	P-1-26	P-1-27	P-1-28	P-1-29	P-1-30
	87 106	87 107	87 108	87 109	87 110	87 111	87 112
SiO ₂	53.28	52.83	53.29	53.80	53.33	52.88	52.34
TiO ₂	0.69	0.71	0.68	0.65	0.66	0.64	0.58
Al ₂ O ₃	14.69	14.59	14.41	15.12	14.76	14.78	15.47
Fe ₂ O ₃	2.13	1.96	2.03	1.82	1.75	1.61	1.52
FeO	7.37	7.43	7.45	7.30	7.43	7.22	6.91
MnO	0.16	0.15	0.16	0.15	0.16	0.15	0.14
MgO	6.39	6.37	6.62	6.64	6.72	6.96	7.05
CaO	10.70	10.77	10.86	11.12	11.11	11.18	11.74
Na ₂ O	2.13	2.06	2.02	2.05	2.02	1.97	1.88
K ₂ O	0.78	0.79	0.74	0.71	0.68	0.72	0.54
P ₂ O ₅	0.11	0.11	0.12	0.10	0.10	0.10	0.09
S	0.04	0.06	0.03				0.01
H ₂ O+	0.98	0.91	0.87	0.99	1.06	1.02	1.49
H ₂ O-	0.44	0.39	0.40	0.40	0.32	0.48	0.59
CO ₂	0.15	0.19	0.13	0.11	0.07	0.09	0.32
Total	100.04	99.32	99.81	100.96	100.17	99.80	100.67
Ba	210	205	195	185	190	180	155
Rb	25.0	24.5	22.0	21.0	19.5	21.5	15.5
Sr	149	148	146	145	144	145	146
Pb	4	5	4	4	4	4	3
Y	19	18	19	17	17	18	16
Zr	87	88	88	82	80	83	65
Nb	4.0	4.5	5.0	4.0	4.0	3.5	3.0
La	10	10	11	10	10	8	8
Ce	22	21	23	22	23	21	18
V	204	215	206	211	207	214	192
Cr	105	114	112	129	120	126	183
Mn	1170	1210	1200	1210	1200	1200	1070
Ni	83	84	81	74	73	63	59
Cu	83	84	81	74	73	63	59
Zn	64	63	64	67	61	62	64
Ga	17.0	16.0	17.0	16.5	16.5	16.0	16.0

	P-1-31	P-1-32	P-1-33	P-1-34	P-1-35	P-1-36	P-1-37
	87 113	87 114	87 115	87 116	87 117	87 118	87 119
SiO ₂	52.69	53.65	53.29	53.60	51.94	51.89	53.53
TiO ₂	0.55	0.61	0.62	0.68	0.55	0.71	0.62
Al ₂ O ₃	15.59	15.44	15.13	14.55	21.64	14.39	14.58
Fe ₂ O ₃	1.40	1.56	1.49	1.90	1.71	2.47	1.82
FeO	7.08	7.05	6.98	6.97	4.53	6.67	7.40
MnO	0.15	0.15	0.15	0.15	0.10	0.16	0.16
MgO	7.37	7.28	7.20	7.14	3.12	6.58	6.98
CaO	11.37	11.47	11.66	11.22	12.16	12.27	11.17
Na ₂ O	1.89	1.93	1.87	1.86	2.42	1.94	1.95
K ₂ O	0.60	0.65	0.68	0.69	0.83	0.60	0.67
P ₂ O ₅	0.09	0.09	0.10	0.11	0.09	0.12	0.10
S	0.05		0.01		0.01		0.02
H ₂ O+	0.88	0.82	0.98	0.91	0.93	2.03	1.01
H ₂ O-	0.35	0.34	0.31	0.32	0.20	0.57	0.30
CO ₂	0.22	0.05	0.13	0.08	0.15	0.11	0.13
Total	100.28	101.09	100.60	100.18	100.38	100.51	100.44
Ba	165	175	175	190	185	170	180
Rb	17.5	19.0	21.5	21.5	31.5	17.0	21.0
Sr	144	139	141	136	206	135	144
Pb	3	4	4	4	4	3	4
Y	14	16	17	19	14	19	18
Zr	69	72	77	85	71	78	79
Nb	2.5	4.0	4.5	4.5	4.0	5.5	3.5
La	8	9	9	10	8	9	9
Ce	18	22	20	23	19	23	22
V	180	187	198	200	129	224	205
Cr	235	217	214	211	27	172	120
Mn	1180	1120	1140	1150	780	1320	1230
Ni	111	107	112	106	39	108	104
Cu	64	67	70	73	49	78	82
Zn	59	58	58	60	43	63	64
Ga	15.0	16.0	16.0	15.0	19.0	16.0	15.5

	P-1-38	P-1-39	P-1-40	P-1-41	P-1-42	P-1-43	P-1-44
	87 120	87 121	87 122	87 123	87 124	87 125	87 126
SiO ₂	51.96	54.20	53.59	54.61	54.27	54.41	54.21
TiO ₂	0.36	0.79	0.62	0.94	0.78	0.82	0.78
Al ₂ O ₃	15.35	14.12	15.09	13.65	14.33	14.22	14.31
Fe ₂ O ₃	1.23	2.55	2.19	2.91	2.13	2.11	1.74
FeO	6.86	7.74	6.91	8.25	7.66	8.02	8.11
MnO	0.15	0.17	0.15	0.17	0.16	0.16	0.16
MgO	8.56	5.77	6.50	5.18	5.86	5.80	6.05
CaO	12.83	10.20	10.99	9.76	10.43	10.31	10.46
Na ₂ O	1.75	2.23	2.00	2.19	2.04	2.13	2.11
K ₂ O	0.43	0.89	0.84	1.05	0.80	0.78	0.81
P ₂ O ₅	0.05	0.12	0.11	0.15	0.13	0.14	0.13
S					0.01	0.03	0.02
H ₂ O+	0.81	1.07	0.99	1.17	1.09	1.21	1.31
H ₂ O-	0.19	0.35	0.40	0.36	0.32	0.41	0.39
CO ₂	0.12	0.10	0.11	0.13	0.14	0.14	0.16
Total	100.65	100.30	100.49	100.52	100.15	100.69	100.75
Ba	110	230	220	290	225	225	215
Rb	15.0	28.5	29.5	31.0	22.5	20.5	19.0
Sr	143	160	153	160	152	153	159
Pb	2	5	3	6	4	5	5
Y	10	21	18	25	22	23	22
Zr	35	96	81	119	106	106	101
Nb	2.0	5.5	3.5	6.5	5.5	6.5	5.0
La	4	12	10	14	11	12	11
Ce	9	28	24	32	27	27	27
V	190	234	191	237	228	237	234
Cr	125	19	112	36	97	89	104
Mn	1230	1270	1170	1230	1260	1250	1260
Ni	130	78	93	60	85	81	90
Cu	53	111	87	115	95	98	97
Zn	53	72	60	73	73	75	72
Ga	14.5	17.5	16.0	17.0	17.0	17.0	17.0

	P-1-45	P-1-46
	87 127	87 128
SiO ₂	63.17	62.77
TiO ₂	0.73	0.58
Al ₂ O ₃	16.15	12.07
Fe ₂ O ₃	0.68	0.52
FeO	5.11	2.95
MnO	0.06	0.54
MgO	2.79	2.19
CaO	0.67	11.06
Na ₂ O	1.36	2.50
K ₂ O	4.48	3.42
P ₂ O ₅	0.18	0.17
S		0.01
H ₂ O+	3.37	1.12
H ₂ O-	0.49	0.22
CO ₂	0.15	0.39
Total	99.39	100.51

Ba	1010	930
Rb	209.0	116.0
Sr	239	182
Pb	26	21
Y	34	30
Zr	238	253
Nb	14.0	10.0
La	43	39
Ce	95	83
V	97	57
Cr	97	58
Mn	460	4750
Ni	43	26
Cu	27	11
Zn	90	54
Ga	22.0	14.5

	87 87	87 93	87 104	87 117	87 122	87 126
SiO ₂	53.49	54.41	53.18	51.94	53.59	54.21
TiO ₂	0.77	0.71	0.74	0.55	0.62	0.78
Al ₂ O ₃	14.41	14.75	14.79	21.64	15.09	14.31
Fe ₂ O ₃	2.12	2.49	2.01	1.71	2.19	1.74
FeO	7.28	6.89	7.48	4.53	6.91	8.11
MnO	0.13	0.16	0.16	0.10	0.15	0.16
MgO	6.02	6.47	6.23	3.12	6.50	6.05
CaO	10.52	10.63	10.65	12.16	10.99	10.46
Na ₂ O	2.10	2.15	2.17	2.42	2.00	2.11
Na ₂ O _{INAA}	1.80	1.75	1.87	2.29	1.74	1.82
K ₂ O	0.74	0.85	0.78	0.83	0.84	0.81
P ₂ O ₅	0.13	0.11	0.14	0.09	0.11	0.13
S	0.02	0.03	0.00	0.01	0.00	0.02
H ₂ O+	1.51	0.92	0.95	0.93	0.99	1.31
H ₂ O-	0.56	0.27	0.31	0.20	0.40	0.39
CO ₂	0.09	0.10	0.07	0.15	0.11	0.16
Cs _{INAA}	0.5	0.5	0.5	0.7	0.7	0.6
Ba	205	230	205	185	220	215
Rb	20	28	24	32	30	19
Sr	142	152	150	206	153	159
Pb	5	4	4	4	3	5
La	12	12	10	8	10	11
La _{INAA}	11.2	10.6	10.6	8.2	9.1	11.5
Ce	26	28	24	19	24	27
Ce _{INAA}	25.0	24.0	24.0	18.6	20.5	11.5
Nd _{INAA}	13.0	12.4	12.4	9.5	10.8	13.2
Sm _{INAA}	3.30	3.05	3.15	2.35	2.80	3.40
Eu _{INAA}	0.96	0.92	0.94	0.86	0.87	0.96
Gd _{INAA}	3.6	3.1	3.4	2.5	3.0	3.7
Tb _{INAA}	0.61	0.61	0.61	0.44	0.53	0.62
Ho _{INAA}	0.85	0.85	0.85	0.60	0.75	0.85
Yb _{INAA}	2.60	2.45	2.45	1.84	2.15	2.60
Lu _{INAA}	0.41	0.38	0.38	0.27	0.36	0.42
Y	22	19	20	14	18	22
Th _{INAA}	3.4	3.1	3.0	2.6	2.8	3.4
U _{INAA}	1.3	0.9	0.9	0.6	0.7	1.1

	87 87	87 93	87 104	87 117	87 122	87 126
Zr	105	92	94	71	81	101
Hf _{INAA}	2.1	2.1	2.0	1.5	1.7	2.0
Nb	5	5	4	4	3.5	5
Sc _{INAA}	39.6	38.7	39.3	21.8	38.8	39.6
V	233	205	218	129	191	234
Cr	106	129	96	27	112	104
Mn	1100	1130	1210	780	1170	1260
Ni	86	86	87	39	93	90
Cu	88	84	84	49	87	97
Zn	72	63	68	43	60	72
Ga	16.5	16.5	16.5	19.0	16.0	17.0

4.3 DATABASE OF VICTORIAN AND SOUTH AUSTRALIAN SAMPLES

	87 129	87 130	87 131	87 132	87 133	87 134	87 135
SiO ₂	46.05	52.04	54.37	47.08	55.46	54.89	55.55
TiO ₂	2.65	0.59	0.64	2.86	0.63	0.75	0.65
Al ₂ O ₃	15.91	14.49	14.91	15.14	15.13	15.06	14.84
Fe ₂ O ₃	4.84	1.48	1.48	3.45	1.29	1.14	1.19
FeO	6.77	8.80	6.64	7.48	6.27	7.90	7.36
MnO	0.15	0.32	0.16	0.15	0.15	0.23	0.17
MgO	6.89	6.03	6.58	7.19	7.21	4.97	7.36
CaO	8.72	10.01	10.59	8.03	11.01	9.84	10.98
Na ₂ O	2.46	1.94	2.00	3.47	1.96	2.13	1.87
K ₂ O	1.74	0.75	0.46	2.51	0.76	0.79	0.71
P ₂ O ₅	0.62	0.08	0.09	0.94	0.09	0.11	0.09
S	0.01	0.02	0.04	0.01	0.02	0.05	0.04
H ₂ O+	2.49	0.73	1.36	1.00	0.55	0.71	0.50
H ₂ O-	1.08	0.67	1.24	0.32	0.54	0.77	0.22
CO ₂	0.25	3.29	0.45	1.06	0.10	1.01	0.16
Cs	0.74	1.00	1.20	0.31	0.66	1.00	1.10
Ba	785	250	640	1420	450	230	195
Rb	37.8	23.6	26.0	48.0	25.1	29.9	26.0
Sr	783.8	130.3	162.1	1029	131.8	133	155.4
Pb	4.5	5.4	4.4	4.4	6.1	6.9	7.2
La	61.1	10.6	13.1	65.5	11.5	17.0	12.6
Ce	123.0	24.2	27.5	132.0	25.7	36.9	28.6
Pr	12.90	2.75	3.56	15.70	2.86	4.91	3.44
Nd	47.7	12.6	13.6	59.5	12.5	19.0	13.7
Sm	8.43	3.10	2.93	9.70	2.93	3.71	2.85
Eu	2.51	0.83	0.83	3.08	0.73	0.94	0.74
Gd	6.30	3.12	3.34	6.56	2.56	3.96	2.74
Tb	0.97	0.65	0.56	0.96	0.55	0.71	0.49
Dy	5.33		3.58	5.57	4.00	4.76	3.39
Ho	1.01	0.85	0.74	1.04	0.95	1.03	0.73
Er	2.62	2.54	2.12	2.49	2.91	2.96	2.17
Yb	1.9	2.48	2.11	1.78	2.52	2.89	2.17
Y	27	24	19	27	23	26	18
Th	6.57	3.04	2.78	4.47	3.00	4.21	3.11
U	1.15	0.86	0.68	0.94	0.77	0.89	0.81
Zr	220	122	70	223	73	105	70
Hf	5.60	3.20	2.10	6.12	2.46	2.95	2.46
Nb	66.0	5.9	3.6	61.0	5.0	5.1	5.0
Sn	6.0	3.4	2.3	5.2	2.5	2.1	2.6
Mo	2.20	0.71	0.55	1.90	0.71	0.46	0.90

APPENDIX 5.

ISOTOPIC DATA

SAMPLE NUMBER	LATITUDE	LONGITUDE	ROCK-TYPE	LOCATION
84 221	42° 40' S	147° 07' E	Chilled Margin	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 216.38 m
84 222	"	"	Chilled Margin	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 219.23 m
84 223	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 227.38 m
84 224	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 235.24 m
84 225	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 243.47 m
84 226	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 250.38 m
84 227	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 265.61 m
84 228	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 283.08 m
84 229	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 298.65 m
84 230	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 315.41 m
84 231	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 331.49 m
84 232	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 347.45 m
84 233	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 361.04 m
84 234	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 379.45 m
84 235	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 392.46 m
84 236	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 409.53 m
84 237	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 426.76 m
84 238	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 434.10 m
84 239	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 445.99 m
84 240	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 456.18 m
84 241	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 464.46 m
84 242	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 475.12 m
84 243	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 486.33 m
84 244	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 497.55 m
84 245	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 504.32 m
84 246	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 516.14 m
84 247	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 528.09 m
84 248	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 538.05 m
84 249	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 549.27 m
84 250	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 559.30 m
84 251	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 569.94 m
84 252	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 579.13 m
84 253	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 588.78 m
84 254	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 598.95 m
84 255	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 609.25 m
84 256	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 621.10 m
84 257	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 630.05 m
84 258	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 641.32 m
84 259	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 651.10 m
84 260	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 662.11 m
84 261	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 668.76 m
84 262	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 681.81 m
84 263	"	"	Dolerite	Ross Borehole 1. Core samples courtesy Dept. Mines Tas. 693.12 m
84 264	41° 39' S	148° 06' E	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 67.69 m
84 265	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 82.75 m
84 266	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 97.76 m
84 267	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 112.75 m
84 268	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 121.75 m
84 269	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 133.78 m
84 270	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 142.79 m
84 271	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 151.75 m
84 272	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 163.81 m
84 273	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 178.81 m
84 274	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 187.82 m
84 275	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 199.83 m
84 276	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 208.81 m
84 277	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 220.81 m
84 278	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 229.82 m
84 279	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 241.78 m
84 280	"	"	Dolerite	Fingal Borehole, GY 68. Samples courtesy Dept. Mines Tas. 253.84 m

APPENDIX 6.

ELECTRON-MICROPROBE DATA

This appendix contains representative pyroxene analyses of the samples discussed in Chapter 3. A representative analysis of the fayalitic olivine preserved in sample 84-104 is also given. Plagioclase data are not included as a broad range in compositions was obtained in each section. Minerals are referred to as OPX (orthopyroxene), CPX (clinopyroxene), PIG (pigeonite) and FAY (for the fayalitic olivine).

SAMPLE	Chilled Margin	84-104	84-104	84-127	84-127	84-134	84-134
PX-TYPE	OPX	CPX	(FAY)	PIG	CPX	CPX	PIG
SiO ₂	53.26	47.12	29.20	48.74	49.87	48.70	48.63
TiO ₂	0.11	0.65	0.09	0.27	0.46	0.54	0.43
Al ₂ O ₃	1.39	0.79		1.24	1.64	1.23	0.92
Cr ₂ O ₃	0.24	0.01		0.06	0.04		
FeO	9.38	27.90	67.32	32.52	19.62	24.92	33.22
MnO	0.28	0.65	1.26	0.48	0.23	0.38	0.59
MgO	30.28	2.24	1.07	10.61	11.18	8.83	9.31
CaO	2.28	18.60	0.30	5.40	16.33	14.77	6.90
Total	97.00	97.60	99.25	99.64	99.71	99.00	100.00
ATOMIC PROP.							
Si	1.935	1.970	0.991	1.954	1.937	1.937	1.954
Ti	0.003	0.020	0.002	0.008	0.013	0.016	0.013
Al	0.059	0.039		0.059	0.075	0.058	0.043
Cr	0.007	0.000		0.002	0.001		
Fe	0.285	0.976	1.911	1.090	0.638	0.829	1.116
Mn	0.009	0.023	0.036	0.016	0.008	0.013	0.020
Mg	1.640	0.139	0.054	0.634	0.647	0.523	0.558
Ca	0.089	0.816	0.011	0.232	0.723	0.630	0.297
Total	4.026	3.983	3.006	4.021	4.005	4.005	4.001
Ca	4.4	41.8		11.8	35.9	31.6	14.9
Mg	81.1	7.1		32.1	32.1	26.2	28.0
Fe+Mn	14.5	51.1		56.1	32.0	42.2	57.1

SAMPLE	84-264	84-264	84-265	84-265	84-267	84-267	84-275	84-275
PX-TYPE	PIG	CPX	PIG	CPX	PIG	CPX	PIG	CPX
SiO ₂	53.28	51.37	51.92	51.64	52.92	52.18	53.18	51.26
TiO ₂	0.14	0.36	0.12	0.25		0.12		0.28
Al ₂ O ₃	0.73	1.32	0.74	1.12	0.55	1.03	0.64	1.20
Cr ₂ O ₃	0.00	0.00	0.02	1.14				
FeO	17.65	14.03	22.90	14.25	19.87	11.07	16.05	13.87
MnO	0.42	0.33	0.50	0.32	0.46	0.29	0.37	0.33
MgO	24.54	15.05	17.78	14.51	20.72	15.93	22.45	14.38
CaO	3.64	17.45	5.60	17.14	4.67	17.84	4.31	16.92
Total	100.00	99.94	99.57	99.25	99.30	98.46	97.01	98.24
ATOMIC PROP.								
Si	1.952	1.946	1.978	1.962	1.986	1.972	2.003	1.965
Ti	0.004	0.010	0.004	0.007	0.003	0.004	0.000	0.008
Al	0.032	0.059	0.033	0.050	0.025	0.046	0.028	0.054
Cr	0.000	0.000	0.001	0.001	0.001	0.000	0.000	0.000
Fe	0.541	0.448	0.739	0.457	0.627	0.350	0.506	0.450
Mn	0.013	0.011	0.016	0.010	0.015	0.009	0.012	0.011
Mg	1.340	0.843	1.000	0.819	1.154	0.898	1.259	0.818
Ca	0.143	0.705	0.228	0.698	0.188	0.723	0.174	0.693
Total	4.025	4.023	3.999	4.004	3.998	4.001	3.983	4.000
Ca	7.0	35.1	11.5	35.2	9.5	36.5	8.9	35.2
Mg	65.8	42.0	50.4	41.3	58.2	45.4	64.5	41.5
Fe+Mn	27.2	22.9	38.1	23.5	32.4	18.2	26.6	23.4

SAMPLE	84-276	84-276	84-281	84-281	84-283	84-283	84-289	84-289
PX-TYPE	PIG	CPX	PIG	CPX	PIG	CPX	PIG	CPX
SiO ₂	50.81	52.10	52.69	50.48	54.32	52.21	55.33	53.13
TiO ₂			0.14	0.39			0.08	0.19
Al ₂ O ₃	0.58	1.10	0.77	1.34	0.42	1.18	1.07	1.70
Cr ₂ O ₃			0.03	0.02			0.18	0.20
FeO	17.33	11.48	16.26	14.33	14.95	13.02	11.09	7.11
MnO	0.43	0.30	0.39	0.33	0.35	0.31	0.24	0.19
MgO	20.62	16.86	24.68	15.22	24.64	15.86	29.48	19.49
CaO	4.55	16.18	4.76	16.82	4.41	16.50	2.35	18.47
Total	94.42	98.13	99.71	98.93	99.09	99.26	99.82	100.48
ATOMIC PROP.								
Si	1.990	1.971	1.941	1.928	1.992	1.967	1.944	1.933
Ti	0.049	0.003	0.004	0.011	0.000	0.005	0.002	0.005
Al	0.027	0.050	0.034	0.060	0.018	0.053	0.044	0.073
Cr	0.001	0.000	0.001	0.001	0.000	0.000	0.005	0.006
Fe	0.561	0.369	0.503	0.461	0.459	0.414	0.327	0.216
Mn	0.014	0.010	0.012	0.011	0.011	0.010	0.007	0.006
Mg	1.209	0.943	1.353	0.864	1.346	0.886	1.609	1.057
Ca	0.190	0.655	0.188	0.688	0.173	0.666	0.088	0.720
Total	3.994	4.001	4.035	4.024	3.999	4.002	4.027	4.017
Ca	9.6	33.1	9.2	34.0	8.7	33.7	4.4	36.0
Mg	61.2	47.7	65.8	42.7	67.7	44.8	79.2	52.9
Fe+Mn	29.1	19.2	25.1	23.3	23.6	21.5	16.5	11.1

APPENDIX 7

ION MICROPROBE U-Pb DATING OF ZIRCONS AND BADDELEYITES

The advantages of dating zircon separates using sensitive high mass-resolution ion microprobe (SHRIMP) mass spectrometry, compared with conventional techniques, have been demonstrated in a number of recent publications (e.g. Compston *et al.*, 1984; 1986). As well as greater ease of sample preparation (i.e. no dissolution and extraction of Pb is required), the major advantage of SHRIMP is that individual grains, and growth zones within grains can be dated separately. Whereas conventional analysis gives the average composition or age of a zircon population or crystal, SHRIMP makes it possible to date each of the several different aged components that may be present in the zircons from one rock.

It was considered that if zircons could be found in samples of Tasmanian Dolerite, this type of dating could prove extremely useful in studying these rocks. Apart from simply dating the dolerites, identification of xenocrystic zircons would give a direct indication of contamination by continental crust. There are three methods of obtaining xenocrystic zircons in magmas. First, the zircons may be introduced into the source and be incorporated into the magma during partial melting, secondly, zircons may be entrained into the magma during ascent and/or emplacement, or thirdly, after emplacement, the larger intrusions may melt and assimilate roof rocks. It is, therefore, most likely that if xenocrystic zircons from the host rocks are preserved, they will be found in the chilled margin dolerites and/or the most felsic differentiated dolerites towards the top of large intrusions. Clearly, if zircons have crystallized from the dolerite magma itself, then they are also most likely to occur in the most silicic differentiates with highest Zr content towards the top of the large intrusions.

Calculations of the zircon saturation in the chilled and silicic dolerites have been performed using the model of Watson and Harrison (1983). The results can only be treated in a general sense as their model is formulated using granitic compositions, and significant extrapolation of their results is required; however, it is apparent that the chilled margin liquids with M-values of 3.1 ($M=[Na+K+2Ca]/[Al.Si]$) and Zr contents of ~95 ppm at high temperatures (~1200°C) will not be saturated in zircon. Zircon will not crystallize from these liquids, and any xenocrystic zircon would be unstable and dissolve in the melt. It should be expected then, that the chilled dolerites would not contain zircons. It may be argued that dissolution requires time, and if contamination occurred shortly prior to quenching with little time available to permit the dissolution of the zircon, then some remnant crystals may be found (cf. Kambalda). Two chilled margin samples (for which ~1 kg of material was available) were crushed and the heavy minerals separated. The absence of zircon in these samples is consistent with the model calculations; either no contamination has occurred, or the xenocrystic zircons have been consumed by the melt.

The type of model calculations performed for the chilled margin samples are not as simple to apply to the strongly fractionated dolerites chosen from Red Hill. This is because the curves used to evaluate the zircon saturation in melts apply to the Zr concentration in the liquids at the

point of saturation. In the slowly cooled differentiates, it is not clear whether zircon should begin to crystallize from a liquid of the bulk composition of this rock, or if this is more likely to occur after the crystallization of other phases.

Using the bulk compositions of samples 84-103 (273 ppm Zr) and 84-104 (334 ppm Zr) from close to the top of the Red Hill intrusion, M-values of 2.2 and 2.3 are obtained respectively. Extrapolating the results of Watson and Harrison (1983) to these values, zircon may be saturated at temperatures below approximately 750-800°C. These extrapolations may not be justified but the results indicate that zircon saturation is possible. Zircon saturation will probably become even more favourable as the Zr content of the residual liquid is increased by the crystallization of other phases in which zircon is incompatible (e.g. plagioclase).

Approximately 0.5 kg of the two samples from Red Hill were crushed and the heavy minerals separated; both samples yielded numerous zircons. The zircons are long prismatic crystals with well developed faceted terminations and are colorless to pale brown. Strong zoning occurs in many of the crystals from sample 84-103 (from the summit of Red Hill) and cores tend to be rather poorly defined. The other zircons, particularly those from sample 84-104 (~120 m lower in the intrusion) do not show strong zoning but do possess similar morphology to the rest of the population. In many of these zircons, the centre of the crystal appears to be long and hollow and is sometimes filled with a dark-brown to black material, the composition of which was not determined.

In addition to the zircons, long prismatic crystals of baddeleyite are also abundant in both samples. This is surprising as it would be expected that in siliceous melts, baddeleyite (ZrO_2) should react to produce zircon, $ZrSiO_4$ (e.g. Buttermann and Foster, 1967; Nasland, 1987). Other occurrences of baddeleyite have been noted in interstitial Zr-enriched melt phases, which are believed to represent late-stage products of magmatic differentiation (e.g. Keil and Fricker, 1974; Williams, 1978b). Despite careful examination of thin-sections of 84-103 and 84-104, neither of the Zr-rich phases were observed. It is, therefore, not clear what the physical relationship is between these minerals and other phases in the matrix. It is possible that the baddeleyites may have formed during quenching caused by late-stage volatile release and are not products of equilibrium crystallization. Alternatively, the baddeleyites may have crystallized in the early stages of cooling when zircon was not saturated in the melt, and were subsequently enclosed in a silica-poor phase which effectively removed them from further interaction with the melt.

The baddeleyite crystals are of similar appearance to the zircons although the former are flatter in cross-section and have striations parallel to their elongation. Typical zircon and baddeleyite crystals are illustrated in Figure 7.1. Between ~30-50 grains each of zircon and baddeleyite were hand-picked from each sample and prepared for ion-probe analysis (Compston *et al.*, 1984).

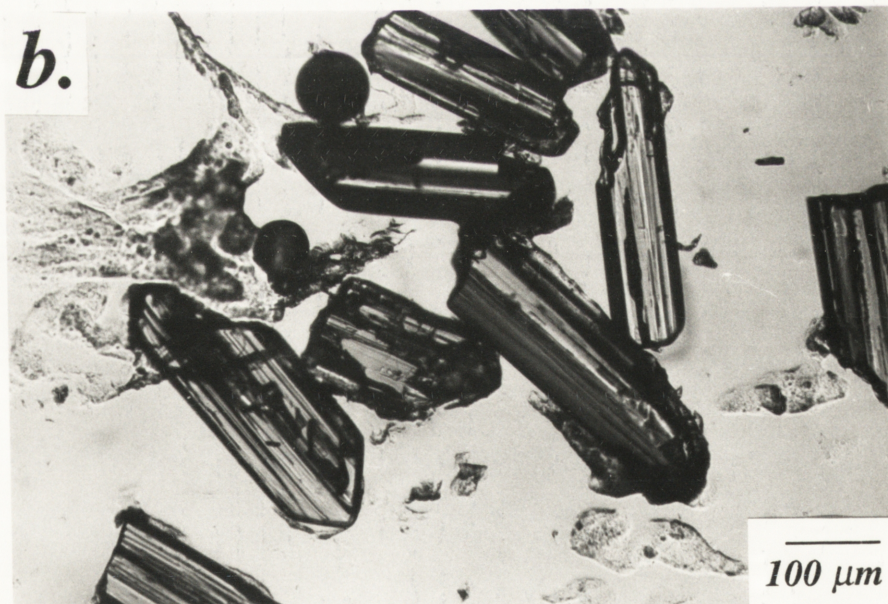
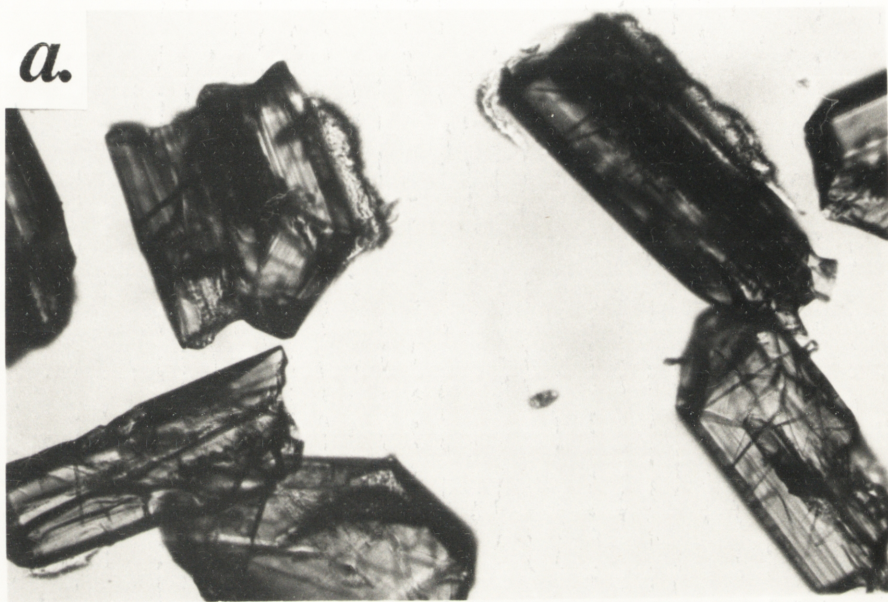


Fig. 7.1 Photographs illustrating the morphology of typical Zr-rich phases from samples 84-103 and 84-104. **a.** Euhedral zircons (broken during crushing) showing well developed crystal terminations and fine-scale zoning. The once hollow core of the crystal in the upper left-hand side has been filled with material. **b.** Prismatic baddeleyite crystals (also broken during crushing) which appear unzoned and have characteristic striations parallel to their length.

7.1 AGE DATA

As relative secondary ion yields under primary ion bombardment are matrix dependent (Shimizu and Hart, 1982 and references therein), separate reference standards were required for the two minerals, and measurements of these were interspersed with analyses of the unknowns. The zircon reference used was a Sri Lankan zircon SL3, which has a concordant age of 555 Ma and $^{206}\text{Pb}/^{238}\text{U}$ ratio of 0.0894. The relationship between Pb^+/U^+ and UO^+/U^+ is such, that a plot of $\ln(\text{radiogenic } ^{206}\text{Pb}^+/\text{U}^+)$ versus UO^+/U^+ is linear. Using the data from multiple analyses of SL3, these parameters were used to plot a calibration curve, and the Pb/U ratios of the unknowns were determined relative to this.

The baddeleyite standard TK-82-8, has a concordant age of 1135 Ma and a $^{206}\text{Pb}/^{238}\text{U}$ ratio of 0.1921. The calibration curve for this mineral is based on $\ln(\text{radiogenic } ^{206}\text{Pb}^+/\text{UO}^+)$ versus $\text{UO}_2^+/\text{UO}^+$. Although linear, this curve is much steeper than that of the zircon reference, and therefore determinations of Pb/U from $\text{UO}_2^+/\text{UO}^+$ are much more prone to uncertainty. The Pb/U of the baddeleyite unknowns were determined by using the value of radiogenic $^{206}\text{Pb}^+/\text{UO}^+$ determined from the curve, and the known Pb/U ratio of TK-82-8.

Corrections have been made for common lead assuming a Broken Hill Pb isotopic composition. The precisions of the zircon analyses were determined almost entirely by counting statistics (Compston *et al.*, 1984) and have been calculated using a computer program developed at the ANU. Unfortunately this program cannot be applied to the baddeleyite data and these have not been assigned individual analytical uncertainties. Instead, duplicate measurements of individual baddeleyite crystals indicate that the ages obtained for the entire data set can be considered to be equivalent to within error (see discussion below).

Analyses were obtained for 11 zircons and 14 baddeleyites including multiple analyses on individual grains (Table 7.1). The $^{206}\text{Pb}/^{238}\text{U}$ data corrected for common Pb using ^{208}Pb (Compston *et al.*, 1984) have been chosen for the zircons as this peak is large and less prone to errors in counting statistics. The zircons are concordant within analytical uncertainty (Fig. 7.2). Even though the very high radiogenic Pb content of the grains makes the determination of radiogenic ^{207}Pb relatively precise, the very small variation in $^{207}\text{Pb}/^{206}\text{Pb}$ is a function of time for very young materials (i.e. $^{207}\text{Pb}/^{206}\text{Pb}$ differs from 0 to 180 Ma by only ~8%), and $^{207}\text{Pb}/^{206}\text{Pb}$ is a very imprecise means of determining the zircon age. The Pb/U ratio, although determined with less precision, is a more sensitive measure of age.

The pooled mean $^{206}\text{Pb}/^{238}\text{U}$ age of 8 spots measured from sample 84-103 is 185.7 Ma, with an observed $1\sigma_{\text{mean}}$ of 1.6 Ma. This is marginally greater than the standard error expected from the precision of individual determinations ($F = 2.2$ cf. ${}^7F_\alpha = 2.01$) and there are no outliers, which almost certainly indicates that the value of 1.76% for the uncertainty in the $^{206}\text{Pb}/^{238}\text{U}$ of the standard was a slight underestimate. The best estimate of the age of 84-103 is therefore $185.7 \pm 1.6\text{Ma}$.

Only three measurements were made of zircons from sample 84-104. These also are concordant within the analytical uncertainties. The mean $^{206}\text{Pb}/^{238}\text{U}$ age of the zircons is 172.7 Ma. The observed $1\sigma_{\text{mean}}$ is 0.9 Ma, significantly less than that expected from the precision of the individual analyses (1.7 Ma). The best estimate of the age of 84-104 is therefore $172.3 \pm$

U-Pb data obtained by SHRIMP analysis of zircon and baddeleyite separates from two silicic differentiates of the Red Hill Dyke.

Sample, grain-spot	$^{206}\text{Pb}/^{238}\text{U}$ $\pm 1\sigma$	$^{207}\text{Pb}/^{235}\text{U}$ $\pm 1\sigma$	Age (Ma) $\pm 1\sigma$
ZIRCONS			
84-103/1-core	0.0301 \pm 5	0.206 \pm 5	191 \pm 3.5
84-103/2-rim	0.0308 \pm 5	0.211 \pm 5	193 \pm 3.5
84-103/2-core	0.0295 \pm 5	0.194 \pm 4	188 \pm 3.5
84-103/3-core	0.0284 \pm 5	0.201 \pm 6	181 \pm 3.0
84-103/4-core	0.0296 \pm 5	0.204 \pm 4	188 \pm 3.5
84-103/5-rim	0.0288 \pm 5	0.196 \pm 5	183 \pm 3.0
84-103/5-core	0.0288 \pm 5	0.194 \pm 5	183 \pm 3.0
84-103/6-rim	0.0285 \pm 5	0.194 \pm 5	181 \pm 3.0
84-104/1-core	0.02696 \pm 5	0.177 \pm 6	171 \pm 3.0
84-104/2-core-rim	0.0271 \pm 5	0.194 \pm 6	173 \pm 3.0
84-104/3-core	0.0274 \pm 5	0.186 \pm 5	174 \pm 3.0
BADDELEYITES			
84-103/1-1	0.034	-0.396	179
84-103/2-1	0.027	-0.022	178
84-103/3-1	0.022	-0.096	164
84-103/4-1	0.024	0.048	191
84-103/5-1	0.029	0.184	170
84-104/1-1	0.024	0.012	215
84-104/2-1	0.025	0.145	176
84-104/3-1	0.028	0.222	148
84-104/3-2	0.026	0.208	186
84-104/4-1	0.028	0.246	156
84-104/5-1	0.029	0.056	186
84-104/6-1	0.025	-0.161	158
84-104/7-1	0.030	0.281	156
84-104/7-2	0.026	0.091	175

1.7Ma. The difference between the two mean ages is significant at the 1% level (i.e. the difference between the means, divided by the standard error of the differences is 11.5).

The ages quoted for the baddeleyites are based on ^{204}Pb corrected $^{206}\text{Pb}/^{238}\text{U}$ data. Despite attempts to reduce the uncertainties in the ages obtained for the baddeleyites, there is a large amount of variation within each rock sample, and even within individual crystals (Table 7.1). More development work on the baddeleyite technique is required. Because of the large uncertainties, the baddeleyites from 84-103 and 84-104 are considered to be of the same age and average estimates from the two samples compare well.

The ages obtained for the two populations of zircons and the averaged baddeleyite results (185.7 ± 1.6 Ma, 172.7 ± 1.7 Ma and 174.1 ± 4.5 Ma respectively), are generally consistent with previous K-Ar dates obtained for this intrusion (174 ± 5 Ma; Schmidt and McDougall, 1977; recalculated using the recommended decay constants of Steiger and Jäger, 1977), and a Rb-Sr isochron age of 180 ± 13 Ma obtained in this study (Fig. 7.3).

Although the baddeleyite data, and zircon results from sample 84-104 are very similar to the K-Ar age, it is not clear why the zircons in sample 84-103 should appear significantly older. It is important to note that these data are consistent with the crystallization of these Zr-rich phases directly from the fractionated magmas. None of the zircon cores show any indication of being xenocrystic. These data support the contention that the silicic rocks towards the upper levels of the Red Hill Dyke are true magmatic differentiates and do not represent metasomatised roof rocks (McDougall, 1962; Chapter 3); therefore, the discrepancy in the age of sample 84-103 remains enigmatic.

Table 7.1 U-Pb age data (in Ma) obtained by SHRIMP analysis of zircon and baddeleyite separates. The mineral separates have been extracted from two silicic differentiates from the Red Hill Dyke in southern Tasmania. Repeat analyses of grains are given in parentheses (in the case of the zircons the rim analyses are listed first and the data for the cores are given in parentheses).

	84-103		84-104	
	Zircon $\pm 1\sigma$	Baddeleyite	Zircon $\pm 1\sigma$	Baddeleyite
	191 \pm 3.5	179	171 \pm 3.0	215
	193 \pm 3.5 (188 \pm 3.5)	178	173 \pm 3.0	176
	181 \pm 3.0	164	174 \pm 3.0	148 (186)
	188 \pm 3.5	191		156
	183 \pm 3.0 (183 \pm 3.0)	170		186
	181 \pm 3.0			158
				156 (175)
Average $\pm 1\sigma_{\text{mean}}$	185.7 \pm 1.6	176.4 \pm 4.1	172.7 \pm 1.7	172.9 \pm 6.6

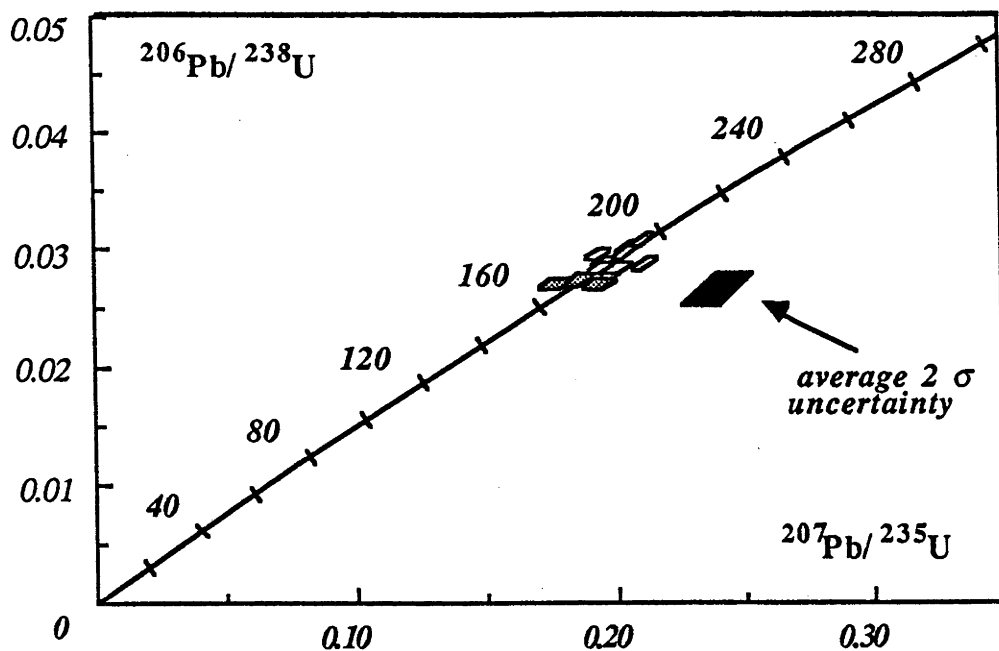


Fig. 7.2 Concordia diagram illustrating the results of the zircon analyses. The symbols correspond to the data for 84-104 (shaded), and 84-103 (open). The data are plotted as 1σ error-boxes and an estimate of the average 2σ uncertainty is included for comparison.

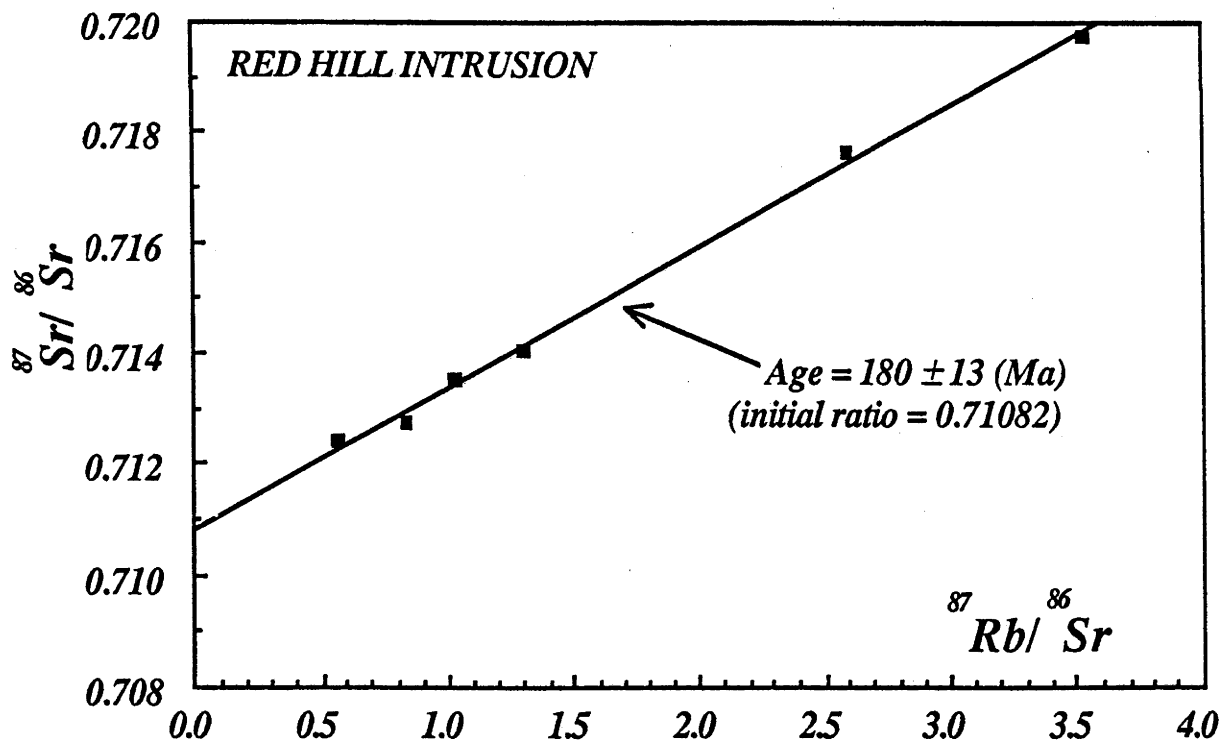


Fig. 7.3 Rb-Sr isochron diagram showing the regression of data from Red Hill (data from Appendix 5). Although the Model 1 regression gives an age of 178.5 ± 1.5 Ma, the MSWD is 52 indicating that the scatter in the data is beyond the analytical precision. The Model 3 fit of 180 ± 13 Ma is therefore preferred.

7.2 CONCENTRATION DATA

In addition to the age data, the concentrations of U, Th and Pb in the zircons are also of interest. The data for both samples are presented in Table 7.2 and indicate that the zircons have unusually high U and Th contents. Zircons from granitic rocks have been shown to contain U and Th contents generally in excess of 300 and 100 ppm respectively (Ahrens, 1965; Speer, 1980). Of the granitic zircons studied by Ahrens (1965) average concentrations of U (1330 ppm) and Th (630 ppm) were obtained, and less than ~5% of the zircons had concentrations of U>4000 ppm and Th>2000 ppm. Ion-probe analysis shows that even in the zircons from one rock, differences in U and Th content of two orders of magnitude are not unusual.

Table 7.2 U, Th and Pb concentrations in the zircons measured from samples 84-103 and 84-104. All concentrations are in ppm. f% is the amount of Pb originally in the crystal (prior to radioactive decay of U) expressed as a percentage of the total Pb present.

Sample/Grain	U	Th	Pb (total)	Th/U	f%
84-103/1 core	4989	3699	167	0.74	0.60
84-103/2 rim	7093	6070	251	0.86	0.80
84-103/2 core	5089	4354	175	0.86	0.57
84-103/3 core	3620	4673	131	1.29	0.76
84-103/4 core	6740	4965	224	0.74	0.89
84-103/5 rim	5966	3923	194	0.66	4.12
84-103/5 core	4746	4163	161	0.88	3.11
84-103/6 rim	4697	5106	164	1.09	2.44
84-104/1 core	2299	2178	75	0.95	4.00
84-104/2 core-rim	2081	1995	68	0.96	2.94
84-104/3 core	2930	1585	87	0.54	3.45

The very high concentration of U and Th in these zircons is particularly interesting as they have crystallized from the silicic differentiates of mafic magmas. It is not clear why the zircons from 84-104 have significantly lower U and Th (and therefore Pb) contents. Although the bulk rock U content of 84-104 is lower than that of 84-103 (i.e. 3.1 ppm cf. 5.2 ppm), the Th content is actually higher in 84-104 (i.e. 11.2 ppm cf. 9.7 ppm).

Ahrens (1965) noted that although the Th/U ratios in granitic rocks is about 3-4, the ratios in the zircons from granites is typically less than 1 and commonly between ~0.2 and 0.6. The Th/U ratios in the zircons from samples 84-103 (bulk rock Th/U=1.9) and 84/104 (bulk rock Th/U=3.6) vary between 0.54 and 1.29 with most falling above 0.85 (Table 7.2). There is no difference observed in this parameter between the two rock samples; that is, the lower concentrations of Th and U in zircons from 84-104 are not reflected in a change in the Th/U ratio.

7.3 SUMMARY

The only zircons (and baddeleyites) obtained from the Tasmanian Dolerites occur in the most silicic differentiates from the Red Hill Dyke. These have not been identified in thin-section and it is not clear where they occur in the rock matrix. The ages obtained on single crystals of zircon and baddeleyite from two samples are similar to those reported for the age of the intrusion obtained by K-Ar dating (Schmidt and McDougall, 1977). These Zr-rich phases crystallized from silicic differentiates of the dolerite magma and there is no evidence to suggest that any of the roof rocks were assimilated. The lack of zircon in the chilled dolerites does not rule out contamination of the dolerite magma (or its source), because zircon is not saturated in the chilled margin liquid compositions and would therefore react with the melt.

APPENDIX 8.

ESTIMATE OF THE MAGMA TEMPERATURE AT THE TIME OF INTRUSION

The low abundance of microphenocrysts in the chilled margin dolerites and the relatively uniform whole-rock geochemical compositions, indicate that these rocks may closely approximate quenched liquids. On this assumption, three samples were selected for 1 atmosphere melting experiments using a Mo-strip heater. The experiments were carried out in an Ar atmosphere to avoid oxidation, and the temperatures were determined with an optical pyrometer. The pyrometer was calibrated using multiple runs of two synthetic basalt compositions of known melting temperature. The correction due to emissivity is between 60 and 70°C which immediately introduces an error of ~10°C. The three samples chosen have slightly different MgO contents (6.84 wt%, 6.66 wt% and 6.08 wt%) but show no range in melting temperature outside the estimated analytical uncertainty (~30°C). The six runs performed give a total range from 1225 to 1255°C with an average value of 1230°C. Although this technique is a rather rough way of determining the melting point of rock samples, it provides reasonable constraints within the errors cited. If the powders used in the melting tests lose their volatiles during heating (which is likely), the estimate of ~1230°C (\pm ~30°C) for the Tasmanian chilled dolerites will be based on anhydrous conditions and must be considered an upper limit.

Using anhydrous 1 atm phase equilibria calculations (based on the models of Nielsen and Dungan, 1983; and Glazner, 1984) temperatures of between 1190 and 1205°C for liquidus plagioclase are calculated from the average chilled margin whole-rock composition. Also, using the equations for low-Ca pyroxene from Nielsen and Dungan, a liquidus temperature of ~1200°C is calculated. These results are consistent with the experimental determination obtained.

Finally, the crystallisation temperature of orthopyroxene in the chilled margins has been calculated using the model of Lindsley (1983). At pressures between ~1 atm and 10 kbar, the temperatures obtained are ~1250 \pm ~50°C.

The three independent methods used to determine the approximate temperature of the magmas at the time of intrusion have given generally consistent results. Simple melting experiments give a maximum temperature of ~1230°C (\pm 30°C); anhydrous 1 atm phase equilibria calculations indicate temperatures between ~1190-1200°C (\pm ~30°C); and pyroxene thermometry indicates values of ~1250°C (\pm ~50°C). It is concluded that the best estimate of the temperature of the intruding magma is probably close to ~1200°C.